

Theodore Eliades
George Eliades *Editors*

Plastics in Dentistry and Estrogenicity

A Guide to
Safe Practice

 Springer

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Foreword

It is estimated that 370 million direct dental restorations are placed in Europe annually [1]. Of these an increasing proportion are resin-bonded composite (RBC) restorations, because of the declining use of dental amalgam. Whereas there have been healthcare concerns over the use of amalgam, along with environmental issues, it is understandable that attention is now focussed upon RBC biomaterials as their deployment expands considerably. In this connection, it is the organic ‘resin’ phase of dental composites that attracts the main notice. This is because the ‘resin’ is supplied as a mixture of monomers that undergo polymerisation to create the composite matrix in situ. Adverse outcomes can ensue if either (a) inadequate polymerisation leaves substantial elutable monomer concentrations or (b) the monomers or polymer network contains or biodegrades to release undesirable substances such as BPA.

Although disputed by some experts, it is evident that BPA is released in non-minimal quantities from various polymers (dental composites, polycarbonates) that are used intra-orally. Many research groups have identified effects in vitro and in animals with concentrations far below the ones measured to be released from materials. Nevertheless, there is absence of proof that dental RBCs and related materials constitute a ‘clear and present danger’ to patient health. That is, a potential risk does not necessarily translate into an actual risk. The reality is that there are considerable variations in the mode of application of dental resins, in the patient’s ages and in the amounts of BPA released from different classes of material that all modify the exposure to hazard. It is likely that as the situation clarifies, different subcategories may have different risk/benefit ratios attached to them.

This book introduces and considers these issues in a careful and responsible manner showing that the evidence is not yet complete. So we need to read and build upon this evidence, meanwhile adopting a cautious attitude to the possible risk.

Manchester, UK

David Watts

Reference

1. European Commission (2012) DG ENV – final report

Contents

Part I Introduction and Overview

- 1 Endocrine Disruptors (Xenoestrogens): An Overview** 3
George Dimogerontas and Charis Liapi

Part II Methodology of Measuring BPA and Its Effects

- 2 Analytical Methods for Determination of Bisphenol A** 51
Dimitra Voutsas
- 3 In Vitro Assay Systems for the Assessment of Oestrogenicity** 79
Harris Pratsinis and Dimitris Kletsas
- 4 BPA Effects In Vivo: Evidence from Animal Studies** 89
Efthymia Kitraki

Part III Bisphenol-A in Dental Polymers

- 5 BPA and Dental Materials** 117
Jill Lewis
- 6 Bisphenol A and Orthodontic Materials** 125
Dimitrios Kloukos and Theodore Eliades
- Index** 139

Part I
Introduction and Overview

Chapter 1

Endocrine Disruptors (Xenoestrogens): An Overview

George Dimogerontas and Charis Liapi

1.1 Introduction

In the last decades, a large number of natural and synthetic chemicals have been identified as interfering with the endocrine system; they are collectively termed endocrine-disrupting chemicals (EDCs) or endocrine disruptors. According to the working definition of the Environmental Protection Agency (EPA), an endocrine disruptor is “*an exogenous agent that interferes with the synthesis, release, transport, metabolism, binding, action, or elimination of natural hormones in the body responsible for the maintenance of homeostasis, reproduction, regulation of developmental processes and/or behavior*” [1]. Endocrine disruptors comprise more than 100.000 synthetic chemical compounds that belong to different classes. A subset of the endocrine disruptors, including synthetic estrogens, natural products, commercial chemicals, industrial compounds, or by-products among which plastics, are known as environmental estrogens or xenoestrogens; they confer estrogenic potential (“estrogenicity”) translated as affinity to the estrogen receptors (ER) (α or β), thus ability to activate expression of estrogen-dependent genes or stimulation of cell proliferation of ER-competent cells [2].

Estrogens consist of an important group of steroid hormones found not only in humans but in all vertebrates, insects, and plants. In humans, estrogens (estradiol, estrone, and estriol) are primarily produced by developing follicles in the ovaries as well as by the corpus luteum of the placenta; adrenal cortex, brain, testicles, liver, and adipose tissue are smaller sources of estrogens but the only source during menopause. Estrogens are formed from the aromatization of either androstenedione or testosterone (immediate precursors) by the enzyme aromatase which is located in many tissues including adipose tissue and brain. Thus, in men, the primary source

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of estradiol is from conversion of testosterone by aromatase. In the plasma, estrogens are bound to the glycoprotein SHBG (sex hormone-binding globulin), which regulates access to the receptor [3]; SHBG serum levels are relatively high until puberty with many sites unoccupied; the large numbers of SHBG binding sites that remain unoccupied in men and women, 44 and 80 %, respectively, and especially in women taking oral contraceptives (they cause a three to five times increase in SHBG levels) are available to bind nonsteroidal ligands [4–6].

Estrogens are regulating the development, maintenance, and function of the reproductive system in both sexes, but they also exert important biologic effects in many tissues and organs, influencing many physiological processes. Given the widespread role for estrogens in many body functions, xenoestrogens, binding to estrogen receptors and acting as inappropriate estrogens, can disturb the physiology not only of the genital system, but they can also influence the integrity of many systems; they cause among others cancer, immunological, and neurological problems [7–9] in a wide range of organisms including, except for mammals, fish, birds, and reptiles [10]. Taking into consideration that the compounds with estrogen-like biological effects are ubiquitous in nature and that the endocrine systems are interlinked with each other and with other systems, EDCs have been considered as a threat for human health and for wildlife species, raising scientific, public, and political concern [11] not only at the level of health issues but also from the consequences to the economy of a country; the Belgian dioxin crisis caused an estimated damage to the Belgian economy of many million euros in addition to the number of cancers that according to estimations could reach the 8,000 as a result of the ingestion of PCBs and dioxin [12, 13]. In view of the importance of the issue, international agencies like the European Commission, the European Parliament, the US Environmental Protection Agency, the Organization for Economical Cooperation and Development, the WHO International Program on Chemical Safety, nongovernmental organizations, and the chemical industry have addressed the issue in an attempt to identify the potential risks and to develop an international research strategy [14].

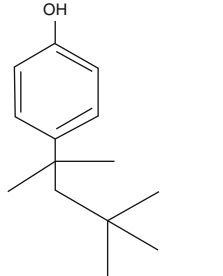
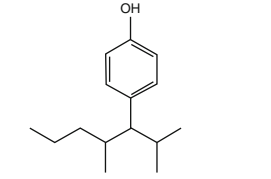
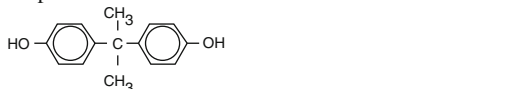

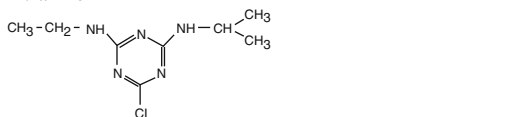

More than 80,000 chemicals with estrogenic activity are used in a vast array of industry and household products, including compounds used as pesticides such as DBCP, vinclozolin, endosulfan, dieldrin, kepone, methoxychlor, o,p-DDT, toxaphene, phenolic derivatives, and polychlorinated biphenyls (PCBs); compounds used in the food industry; antioxidants such as t-butylhydroxyanisole; plasticizers such as benzylbutylphthalate and 4-OH-alkylphenols; products associated with plastics such as bisphenol A and phthalates; industrial chemicals and by-products such as polychlorinated biphenyls (PCBs), dioxins, and benzo(a)pyrene; and heavy metals [9, 10]. The chemistry of these compounds is significantly different from the hormones they mimic, and their chemical structure does not predict the estrogenic activity they dispose (Table 1.1). Some of the most common products, except for pesticides, are flame retardants, electronic enclosures, wood preservatives, glues, cleansing and degreasing agents, polyesters, textiles, paints, lubricants, toys, personal care products, cosmetics, food and beverage containers, and dental material [15–17]. Thus, the exposure to EDCs can be through different routes such as diet, drinking water, air, and skin; dermal and inhalation exposure in industry

Table 1.1 The chemical formulas of the main estrogenic substances

Natural estrogens		
Estradiol	Estrone	Estriol
Phytoestrogens		
Genistein	Coumestrol	Equol (4',7-isoflavandiol)
Synthetic steroid estrogens		
17 α -ethinylestradiol	Mestranol	
Synthetic nonsteroid estrogenic compounds		
Chlorinated hydrocarbons		
	DDT	Methoxychlor
Polychlorinated Biphenyls (PCBs)		
Polychlorinated Biphenyls (PBBs)		
Aromatic heterocyclic compounds		
	PCDDs (Dioxins)	PCDFs
Polycyclic Aromatic Hydrocarbons (PAHs)		
	B[a]P: Benzo-a-pyrene	DB[a,1]P: Dibenzo-[α ,1]-pyrene
Aromatic Amines (AA) and Heterocyclic Aromatic Amines (HAA)		
	PhIP: 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine	4-ABP: 4-aminobiphenyl

(continued)

Table 1.1 (continued)

Alkylated phenols	<p>Octy-phenols</p> 	<p>Nonyl-phenols</p> 
Monomers of polymeric plastics	<p>Bisphenol-A</p> 	
Synthetic pyrethrines	<p>Allethrin</p> 	
Triazines	<p>Atrazine</p> 	
Pharmacological substances	<p>Diethylstilbestrol (DES)</p> 	

workers and exposure to agriculture workers are common ways of occupational exposition [18].

Surface water, municipal effluents from sewage treatment plants, and sediments are among the important contamination sources in many European and other countries [19–21] with consequent adverse effects in wildlife (fish, roach, etc.) [22, 23]; the major intake of estrogenic chemicals is considered to be through food [24, 25]. Fish products may represent an important dietary source of EDC contamination in food, but edible plants may also take up estrogenic compounds from terrestrial or aquatic environments [26, 27]. Note that weak estrogenicity has also been detected in mineral water and milk as a result of the leach from the polyethylene terephthalate (PTE) in baby bottles [28–30]. EDC chemicals are present in higher amounts in humans because humans are at the top of the food chain, having ingested plants and animals that contain low levels of these persisting compounds.

EDCs share physical and chemical properties such as chemical stability, lipid solubility, accumulation in fat, slow rate of biotransformation, and biodegradation. They are weak estrogens (most of them about 1/1,000 to 1/1,000,000 of the activity of estradiol), but small changes in more innocent compounds can give rise to persistent and bioaccumulative compounds (replacement of chloride by bromide leads to lipophilic brominated organic compounds that, although they show a weak estrogenic activity, tend to accumulate much more in fat compared to chlorinated ones) [31]. The major difference between naturally occurring biochemical molecules and man-made compounds is that the former are assembled and disassembled very rapidly in the human body, while the latter resist biodegradation in the environment and consequent bioaccumulation and biomagnification within various food chains. Thirteen years after Yu-cheng accident (literally oil symptoms), in which people in Taiwan had consumed PCB- and PCDF-contaminated cooking oil for 9 months (estimated consumption 1 g of PCBs and 3.8 mg of PCDFs), the concentrations in women that had born a child were 7- up to 130-fold higher (depending on the compound) compared to nonexposed population [32].

In contrast to endogenous hormones that bind to carrier proteins and thus become biologically inactive, EDCs remain unbound and active. The half-life of these compounds is ranging from weeks to years (i.e., half-life of methoxychlor is 2 weeks; of DDT, 6 months; of PCBs, PCDDs, and PCDFs, 7–10 years) [33].

Many estrogen-like compounds with high biologic activity are present in trace amounts, but since man is exposed to a plethora of these chemicals, the overall estrogenicity might be important and may contribute to overall risk and health implications [34]. Because of the long half-life and bioaccumulation of many EDCs, the “safe” concentrations today may become responsible for adverse effects in the following years [35].

1.2 Mechanism of Action of Estrogens and Xenoestrogens

1.2.1 Estrogen Receptor Signalling Pathway

The pleiotropic effects of estrogens in the body are mainly effectuated by binding to the estrogen receptors, ER α and ER β [36], representing products of two different genes localized on human chromosomes 6 and 14, respectively [37]. Although both isotypes exist in the various systems, ER α is the main isotype in the genital system and mammary gland, while ER β is the main isotype in the central nervous, the cardiovascular, and the immune systems; the urogenital and gastrointestinal tracts; the kidneys; and the lungs [38–42]. Various ER α and ER β isoforms and splicing variants (hER β 1 long, hER β 1 short, hER β 2, hER β 4, hER β 5, hER α -46) have been described [43, 44].

The ERs (α , β) are composed of three independent but interacting functional domains: the NH₂-terminal transcriptional AF1 (activation function-1)

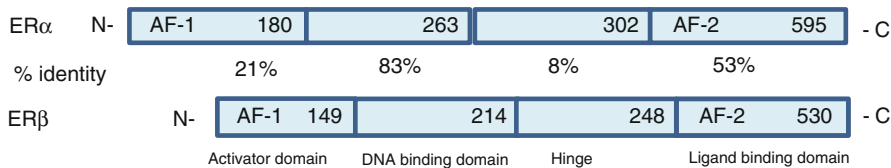


Fig. 1.1 The functional domains of ER α (*top*) and ER β (*bottom*) with the amino acids counting and the identity percent's (%) are shown. The DNA-binding domain and the hinge region are highly conserved between the two receptors. *AF-1* Activation function 1; *DBD* - DNA-binding domain; *H* Hinge; *LBD* Ligand binding domain; *AF-2* Activation function 2

domain, the DNA-binding domain, and the ligand-binding domain that contains a ligand-dependent transcriptional AF2 (activation function-2) domain [45]. Although the DNA-binding domains of ER α and ER β show a high degree of homology (only three amino acids difference), the ligand-binding domain shows only 53 % homology (Fig. 1.1).

The classical mechanism of activation of ERs, through which genomic effects take place, depends on ligand binding to the receptors, after which the receptors dimerize and bind to estrogen response elements (EREs) located in the promoters of estrogen-responsive genes to activate gene transcription [46, 47]. ER α (but not ER β) has also the ability to bind to the orphan nuclear hormone receptor SF-response elements (SFREs) that serve as its EREs [48].

Maximum transcriptional activity requires the concerted actions of the ligand-independent AF1 domain and the ligand-dependent AF2 domain. Regulatory cofactors of the transcriptional activity include coactivators, corepressors, and chromatin-remodeling complexes (chromatin is regulating the basal activity of many promoters) [46, 49–52] (Fig. 1.2).

Most of the coregulators of the activator protein-2 (AP-2) (i.e., RIP 140, TIF-2, SRC-1, and SHP) interact equally well with ER α and ER β [53–55], while others, such as the TRAP 220 coregulator, show significant differences in the interactions with ER α and ER β [56, 57]; corepressors preferentially associate with ER antagonist [58–60]. Since distinct ER α and ER β ligands are known to effect preferential recruitment of different coactivators [61, 62], the selective receptor/coactivator interactions represent an efficient system through which the pleiotropic effects of ER ligands might be mediated and are likely further determined by tissue-specific patterns of posttranslational modification of coactivators [63].

In summary, transcriptional activity of ERs is strongly influenced by ligands and the conformational changes induced upon ligand binding of ER α or ER β , the formation of dimers (i.e., ER α/α and ER β/β homodimers or ER α/β heterodimers), and the cofactor recruitment including interaction with chromatin. The co-expression of ER α and ER β in different tissues results in a heterogeneous pool of pro-proliferative ER α/α and antiproliferative ER β/β homodimers and in ER α -ER β heterodimers that have different biologic effects than the homodimers [64–69]. Thus, according to receptor subtype and the cell type [70, 71], gene activation or repression can happen [72–75].

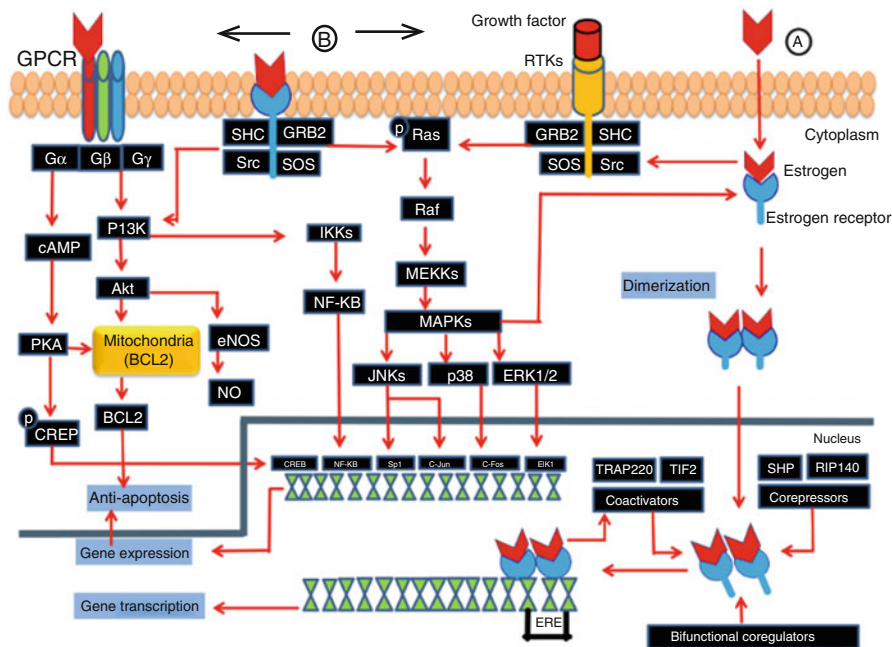


Fig. 1.2 Schematic Illustration of Classical (genomic) and Non-Classical (non-genomic) estrogen signaling pathways: (A) Classical pathway: ER complex, homodimerization and translocation from cytoplasm to to the nucleus. In the nucleus it induces two pathways: 1. Direct binding to responsive elements in the target gene promoters, subsequently the receptor-ligand complex binds to the palindromic ERE, and stimulates gene transcription with the recruitment of coregulators (coactivators, corepressors and bifunctional coregulators). 2. ER complex interacts with transcription factors such as NF- κ B, activator protein-1 and SP1 to influence gene transcription. (B) Non-Classical pathways (non-genomic): Estrogen interaction with Nonsteroidal hormone receptors or Steroid hormone receptors in the membrane. Both non-classical pathways activate kinases that ultimately regulate transcription of specific genes. These signaling cascades recruit second messengers including NO, RTKs, GPCRs, and protein kinases including PI3K, serine-threonine kinase Akt, MAPK family members, and PKA and PKC. A typical example is the induction of antiapoptosis: ER associated MAPK pathway induce rapid phosphorylation of the adaptor proteins, Src and SHC, resulting in a SHC-GRB2-SOS complex formation; this leads to the subsequent activation of Ras, Raf, and MAPKs, including ERK1/2, JNK, and p38. They are then translocated to the nucleus and participate in gene transcription. (Courtesy of Hussam Al-Humadi, MD)

Peptide growth factors are also capable of eliciting estrogen receptor-dependent activation of an ERE of DNA; ER-dependent transcriptional activation can also be elicited by both protein kinase A and protein kinase C pathways [76–79]. In some cases, genomic effects are effectuated through protein–protein interactions [80]. In the absence of an ERE (around one third of the genes in humans that are regulated by ERs do not contain ERE-like sequences [81]), the ER–ligand complexes can bind to activator protein-1 (AP-1, a transcription factor which is a heterodimeric protein composed of proteins belonging to the c-Fos, c-Jun, ATF, and JDP families)

or interact with transcription factors NF- κ B (nuclear factor- κ B), and the SP (specific protein-1) to influence gene transcription [70, 72, 82, 83].

All abovementioned genomic mechanisms of action of estrogens mediated through the ERs activation upon ligand binding take time to be effectuated, but estrogens exert rapid and transient membrane-initiated effects as well; these effects occur within seconds or minutes, are known to involve several signalling cascades, and may also influence gene transcription in the nucleus (Fig. 1.2) The second messenger signalling events include stimulation of adenylate cyclase and production of cAMP [76, 84], mobilization of intracellular calcium [85], stimulation of PI3K and PKB [86, 87], and activation of MAPK pathway of Src with consequent activation of the extracellular-regulated kinases Erk1 and Erk2 [88–92].

Although most of the rapid effects of estrogens are believed to be mediated through activation of nuclear ERS (ER α and ER β) localized near the cell surface (a small amount, approximately 2 %, of either ER α or ER β can associate with the cell membrane [93]), novel membrane ERs (mERs) have been identified in a number of tissues. Membrane receptors are located in caveolae (specialized membrane invaginations enriched in the scaffold protein caveolin-1) at the membrane [94] and can bind to caveolin-1, G proteins, PI3 kinases, Src kinase, Ras, etc. (Fig. 1.2) [95–101].

An indirect induction of nongenomic effects can indirectly activate the gene transcription (i.e., the activation of a nuclear ER through phosphorylation by both Src/Erk and PI3K signalling in the absence of a ligand), and thus, the modulation of the functions of ERs by nongenomic actions of estrogens may augment the classical mechanism of ER action.

The possible convergence of genomic and nongenomic actions on target genes is an attractive mechanism by which ERs can finely regulate gene expression [78, 102]. It has been suggested that some of the responses to selective estrogen receptor modulators (SERMs) are mediated through nongenomic actions, which subsequently lead to genomic responses [103].

Similarly to estradiol, EDCs with estrogenic activity interfere with the functioning of the complex endocrine system acting through the ER α and ER β receptor-mediated mechanism. The EDC receptor–ligand complex results in conformational changes and may activate EREs in a different way than the natural estrogen and thus influence the response in a qualitative and quantitative way, i.e., by mimicking the action of naturally produced hormones, they set off similar chemical reactions in the body, and by blocking the receptors in cells, they prevent the action of normal hormones, sometimes in a nonreversible manner [104]. Xenoestrogens seem to have equal binding affinity either to ER α or to ER β [105] [the final effect in a specific tissue seems to be regulated by the ratio of the two ER isoforms (ER α , ER β)], and they can selectively activate or repress estrogen-responsive genes in a different mode than the natural estrogens [104].

EDCs can bind either to estrogen receptors acting as estrogens or antiestrogens [106] or to androgen receptors acting as androgens or antiandrogens [106], but some EDCs can activate both receptors (bisphenol A binds to ER and acts through the genomic pathway [107], the pesticide o,p'-DDT also binds and activates the ER

[108, 109], while the p,p'-DDE (the DDT metabolite) acts as an androgen antagonist but also as a weak estrogen receptor agonist compared to o,p'-DDT [110, 111]). The chemical structure of these compounds does not predict their activities, and small changes can alter affinity for the receptor; a typical example is the 5-carbon DPP that has 3-fold increased antiandrogenic potency compared to 4-carbon DBP. It seems that the structure function relationship is very complex.

Except for the genomic effects, some xenoestrogens, such as endosulfan, nonylphenol, and o,p'-DDE, induce rapid nongenomic effects by binding to membrane estrogen receptors (mER α , mER β , and GPR30); the consequent activation or inhibition of several kinases including Erk1/Erk2, PI3K, MAPK, PKC, and PKA kinases triggers signal cascades. Activation of Ca²⁺ and K⁺ channels, intracellular Ca²⁺ concentration signals, cell proliferation, and apoptosis are effectuated through these pathways in several cell types [112–115].

Nongenomic effects of xenoestrogens have been observed in many cell types including pituitary cancer cells, breast cancer cells, cells of the immune system, neuronal cells, and bone tissues [109, 115–119].

The different classes of EDCs show a diversity of effect patterns and a distinct effect profile: they can induce genomic (nuclear) and nongenomic (extranuclear) effects or both of these effects, independently of each other and thus in conjunction with the activation or inhibition of other signalling pathways (e.g., PI3K); this might lead to an indirect promotion of the transcriptional activity of the ER [107, 120]. The substantial differences in the way they exert their effects through steroid receptors and the ability of compounds to activate either or both pathways are mostly influenced by their chemical structure: as an example, the long-carbon-side-chain alkylphenols show weak estrogenic activity in genomic assays and the shorter-side-chain versions even weaker, while the short- or long-carbon-chain variants show quite robust nongenomic activities [121–123].

Some EDCs, such as the chlorinated hydrocarbon β -HCH, although not binding to ER, are capable of activating ER target genes in a pattern very similar to the profile observed with estrogens. In order to explain these type of estrogenic effects of some compounds, Norman et al. [124] have proposed for the ERS the “two ligand-binding domains,” the “*classical*” and the “*alternative*” ligand-binding domain, responsible for the prolonged genomic events and the rapid nongenomic signalling, respectively. According to this theory, a ligand binds to the binding site it better fits; the conformation of a membrane-bound receptor favors binding to alternative site. Thus, compounds such as β -HCH and p,p'-DDE might have different affinities for the two proposed binding domains of the ER: p,p'-DDE fails to interact with the “*alternative*” domain in the membrane ER and consequently nongenomic effects cannot happen; on the contrary, β -HCH shows affinity to the “*alternative*” domain and thereby a sustained activation of Src/Ras/Erk pathway that may also lead to the strong activation of a number of other signalling cascades (such as PI3K and PKC), in addition to Src/Ras/Erk pathway [125–127].

As many estrogen agonists and antagonists, xenoestrogens have the ability to selectively bind membrane estrogen receptor [selective membrane receptor modulators (SmERMs)].

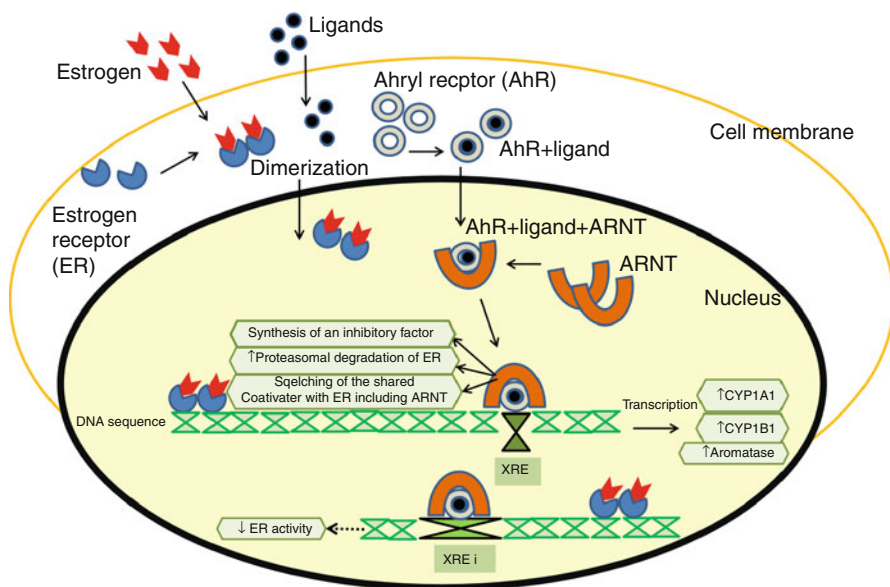


Fig. 1.3 Activation of ER and AhR receptor including relevant proposed mechanisms of cross-talk between their signaling pathways from the step of heterodimerization for AhR with ARNT and homodimerization for ER. AhR has been reported to inhibit ER activity through a combination of several different mechanisms: direct inhibition by the activated AhR/ARNT heterodimer through binding to inhibitory xenoestrogen receptor (iXRE) present in ER target genes; squelching of shared coactivators, including ARNT; synthesis of an unknown inhibitory protein; increased proteasomal degradation of ER; and altered estrogen synthesis/metabolism through increase in aromatase, cytochrome P450 1A1 and 1B1 expression. ((Courtesy of Hussam Al-Humadi, MD)

Although many of the EDCs' effects are through binding to estrogen receptors, acting as agonists or antagonists, some of them bind to androgen or aryl hydrocarbon receptor (AhR) [128] (Fig. 1.3). The aryl hydrocarbon receptor (AhR) is a member of the basic helix–loop–helix Per (Period)–ARNT (aryl hydrocarbon nuclear translocator)–SIM (single-minded) (bHLH-PAS) family [129]. Upon ligand binding, the AhR translocates from the cytoplasm to the nucleus where it binds its dimerization partner ARNT. The activated AhR/ARNT heterodimer complex binds to xenobiotic response elements (XREs) and activates the expression of AhR target genes, such as cytochrome P4501A1 (CYP1A1) and CYP1B1 [130]. As shown in AhR-null animals, the AhR mediates most, if not all, of the toxic effects of 2,3,7,8-tetrachlorodibenzo-[p]-dioxin (TCDD) [131]. Although its physiological role is unknown, the AhR has been shown to be important in liver development and female reproduction [132]. Since AhR is a receptor for many ligands, striking synergistic effects can be anticipated.

Inhibitory crosstalk between the AhR and ER signalling was suggested by early experiments examining the long-term effects of TCDD treatment in Sprague Dawley rats [133]. The first observations, that the incidences of both mammary and uterine

tumors decreased in female rats [133] after exposure to dioxins, were supported by other reports demonstrating that TCDD inhibits the formation of 7,2-dimethylbenz[a]anthracene (DMBA)-induced mammary tumors [134].

The precise molecular mechanisms for this crosstalk are unclear and may be a combination of several different mechanisms. Several studies have reported that the activated AhR inhibits the expression of E₂-induced genes [135, 136], causes decrease of the levels of nuclear ER, and is involved in mediating the antiestrogenic responses in target cells/organs [137–140]. Thus, it is possible that the primary antiestrogenic action of TCDD is to downregulate expression of the ER gene, thereby reducing cellular ER levels.

EDCs can disrupt the homeostasis of a multicellular tissue by inhibiting the gap junctional communication (the intact communication between adjacent p cells through the connexin-lined gap junctions (Gjs) is a requisite for maintaining homeostasis) [141]. Furthermore, they can dysregulate the effects of effects in an indirect way by disrupting hormone levels; they can inhibit or activate the expression of the P450 enzymes, with consequent alterations in the synthesis, transport, metabolism, and excretion of endogenous hormones, i.e., inhibition of enzymatic activity of the P450 family members CYP19 and CYP3A1, which convert testosterone to estradiol, decreases the hormone synthesis.

EDCs may act at the cellular and molecular levels, binding to both steroid and aryl hydrocarbon receptors exhibiting both dependent and independent receptor modulations of specific gene transcriptional elements [142–144]. As a result, xenoestrogens have the potential to variably modulate cell proliferation, cell cycle progression, apoptosis, and cytokine production in much the same way as 17- β -estradiol does [145, 146].

1.2.2 Cytochrome P (CYP) Induction

Exposure to EDCs can interfere with the induction of the phase I enzymes of the cytochrome P450 family. An example is CYP1A that is inducible by several classes of EDCs including the halogenated aromatic hydrocarbons (HAHs), the polycyclic aromatic hydrocarbons (PAHs), and the dioxins [2,3,7,8-tetrachlorodibenzo-p-dioxin (TCDD)] [147]. In the presence of a ligand, the AhR/ARNT heterodimer binds to the xenoestrogen responsive elements in the promoter of the CYP1A gene [147, 148] while at the same time the induction of many other genes happens [149, 150] (Fig. 1.3). CYP induction that occurs by a process involving de novo RNA and protein synthesis [151] has been shown to be important in the metabolism of xenoestrogens and the generation of reactive genotoxic metabolites [152, 153]. By this procedure, weakly active procarcinogens can be transformed into electrophilic intermediate metabolites capable of reacting with DNA, raising the risk of developing cancer; a typical example is the case of breast cancer in which the induction of the ile/val and val/val alleles of the cytochrome P450 1A1 gene, under certain circumstances, may result in increased risk [154].

1.3 Effects

The endocrine system regulates complex functions and thus hormone dysregulation results in a wide array of effects. Endogenous estrogens (17- β -estradiol, estrone, and estriol) are not only regulating the development, maintenance, and function of the reproductive system in both sexes [155–157] but they also exert important biologic effects in many tissues and organs: they affect cognition and behavior in the central nervous system; they are involved in the cardiovascular health, have a significant impact on cell-mediated and humoral immune and autoimmune responses, and play a role in adipocyte development and function as well as in bone growth and epiphyseal plate closure in both sexes [158–163]. They are implicated in the development or progression of numerous diseases including breast and colon cancer, osteoporosis, cardiovascular and neurodegenerative diseases, endometriosis, and obesity [164–167].

In relation to estrogens, EDCs can have direct toxic effects on an endocrine gland and indirect endocrine toxicity to non-endocrine organs [168]. The effects of EDCs in wildlife have been documented by many studies; the most prominent include masculinization in snails, hermaphroditism in fish, distorted sex organ development and function in reptiles (alligators and turtles), abnormal nesting behavior and induced eggshell thinning in birds, and disturbed reproduction and immune functions in grey seals [169, 170].

In humans, xenoestrogens have been mainly accused for cancer, neurological and immunological effects, reproduction failure, and osteoporosis, but data are still contradictory. The link between man-made chemicals and adverse effects that usually appear as domino effect is not quite clear. The causative role of chemical substances in diseases and abnormalities related to endocrine substances has not been well documented in human health, even though various articles have appeared describing the growing evidence that man-made chemicals are causing adverse effects in both humans and wildlife by poisoning the hormone system. The fact that adverse effects in animals do not predict the same results in humans and many effects appear to be species specific makes the issue even more difficult; i.e., exposure to phthalates causes suppression of testosterone in rat and stimulation of testosterone in the mouse, while no clear effects have been demonstrated in humans [171, 172]. The accidents in Seveso and in Taiwan (Yu-cheng disease) gave a lot of information about the connection of EDCs and human health [171, 173], but in order to establish a clear cause–effect relationship, geographical, social, diet, lifestyle and inter-population variations should be taken into consideration.

The diversity of mechanisms, the complexities and interactions of endocrine signalling mechanisms, the variety of possible end points, and the broad range of chemicals possibly involved in the adverse effects in humans and wildlife make the issue difficult to understand in its various aspects; thus, before the hypothesis becomes a certainty, it might take a lot of time.

Some of the effects associated to the exposure to xenoestrogens are presented in the following sections.

1.3.1 *Effects on Female Genital System*

The fundamental role of estrogens in females during puberty and reproductive cycling is well known [174, 175]. Estradiol exerts complex effects on gonadotropin-releasing hormone (GnRH) neuronal function including long-term genomic effects through binding to ER α and/or ER β subtypes and rapid nongenomic effects such as glutamate-induced currents in hippocampal neurons and second messenger cascades in hippocampal or hypothalamic neurons [85, 176–179].

Xenoestrogens seem also to affect the physiology of the genital system in women, since in epidemiological studies, they have been associated with menstrual disorders, abnormal ovulations, endometriosis, and spontaneous abortions.

Similarly to estrogens, EDCs, in particular *o,p'*-DDT, can modulate the GnRH secretion in vitro in the immature female hypothalamus through both slow and rapid effects; in these effects, glutamate plays an important role with the participation of genomic and nongenomic pathways involving several receptors (ERs, AHR, and AMPA) and intracellular kinases (A, C, and MAPK) [180]. The early onset of puberty observed after exposure to EDCs is probably connected to GnRH stimulation [181–183].

Epidemiological studies have shown that women who consumed fish from Lake Ontario, polluted with organic pollutants, had reduced cell cycle length, while endocrine dysfunction was found in women exposed to pentachlorophenol [184–186]. Endometriosis represents a common gynecological condition reaching 5–15 % of childbearing-age women and up to 3–5 % of postmenopausal women. Although endometriosis has developed in rats [187, 188] and in monkeys [189] after exposure to dioxins, the data in women are still conflicting; thus, in some studies, EDCs acting through an Ah receptor mechanism [190] have been associated with endometriosis, while in others, there is no evidence of such association [191, 192]. Early menopause has also been referred in women exposed to perfluorocarbons [193].

The concentrations of estrogens in the plasma seem an important factor for the manifestation of the effects after exposure to chemicals with estrogenic activity; thus, menopausal women under hormone replacement therapies are less vulnerable than those who do not take estrogens, while prepubertal girls are more vulnerable compared to older ones [194]. Several of the effects of the estrogenic compounds are also due to the alterations in the aromatase activity and thus changes in the estrogen concentrations [195, 196].

Reproduction problems have also been connected to the extensive use of EDCs. A decrease in fertilization rates after IVF has been found in couples in which the husband was exposed to EDCs (pesticides) [197]. Furthermore, exposure to organochlorides has been associated in some studies with spontaneous abortions [198–200], but according to other reports, pregnancy outcome was not affected [201].

A causal relationship between malformations in the urogenital system and exposure to EDCs had been strongly suggested by the “feminization” of the population in some areas with high discharge of these compounds [202]. The prevalence of birth of more girls than boys from young fathers in the Seveso accident in 1976, a fact observed

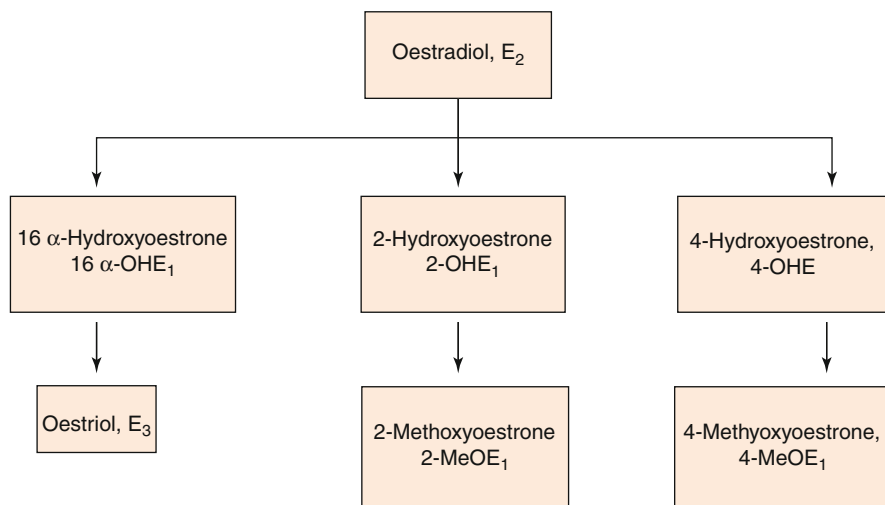


Fig. 1.4 Metabolic products of Oestradiol (E₂)

and in many industrial countries, has been connected with exposure to dioxins [203, 204]. Structural and functional defects in the female reproductive tract have been observed after exposure to diethylstilbestrol and other xenoestrogens such as the pesticide methoxychlor; these compounds have been shown to disrupt the development of the female reproductive tract by altering HOX gene expression (HOX gene determines the differential developmental identity of the Müllerian duct) [205, 206].

Estrogens have also been implicated in endometrial cancer through involvement of G protein-coupled estrogen receptor GPR30 and consequent activation of the PKC pathway [207]. Another possible mechanism is that the 2-OH and 4-OH estrogen metabolites can be further oxidized to semiquinones and quinones, which can form bulky DNA adducts and initiate carcinogenesis (Fig. 1.4).

1.3.2 Effects on Male Genital System

Estrogens have a fundamental role in male genital system and consequently in male fertility. Estrogen receptors have been found in mature and fetal testis and in epididymis as well, indicating their importance in regulation of spermatogenesis: ER α is mainly localized in Leydig cells, and ER β is mainly localized in Leydig and most germ cells, while aromatase, the enzyme that converts testosterone or androstenedione to estradiol, is found in Leydig cells, Sertoli cells, and germ cells [208–210]. Estrogens induce also both proliferative and antiproliferative effects; some of these effects are mediated through binding to ER β , and consequent downregulation of the androgen receptor ending in induction of apoptotic mechanisms [211].

In view of the important role of estrogens in the male genital system, and since men are routinely exposed to estrogen-like compounds, various changes in male physiology and fertility have been attributed to these chemicals [212, 213].

Studies in wildlife have shown that some of the EDCs, as the metabolite of p,p' DDE, exhibit antiandrogenic activity; exposure of the alligators in Lake Apopka to DDT showed a progressive decline in the population and abnormal genital structure [214]. Impaired fertility was also shown in experimental animals after exposure to lindane or PCBs [215, 216].

Prenatal exposure of experimental animals to DES resulted in increased incidence of cryptorchidism, urethral abnormalities, testicular hypoplasia, poor semen quality, rete testes adenocarcinoma, and cell hyperplasia [217–219]. Similarly, DES-exposed males have shown pseudohermaphroditism, genital malformations (small testes, testicular abnormalities, microphallus), and reduced semen quality [220]. The genital system seems to be vulnerable when the exposure to estrogenic chemical compounds happens only at a critical period of neonatal life [221]. The effects of some xenoestrogens on sperm quality seem to be effectuated through a nongenomic pathway [222].

Epidemiological studies in various European countries including France, Sweden, Scotland, and Greece have shown a progressive decline in sperm analysis attributed to exposure to compounds with estrogenic activity (i.e., the pesticide dibrochloropropane, DBCP) [223–227]. Reduced sperm concentration and motility was found in higher prevalence in semirural and agricultural areas compared to more urban areas. Workers exposed to dioxin had decreased serum testosterone and increased LH [228]. The increased incidence of hypospadias and cryptorchidism in some countries has also been associated with prenatal or paternal exposure to EDCs [229, 230].

Increased incidence of testicular cancer, a malignancy more common in young men, has been observed in many countries [231–233]. Etiologic agents or conditions for testicular cancer include, among others, exposure to pesticides and field exposure to hydrocarbons and polyvinyl chloride, but many authors have linked the increased incidence with embryonal exposure to EDCs [234–237]. The mothers of men with testicular cancer showed higher concentrations estrogenic compounds [238].

Although the data are not conclusive and sometimes are contradictory, the most possible etiology for the testicular dysgenesis syndrome (TDS) (a disorder of the male reproductive function including decrease of sperm count and increased incidence of testicular cancer and hypospadias and cryptorchidism), which has shown an increase in a small time period, seems to be rather an environmental and not a genetic factor [234, 238, 239].

According to recent studies, xenoestrogens can affect male fertility through a transgenerational epigenetic action on male reproduction system; thus, a transient in utero exposure to a xenoestrogen influences the embryonic testis transcriptome and through epigenetic effects results in abnormal germ cell differentiation that subsequently influences male fertility [240].

1.3.3 Breast Cancer

Estrogens are hormones with genotoxic potential and may act as carcinogens at non-physiological doses. Their carcinogenic effects seem to be independent of ERs,

although ERs could play a role in the early stages of cell transformation, invasion, and tumorigenesis [241]. Increased expression of specific proteins and induction of oxidants and aldehydes ending to and cause lipid peroxidation are among the mechanisms estrogens cause cancer. Furthermore, several oncogenes have been shown to encode the growth factors and their receptors that are activated by estrogens (close relationship), i.e., *c-erb-1* oncogene encodes the EGF-r (transmembrane receptor protein, whose extracellular domain is overexpressed in many cancers) [242, 243]. Estrogenic compounds, given their capacity to perturb normal hormonal actions, have been associated to the development of hormone-dependent cancers, such as breast and endometrial cancers and testicular cancer [244].

Similarly to estrogens, many EDCs induce an increased activity in a series of genes in which transcription products are growth factors involved in the carcinogenic process, i.e., EGF, TGF α , IGF1, and their receptors; this fact makes proliferation uncontrollable [245]. The first well-studied case of the association between cancer and estrogenic compounds is the example of diethylstilboestrol (DES), in which the daughters of the pregnant women that had been exposed to DES developed a clear cell adenocarcinoma of the vagina and the cervix [246].

Cancers are traditionally presumed to occur without a threshold; as a consequence, any dose of a carcinogen is associated with an increased risk.

Estrogens have been implicated in the development of breast cancer. In the USA, each year 44,000 women die of breast cancer, making it the leading cause of cancer deaths among American women that do not smoke and among those aged 40–55 years. Increased incidence of breast cancer in all age groups has been shown in various countries. The elevated incidence of breast cancer has been associated with prolonged and cumulative exposure to high levels of estrogens, i.e., early onset of menarche and late menopause, obesity, and hormonal replacement therapy (HRT) [247–251]. Decreased levels of SHBG (sex hormone-binding globulin) have also been associated with increased incidence of breast cancer due to increased levels of free estrogens. Estrogen metabolites have also been implicated in the increased incidence of breast cancer. A shift of the normal metabolic pathways of estrogens to alternative routes may involve carcinogenic metabolites; thus, if instead of the activation towards the 2-hydroxyestrone metabolite (2OHE1) production that acts as weak antiestrogen and is not carcinogenic, a shift to 16 α OHE1 pathway occurs, it gives rise to fully potent active estrogens (Fig. 1.4). The 16 α OHE1 pathway metabolites are genotoxic and carcinogenic; they circulate in very small amounts but they remain unbound due to their low affinity for SHBG and thus are free to covalently bind to the nuclear ER and to form stable adduct interacting with nuclear histone proteins [252].

ER α , the important receptor for breast development [253], is the mammary mediator of estrogenic effects in breast cancer (both in cell cultures and in breast tissue) [254]. Several sequence variations or single-nucleotide polymorphisms (SNPs) in the ER α gene (*ER1*) have been associated with increased risk of cancer [255].

Reduced binding of estradiol to SHBG may increase risk for breast cancer development [256].

Increased levels of androgens have also been implicated in breast tumor development mainly serving as substrates for estrogen [257].

In a number of epidemiological and cross-sectional studies, EDCs, such as PCBs, DDE, and dieldrin, are included among the risk factors for breast cancer [258, 259]. Increased risk for breast cancer has been shown in countries with medium or high levels of exposure to various EDCs (i.e., triazine pesticides) [260]; a positive correlation between organochloride concentration in adipose tissue and the development of breast cancer [261, 262] has been shown as well. A possible connection between the levels of some pesticides acting as xenoestrogens in breast milk and adipose tissue cannot be excluded [261, 263]. The fact that in Seveso a decrease incidence in breast cancer was reported shows the complexity of the whole issue [264].

Epidemiological data have linked early-life TCDD exposure and diets high in fat to increased risk for breast cancer in humans; high-fat diet has been shown to increase sensitivity to maternal TCDD exposure, resulting in increased breast cancer incidence, by changing metabolism capability [265]. Even a single exposure to a xenoestrogen, if it happens during a critical period of life, may alter epithelial differentiation and lead to increased multiplicity of tumors or decreased latency of tumor formation [266]. P 53 mutations may also be implicated in the susceptibility to EDCs for breast cancer development [267].

The carcinogenic or noncarcinogenic effects of estrogens have been associated to the initiation of estrogen metabolism by cytochrome P450 enzymes CYP1B1, CYP1A1, and CYP1A2 [147, 268–270]. Estrogenic compounds like dioxins, PCDFs, and some PCBs, acting in a similar way, induce CYP1A1, CYP1A2, and CYP1B1 gene expression by serving as aryl hydrocarbon receptor (AhR) agonists; CYP1A1 and CYP1B1 catalyze hydroxylation of the A-ring of estradiol (E2) to form the catechol estrogen 2- or 4-hydroxylestradiol (2-OH-E2 or 4-OH-E2, respectively) (Fig. 1.4) [271].

The discrepancies between EDCs as risk factors and breast cancer found in various studies are probably due to the fact that in the breast cancer development a complex mixture of estrogenic chemicals is involved and not only one factor [272].

1.3.4 Obesity

Nowadays, obesity has risen dramatically not only in industrialized countries but also in poorer countries reaching epidemic proportions [273].

Estrogens through ER α and ER β are also involved in the regulation of body fat distribution and metabolism [274–277]; ER β has been shown to have anorectic effects mediated via the central nervous system [278], while disruption of ER α in the ventromedial nucleus of the hypothalamus leads to weight gain, increased visceral adiposity, hyperphagia, hyperglycemia, and impaired energy expenditure in female mice [279].

Studies in DES-exposed mice have indicated that the increase in body weight was associated with an increase in the percent of body fat [280, 281], and significant alterations in genes involved in fat distribution were altered [282]. Similarly to endogenous estrogens, a link between exposure to environmental chemicals (such

as estrogenic chemicals, BPA, PCBs, DDE, and persistent organic pollutants and heavy metals) and the development of obesity has been shown in epidemiologic studies, in support of the findings in experimental animals and show [283–287].

Polychlorinated biphenyls (PCBs) and organochlorine pesticides have been associated with high levels of total serum lipids, fat mass, and BMI, while non-dioxin PCBs were shown to be inversely associated with BMI [288–291]. Moreover, prenatal and early-life PCB exposures have been associated with increased weight in boys and girls at puberty [292]. Other studies report a link between some persistent organic pollutants and increased body weight and diabetes [293].

1.3.5 Diabetes

Diabetes mellitus represents one of the most serious health problems worldwide with more than 177 million people suffer from it, and it is among the leading causes of death [294]. Estradiol seems to play an important role in energy balance, lipid metabolism, and glucose homeostasis [294–299]. Estradiol increases insulin biosynthesis and release in an ER α -dependent manner [300, 301], rapid nongenomic insulinotropic action on β cells is effectuated through ER β [302], and both receptors (ER α and ER β) can modulate GLUT4 expression in skeletal muscles of mice [303]. Similarly to E₂, exposure to BPA and other persistent organic pollutants (POPs) like dioxins, furans, polychlorinated biphenyls (PCBs), or organochlorine pesticides, stored in white adipose tissue, have been strongly associated with type 2 diabetes and with most of the components of the metabolic syndrome; cardiovascular disease and liver enzyme abnormalities are established in several cross-sectional studies [290, 304–307]. BPA exposure disrupts pancreatic β -cell function and causes hyperinsulinemia [300] and mild insulin resistance, increases basal and insulin-stimulated glucose transport (due to an increased amount of GLUT4 glucose transporter) [308], stimulates adipogenesis [309, 310], and inhibits adiponectin release leading to increased risk for metabolic syndrome [311]. A significant relationship between BPA concentration in urine and type 2 diabetes has been found [304].

1.3.6 Neurologic Defects

The development of the central nervous system both in utero and during childhood is a continuous process in which many morphologic changes take place. Estrogens play an important role in neural development. Both ER receptors are highly expressed in the brain; ER α receptors are present in higher concentrations in the hippocampus, and ER β receptors are present in higher concentrations in the basal forebrain and cerebral cortex. The neuroprotective effects of estrogens against neuronal cell death [312, 313] have been documented both in vitro and experimental animals: estrogens regulate the dopaminergic neurotransmission [314–316], promote the growth and survival of cholinergic neurons, and increase cholinergic

activity, but they also have antioxidant and antiapoptotic effects [317–320]. The data from clinical studies in neurodegenerative diseases (Alzheimer’s disease and Parkinson disease) are inconsistent and even controversial; estrogens have been positively, negatively, or with no effect correlated with the onset and the severity of the diseases [319, 321–324]. Increased risk of dementia has been associated with lower lifetime endogenous estrogens [325, 326]. Male–female differences in the clinical and cognitive characteristics of several diseases have also extensively been discussed [327–332]. The protective effects have been connected to ER α rather than ER β activation, through ER-dependent and ER-independent mechanisms or both are involved [313, 333].

Similarly to estrogens, EDCs act directly on CNS, and since the brain is a very sensitive target of steroid action, especially during development, EDC exposure might cause severe problems. Reproductive behavior, learning and memory, and other functions are permanently impaired after perinatal exposure of experimental animals [334]; male rats exposed to TCDD during the perinatal period have shown altered sexual differentiation in the brain, involving sexually dimorphic reproductive and nonreproductive neural end points. EDCs also affect neuronal synapse formation [335]. Many of the effects of the EDCs in CNS are mainly effectuated through the ER β receptor, but the effects vary depending on the chemical compound, i.e., bisphenol A and methoxychlor affect the dopaminergic and noradrenergic systems in rodents [334, 336], but they are associated to sensory or cognitive deficits after exposure during the neurodevelopment period [337]. The effects of TCDDs though seem to be effectuated through the Ahr receptors found on GABAergic neurons (GABA and glutamate regulate learning and memory functions, stress responses, social behaviors, and mood [338, 339]).

Exposure of humans to EDCs has resulted in effects on behavior changes, learning problems, memory, attention deficit, and impairment of sensory and psychomotor development [340, 341]. Neurologic disorders and cognitive (impairment of memory and attention) and behavioral problems had been reported in young children whose mothers had consumed food contaminated with PCBs in addition to growth retardation [342, 343].

Exposure to PCBs has been associated with a memory deficit at 7 months and at 4 years of age, while up to 11 years of age, a negative association was reported between deficit of IQ (intelligence quotient) and PCBs’ concentrations index (the index was comprising maternal and cord serum and maternal milk concentrations). Other studies have reported a significant decrease in mental developmental index score as a function of maternal breast milk levels of PCBs at 2 weeks postpartum, which probably reflects maternal body burdens during pregnancy. Most of the studies showed a decreased IQ and poorer cognitive functioning in preschool children [344–346]. Verbal functions are longlasting, while visual–spatial functions, episodic memory, and sustained attention may be less sensitive to prenatal PCB exposure [347].

Some EDCs cross the placenta readily and the blood–brain barrier in the fetus; thus, exposure to these agents can impair mental and physical ability due to altered bioavailability. It is becoming clear that developmental exposure to EDCs and dioxin-like compounds can permanently impair neuroendocrine functions. Several studies, in various countries, have been conducted in order to establish the

association between prenatal exposure to xenoestrogens and several aspects of psychomotor development [348].

The data are not conclusive and the discrepancies found between the clinical studies are probably due to the different methodologies used for the assessment of the neuropsychological problem and the parameter examined [349, 350]. Another possible reason is that isolated xenoestrogens do not reflect the effect of exposure to a mixture [351].

1.3.7 Immunologic Effects

The relationship between autoimmune system and endogenous estrogen levels is well established [352]. Estrogens mediate their effects via estrogen receptors (nuclear isoforms and/or membrane receptor) in different cell types of the immune system (B cell, T cells, dendritic/macrophages, monocytes) [353]. They regulate T cytokine gene expression via ER-mediated pathways, either directly through EREs or indirectly through interaction of ER with other transcription factors including NF- κ B and AP-1 [354, 355]; NF- κ B response elements have been found in the promoter of several cytokine genes like IL-6, IL-10, TNF- α , IL-1 β , IL-12, and IL-2 [356–358]. Thus, estrogens by acting via their receptors and their crosstalk with other transcription factors in immune cells and organs can modulate immunological parameters [359].

Exposure to various classes of EDCs (such as DES, TCDD, PCBs, organochlorides) has been shown to cause immunosuppression and potential disease susceptibility [360, 361] both in humans and animals; dolphins exposed to EDCs (DDT, PCBs) showed impaired immune function [362]; decreased immune function and increased incidence of infections has also been observed among affected people [363]. After the incidence of Japan in 1968 and in Taiwan 1979 from contaminated rice oil, increased incidence of rheumatoid arthritis had been found, while the patients affected by the Yusho disease suffered respiratory infections for a long time [364–366]. Perinatal exposure to estrogenic compounds (i.e., dioxin) has been associated with increased incidence of infections (respiratory infections, otitis) [367], lower white blood cell count during the first years of life; reduced thrombocytes have also been reported to dioxin exposure [368, 369]. Allergic asthma has been associated with phthalate exposure since they induce enhancement of mast cell degranulation and eosinophilic infiltration which are important parts in the early inflammation phase [370]. Exposure to EDCs has also been associated with increased prevalence of thyroid antibodies [371].

PCDDs and related compounds may be related to immune diseases, such as atopic dermatitis. The effects of these compounds on the immune system were very clearly shown on the babies of young Japanese after the oil accident [372]. But important questions of clinical relevance of real-life exposure and identification of molecular targets that can explain the interactions remain to be answered.

1.3.8 Effects on Bones

Estrogens regulate skeletal homeostasis in both men and women. They enhance osteoblast bone formation, and their deficiency has been associated with increased bone resorption and osteoporosis [373, 374]. Exposure to environmental chemicals that are able to disrupt the hormonal equilibrium might represent another risk factor for this disease [254].

Estrogen receptors ER α and ER β have been found in both osteoclasts and osteoblasts. They are differentially expressed in the growth plate and mineralized bone, ER α is more highly expressed in cortical than in cancellous bone, and ER β is most evident at cancellous than cortical sites, suggesting that they may have different functions [375–377]. The role of ER α is clearer compared to ER β in bone formation [378].

The effect of estrogens in bone seems to be age and sex specific [379]. The importance of estrogens in males has been well documented from the fact that a loss of function mutation in ER α gene in a man has been connected with osteopenia [380].

In view of the important role of estrogen deficiency in osteoporosis, EDCs, since they interact with the ERs modulating the estrogen signalling pathway and altering estrogen metabolism, have been implicated in the pathogenesis of osteoporosis. Polychlorinated biphenyls, β -hexachlorocyclohexane, and 2,3,7,8-tetrachlorodibenzo-p-dioxin are among the compounds that have been associated with osteoporosis [381, 382], but the relation between organochlorine exposure and bone quality and osteoporosis is not clear; thus, further studies are needed [383].

1.3.9 Exposure In Utero and During Lactation

The exposure of the embryo to EDCs has been a major concern. The exposure to these chemicals begins from the first days of the in utero life since the placenta easily permits substances with low molecular weight to enter the fetal circulation, suggesting that these compounds can affect organogenesis [384]. Since these chemicals are lipophilic, they tend to accumulate in the adipose tissue of the pregnant women [385] and from there to be transferred to fetuses and infants through the placenta and breastfeeding. Since EDCs cross the placenta, the embryo is exposed to these chemicals and their metabolites; the neonate is further exposed to relatively high EDCs concentrations found in milk.

Even in the absence of epidemiological studies, concern over adverse effects of xenoestrogens is warranted given the unique vulnerability of the developing fetus and child [386]; placenta does not protect the embryo, and the embryo and young children lack the protective mechanisms an adult disposes, including liver metabolism, detoxifying mechanisms, and blood–brain barrier. Maternal exposure to phthalates has been connected to sex steroid hormone status in fetal and newborn

stages. Prenatal exposure to DES has been connected with adverse effects on the reproductive tract both in male and female offspring, including pseudohermaphroditism, genital malformations, and a reduced semen quality [218, 387]. Prenatal BPA exposure has the potential to alter neurodevelopmental, reproductive, and metabolic end points throughout the life span [388–390] at low doses, while at high doses fetal viability is compromised [391].

Although many efforts have been done to establish the lower threshold doses for toxicity, the issue has not been resolved for many products [392]; since there is no threshold in endocrine systems and no safe doses that exist. Actually an effect can be observed in lower doses, while in higher doses little or no effect is shown [393]. One of the reasons the data so far are conflicting is that many methods used lack sensitivity and precision [394]. Furthermore, it is important to take into consideration that the effects on the embryo also depend upon the developmental stage when the exposure happens and they are gender specific. Prenatal and perinatal stages are the most susceptible to the vulnerable effects, but the particular window of exposure during prenatal life makes the whole issue more complicated and the conclusive decisions about the harmful effects difficult. In this context, DDT use has been shown to increase preterm births, which is a major contributor to infant mortality [395].

EDCs can modify gene transcription disrupting the normal signalling systems that determine fetal development, and according to Colborn, they can impose a life sentence on the embryo [396]. Such an example could be considered the higher risk of overweight and obesity later in life that is associated with exposure to EDCs during development.

Further exposure of the neonate is with lactation. Maternal adipose tissue is catabolized and mobilized during lactation to provide 60 % of the fats in milk fat [397]. The adipose tissue catabolism results in the release of persistent EDCs that have been accumulated over the years (i.e., PCDs, DDT) to breast milk. It is interesting that the higher concentrations of the compounds in breast milk are consumed by the first child because the mother's fat stores are depleted with each subsequent child [398]. Human infants are exposed by breastfeeding, on a bodyweight basis, to doses of xenoestrogens that exceed the doses of adults by at least two orders of magnitude. In a group of breastfed children exposed to a PCB home environment, it was shown that the PCB concentrations were markedly increased with the duration of breastfeeding and were about five times higher than in the non-breastfed children [399].

An analysis of human milk revealed increased concentrations of several cosmetic chemicals (i.e., UV filters, synthetic musks, parabens); their concentrations were correlated to the frequency of use of the cosmetics [400]. Human breast milk levels of polybrominated organic compounds have increased 60-fold in the past 30 years and doubled in the last 5 years [401].

In view of the persistency of these chemicals, the ideal solution should be to place the mother and the infant in a protected environment, with no contamination of air, water, and food! But the practical solution for a pregnant and lactating woman should be to avoid consuming contaminated fish, such as fish from freshwater containing PCBs, and minimize exposure to products containing EDCs (cleaning products, paints, etc.). Single products may not be so hazardous but mixtures are [402].

The Greater Boston Physicians for Social Responsibility created a site for lactating women, where one can find information about the contaminated products with EDCs (<http://www.igc.org/psr/breastfeeding.htm>).

1.3.10 Risk Assessment

It is still not clear what the relationship between observed or assumed effects in humans and wildlife and exposure to man-made estrogenic compounds is; apart from the uses they are designed for, they may have unforeseen adverse effects or synergistic effects.

To make a prediction on human health consequences of the exposure to a range of substances which are suspected of interfering with the endocrine system (i.e., a risk assessment) is a very complex procedure since the following steps are important: identification of the substance, identification of the dose–response relationship, identification of a threshold dose which protects human health, quantification, and qualification of the adverse health effects. Thus, to predict a risk is a time-consuming procedure that needs knowledge of the mechanism of action, identification of the exposed population, and knowledge of the heterogeneity of the population, i.e., genetic predispositions, age and gender, diet, work exposure, and special conditions, such as pregnancy and lactation, route which exposures might occur and estimation of the magnitude, and duration and timing of the doses that people might receive as a result of the exposure. Together these studies indicate variable sensitivity to disruption by environmental chemicals during the developmental period and underscore the complexity of the mechanisms involved in their effects.

In order to examine the acute and chronic toxicity, carcinogenicity, genotoxicity, and effect on reproduction and development (organogenesis and fetal period) caused by EDCs, various pharmacological and toxicological tests have been developed [403].

A major problem in the case of EDCs is that one cannot make accurate predictions about the toxicity of xenoestrogens because of the absence of unambiguous dose–response relationships. Dose–response relationships are known for a number of individual EDCs; they present not a monotonic but a nontraditional dose–response curves such as an inverted “U” or even multiple “U”-shaped curves of their effects. Some effects are dose related, others dose independent or inversed according to the dose; some effects are reversible and others are not [404]. Due to the fact that the threshold model does not exist in the endocrine system, higher doses results are not predictive for lower doses effects [405], i.e., no dose effect has been found in dioxin that caused clear cell adenocarcinoma [406], while in the case of bisphenol A, very small doses have been shown to cause sex reversal in turtles [407]. In Belgium, the dioxin crisis was caused by only roughly 1 g of dioxin contained in 25 l of old PCB transformer oil [12, 13]. Furthermore, the fact that effects of many chemicals across species differ, the dose effect experiments in animals do not permit safe conclusions about the adverse effects in humans.

In view of the exposure to mixtures, the whole procedure becomes even more complex. Little is known about the interactive effects of mixture; the consequences

on human health are multifactorial depending upon the concentration (ppm) of a certain compound in the product: i.e., in cosmetics, the volume of cosmetic used (ml) per application, times of applications per day, and rate of absorbance depending upon route of administration should be determined.

Lower observed effect levels (LOEL) and no effect levels (NOEL) are two of the indexes used, but for many EDCs, the current tests do not provide evidence for the existence of a NOAEL function. Synergistic or antagonistic effects are for the most part unknown; compared to prescribed medication, a medication's license can be withdrawn upon reports on adverse effects, while it is very difficult to revoke the license of a non-pharmaceutical chemical because the data will be partially confounded by the mixture problem and means of effect is almost always lacking.

A major problem is that the exposures to EDCs are involuntary, often chronic in duration; nontoxic effects are known a priori, and the appearance of a new illness or pathology is documented only after high-dose disaster happens [408]. Then, while acute, subacute, and chronic exposure and reproductive and genotoxicity tests in animals are routine for pharmaceuticals for the majority of the chemicals, this does not apply. In general, any chemical with a production level of <1,000 tonnes/year in a country will require little testing. Thus, a considerable amount of research is still required to ascertain the scope and the seriousness of endocrine disruption, including confirmation of epidemiological studies.

The exposure levels that could have deleterious health effects are somewhat difficult to determine and are actively debated [409, 410], but a most critical point is the concentrations at the target organs are most critical. Thus, data on breakdown, excretion, and bioaccumulation are very essential.

Newer methods have been developed in order to measure the amount of EDCs in biological fluids and tissues (i.e., exposure to the phthalates in perfumes and their concentration in hair) [411, 412]. In a recent study, by use of a bioassay, the exposure to EDCs, estimated in total 17 β -estradiol equivalents (EEQs), was found increased in occupational exposure to pesticides, disinfectants, and exhaust fumes [413].

Recently, a scalable and statistical method has been developed, in an effort to predict known and novel associations of several chemicals with prostate, lung, and breast cancers, using publicly available data (e.g., on estradiol and bisphenol) [414].

1.4 Importance of Identification of Compounds with Estrogenic Activity

The majority of the EDCs are compounds structurally unrelated, and the prediction of the estrogenic activity is very difficult, and only in a very small number of chemicals can be done. Thus, in view of the continuously increasing number of these compounds, governmental agencies were forced to examine the whole issue [415] setting as first target the identification of the compounds. Several screening tests have been developed; the principal requirement is to assess the potential of

these compounds to interact with the endocrine system of man and wildlife in order to anticipate adverse effects and then to elucidate the mechanism of action. By the *in vitro* screening methods, the affinity to the nuclear ER (α , β) was evaluated [416–418]. From 58,000 synthetic compounds that were checked in 2002, including synthetic estrogens, natural products, several plasticizers, commercial chemicals, and impurities, 6,903 were found to dispose weak estrogenic activity (at least 1,000-fold less compared to E₂) [419]. Among the bioassays, the E-SCREEN assay is a simple, fast, reproducible, reliable, and quite sensitive assay [109]; it has allowed the identification not only of the chemicals with estrogenic activity but their discrimination into estrogen full and partial agonist and antagonist compounds by measuring the cell proliferation on cell lines as well. Other assays that have been used are the binding receptor assay [109, 420], the cell proliferation assay [421], and the gene expression assay [422], but none of them can distinguish between agonist and antagonist. Although useful, *in vitro* assays suffer from problems associated with the absence of effective means to metabolize chemicals. Thus, the big problem is that EDCs must be evaluated in intact organisms; *in vitro* assays are of value just for evaluation of mechanisms of action or prescreening chemicals for potential endocrine-disrupting properties and for setting priorities for in-depth *in vivo* testing. Big efforts have been made in the USA, EU, Japan, and OECD in establishing appropriate tests and to harmonize the strategy efficiently. A two-tiered program is currently running that includes a combination of *in vitro* and *in vivo* assays in order to identify and classify substances in relation to their potential to interact with the endocrine systems (tier 1) and then to develop concentration response curves in animal models (tier two, under validation) [423]. The ultimate goal is to clarify the biological responses of these compounds in whole organism, but in order to test approximately 80,000 compounds, millions of animals should be sacrificed. In view of the ethical and economical problems, animals have been replaced by cell lines or simplified systems (i.e., yeast); compared to the complexity of an organism, often the conclusions drawn are different from *in vivo* experiments [424, 425].

Pretests of new chemicals before they are marketed and a group classification of the EDCs based on their chemical biochemical and biological activities should be done; the problem is difficult to resolve since there are no adequate tools to test complex mixtures.

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Part II
Methodology of Measuring
BPA and Its Effects

Chapter 2

Analytical Methods for Determination of Bisphenol A

Dimitra Voutsas

2.1 Introduction

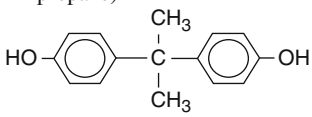
Bisphenol A (2,2-bis-(4-hydroxyphenyl)propane, BPA) is an industrial chemical used in a wide range of applications. It is formed through an acid-catalyzed condensation reaction of phenol and acetone [1]. The chemical structure and the physico-chemical properties of BPA are shown in Table 2.1. BPA is a moderately water-soluble compound, with low volatility that is easily biodegraded.

Bisphenol A is a chemical manufactured in large quantities. It is estimated that 1,150 tonnes/year are produced and used in Western Europe. Almost 96 % of BPA is used as a monomer for the production of polycarbonate and epoxy resins. Other applications include its use as stabilizing agent in plastics, as antioxidant in tire production, and as basic chemical in the production of certain flame retardants. The BPA-based materials are used in food and beverage containers, protective coating, automotive lenses, optical lenses, adhesives, powder paints, building materials, compact disks, thermal paper, paper coatings, dental, surgical, and prosthetic materials [3, 4]. The production and extensive use of these materials result in the release of this compound into the environment during processing, handling, and transportation of final products. It was estimated that 39.5 % of the total environmental release of BPA comprised total air release, 1 % water release, 54 % land release, and 5.4 % underground injection [3].

Various *in vitro* and *in vivo* assays showed that BPA presents estrogenic activity, and consequently, it is considered as important organic pollutant [3]. BPA may cause a variety of adverse effects on reproduction and development of exposed organisms, being more striking and irreversible during embryonic development. These effects may occur even at doses of BPA well below those showing adverse effects in routine toxicity studies [3–7].

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Table 2.1 Physicochemical properties of BPA [2]

CAS no.	80-05-7
Organic compound	Bisphenol A (2,2-bis-(4hydroxyphenyl) propane)
Chemical structure	
Formula	C ₁₅ H ₁₆ O ₂
Molecular weight	228.29 g/mol
Boiling point	398 °C at 760 mmHg
Melting point	153–157 °C
pKa	9.6–11.3
Water solubility	120–300 mg/L
Vapor pressure	0.2 mmHg at 170 °C
Log Kow	2.20–3.82
Henry's constant	1.0 × 10 ⁻¹⁰ atm m ³ /mol

The US-EPA, under the Toxic Substances Control Act, intends to consider initiating rulemaking to identify BPA on the Concern List as a substance that may present an unreasonable risk to the environment on the basis of its potential for long-term adverse effects on growth, reproduction, and development in aquatic species at concentrations similar to those found in the environment [4]. BPA is candidate to be among the first substances to go through Registration, Evaluation, Authorization, and Restriction of Chemicals (REACH) in EU registration (EU Regulation No 1907/2006). Canada was the first country that has classified BPA as toxic substance and announced restriction of imports, sales, and advertising of polycarbonate baby bottles containing BPA. Recently, the European Commission published a new directive (2011/8/EU) to restrict bisphenol A in feeding bottles that are intended for use by infants under the age of 12 months [8]. According to this directive, member states are required to prohibit the manufacture of polycarbonate feeding bottles containing BPA as well as their import and sale in EU.

The study of the occurrence of BPA in various environmental compartments, in food, in dental materials, and in biological fluids contributes to the knowledge on the environmental fate of this compound, the possible pathways of exposure, the biotransformation mechanisms, and the possible risks. In order to identify and determine trace levels of BPA in complex matrices, sensitive analytical methods are required.

A number of analytical methods have been developed for the determination of BPA. The general scheme of analysis usually comprises isolation from samples through extraction, cleanup steps, and determination by employing a sensitive analytical technique. The major problems associated with the analysis are possible loss or contamination during sampling and storage, the need of preconcentration and, possibly, of cleanup, as well as the need for highly efficient separation procedures

and selective detection techniques. Reliable analytical procedures require detailed method validation and careful evaluation. In addition, sampling and sample preparation should be considered integrally with the characterization of an analytical procedure.

2.2 Sampling and Storage

The first step in the measurement of BPA involves representative sampling and maintaining sample integrity prior to analysis. The sampling strategy should reflect the known or expected variability of the system.

All the equipment that may come into contact with sample or the extract should be free from interfering compounds. The sampling containers should be made of materials that do not change the sample during the contact time. Plastics and other organic materials should be avoided during sampling, sample storage, or extraction. Glass brown bottles with glass stoppers or with PTFE-lined screw caps, carefully precleaned, are recommended for sampling and storage of samples. Rinsing with acetone is recommended for all glassware used in the analysis. Alternatively, non-volumetric glassware may be heated to at least 200 °C for a minimum of 2 h. The samples should be analyzed as soon as possible; otherwise, they can be stored at 2–5 °C for 2 weeks [9].

2.3 Extraction Techniques

Sample preparation is an important stage in the analytical process when trace analyte determination is needed. The analysis of pollutants at low concentrations in complex matrices requires the elimination of interferences and the reduction of final extract volumes to attain higher preconcentration of target analytes. Generally these pretreatment methods are necessary in order to improve the detection and quantification limits, avoid matrix implications, limit background noise, and extend the life of the analytical column.

The analysis of BPA in environmental, food, and biological liquid samples employs a wide range of sample extraction techniques. Solid-phase extraction is frequently used for isolation and preconcentration of BPA (Tables 2.2, 2.3, 2.4, and 2.5). Moreover, other techniques such as the traditional liquid–liquid extraction, solid-phase microextraction, and stir bar sorptive extraction have been used.

Liquid–liquid extraction (LLE) is a reliable technique for the extraction of BPA from liquid samples. LLE is proposed for the recovery of BPA along with other compounds (NP, tOP, NPE1EO, NPE2EO) from environmental waters in ASTM D7065-06 method. LLE has the advantage of low equipment costs, but there is

Table 2.2. Analytical methods and concentration range of BPA in dental materials

Dental materials	Samples analyzed/ pretreatment	Analytical method	LOD	Concentrations of BPA	Reference
Core build-up materials	Eluates in ethanol 75 %	LC-MS/MS Column: CC 125/4 Nucleodur 100-5 C18 Mobile phase: 0.1 % formic acid/ acetonitrile	0.5 µg/mL	BDL-6.14 µg/mL	Brenn- Struckhofova and Cichna- Markl [43]
Orthodontic adhesives	Eluates in alcohol 99 %	Diagnostic ions: <i>m/z</i> 227 HPLC-UV/Vis Column: C18 Mobile phase: acetonitrile/water (60:40 v/v) Wavelength: 228 nm	0.1 ppm	BDL	Fontana et al. [57]
Dental sealants	Eluates in ethanol 95 %	HPLC-UV Column: Nova Pak C18 Mobile phase: A acetonitrile/water (50:50 v/v) B acetonitrile Wavelength: 215 nm	0.0001 µg/ mg	BDL	Cunha et al. [58]
Dental sealants	Eluates in distilled water	HPLC-UV/Vis Column: C18 resolved column Mobile phase: isocratic 70 % methanol Wavelength: 215 nm	–	BDL	Salafranca et al. [59]
Orthodontic adhesives	Eluates in distilled water SPE (Oasis HLB) Elution with acetone Derivatization with BSTFA	GC-MS, EI, SIM Column: 5 % diphenyl-95 % dimethyl polysiloxane Diagnostic ions: <i>m/z</i> 357.2, 358.2 Internal standard BPA- <i>d</i> 16	2.3 ng/L	0.116–2.90 µg/L	Maragou et al. [44]

Composites/sealants	Vigorous agitation with distilled water (37 °C) at various pH values (1–12)	HPLC-UV Column: C18 Mobile phase: gradient A acetonitrile/water (1:1 v/v) and B acetonitrile Wavelength: 280 nm GC-MS Column methyl silica Diagnostic ions: <i>m/z</i> 213, 228 HPLC-UV/Vis Reversed phase column Mobile phase: water/acetonitrile (43/57) Wavelength: 215 nm GC-MS Column: methyl silicon DB-1 GC-MD, NCI Column: 5 % phenyl-1-methyl-polysiloxane Diagnostic ions: <i>m/z</i> 407, 299 Internal standard ¹³ C ₁₂ -BPA	0.20 µg/mL	0.3–116.1 µg/mL in polymerized materials <0.2–179.5 µg/mL in unpolymerized materials	[60]
Dental sealants/ adhesive resins	Eluates in water/acetonitrile (43/57)		100 pg	BDL	[60]
Dental sealants	Saliva Urine SPE C18 Elution with methanol Derivatization: pentafluorobenzyl bromide		0.1 ng/mL	0.17–96.2 ng/mL 0.6–112.2 ng/m	Kawaguchi et al. [46]
Composite resins	Saliva	ELISA "EIKEN" kit		0.3–100 ng/L	Shao et al. [39]

(continued)

Table 2.2 (continued)

Dental materials	Samples analyzed/ pretreatment	Analytical method	LOD	Concentrations of BPA	Reference
Dental sealants	Saline solution (37 °C) Saliva Serum SPE C18 Elution with acetonitrile	HPLC-FLD Column: Supelcosil LC-C18	5 ppb	BDL BDL	Chang et al. [45]
Restorative composites Dental sealants	Eluates in ethanol Saliva	Mobile phase: acetonitrile/water (50:50 v/v) Exc/Emiss vv: 278/315 nm HPLC-diode array Column: S5 ODS Mobile phase: gradient A acetonitrile/ water (1:1) and B acetonitrile Wavelength: 235 nm GC-MS confirmation		BDL-84.4 µg/100 mg 3.5-30 µg/mL	Kawaguchi et al. [61]

BDL below detection limit

Table 2.3 Analytical methods and levels of BPA in environmental samples

Samples/Country	Pretreatment	Analytical method	LOD	Concentration of BPA	Reference
River water	LLE with dichloromethane	GC-MS, EI, SIM	0.5 pg/ μ L	17–776 ng/L	Heemken et al. [21]
Sea water	HPLC clean up	Column: 5 % phenylmethyl silicon		BDL–249 ng/L	
Germany	Derivatization with HFBA	Diagnostic ions: <i>m/z</i> 315, 331, 407			
Freshwater	Filtration	Internal standard BPA- <i>d</i> 16			
		GC-MS, EI, Full-scan	20 ng/L	BDL–1, 924 ng/L	Quednow and Püttmann [22]
		Column: BP-X5			
Germany	SPE (Bod Elute OOL)	Diagnostic ions: <i>m/z</i> 213, 228			
Surface waters	Elution with methanol/acetone/nitrile				
	SPE (LiChrolut)	GC-MS/MS, EI	0.1 ng/L	0.5–410 ng/L	Fromme et al. [23]
Sewage effluents	Elution with acetone	RP-HPLC-FLD (Exc/Emiss wv 228/310 nm)			
Germany		Mobile phase: gradient A hexane	2.0 ng/L	18–702 ng/L	
Surface waters	Filtration	B hexane/methanol/isopropanol (40/45/15)			
		GC-MS, EI, SIM	2.4 ng/L	9–76 ng/L	Voutsas et al. [24]
		Column: DB-5			
Switzerland	SPE (Oasis HLB)	Diagnostic ions: <i>m/z</i> 357.2, 358.3			
	Elution with acetone	Internal standard BPA- <i>d</i> 16			
	Derivatization with MSTFA/2 % Sylon BTZ				
Surface water	Filtration	LC-MS/MS, ESI, NI, MRM	1.1 ng/L	2–46 ng/L	Jonkers et al. [25]
		Column: 100 RP18ec			
Wastewaters	SPE (Oasis HLB)	Mobile phase: gradient A water 4 mM ammonium acetate B methanol		1.3–1, 640 ng/L	
		Precursor ion: <i>m/z</i> 227.02			
		Product ion: <i>m/z</i> 211.8			
Switzerland	Elution with MTBE/2-propanol	Internal standard BPA- <i>d</i> 16			

(continued)

Table 2.3 (continued)

Samples	Pretreatment	Analytical method	LOD	Concentration of BPA	Reference
Wastewaters	SPE (Oasis HLB) Elution with methanol-diethyl ether (10:90 v/v)	GC-MS, EI, SIM Column 95 % dimethyl-5 % phenylpolysiloxane Identification ion: <i>m/z</i> 358 GC-MS/MS, EI Precursor ion: <i>m/z</i> 358 Product ions: <i>m/z</i> 191, 267, 357	0.5 ng/L	450 ng/L	Jeannot et al. [26]
France	Derivatization with BSTFA				
Surface waters	Filtration SPE (Oasis HLB)	GC-MS, EI, SIM Column: 5 % diphenyl-95 % dimethyl polysiloxane	2.3 ng/L	15–460 ng/L 15–56 ng/L	Arditsoglou and Voutsas [27, 28] Pothitou and Voutsas [29]
Coastal waters	Elution with acetone	Diagnostic ions: <i>m/z</i> 357.2, 358.2		15–2, 358 ng/L	Arditsoglou and Voutsas [30]
Wastewaters	Derivatization BSTFA	Internal standard BPA- <i>d</i> 16			
Greece					
Surface waters	Decantation	LC-MS/MS, RP, ESI, API Column: Synergi Polar RP Mobile phase: water/acetonitrile Precursor ion: <i>m/z</i> 227 Product ions: <i>m/z</i> 133, 212	2 ng/L	3–175 ng/L	Loos et al. [31]
Wastewaters	SPE (Oasis HLB)				
Italy-Belgium	Elution with ethanol/acetone/ ethyl-acetate (2:2:1)				

Lagoon water	SPE (ENVI-18)	HPLC-MS, ESI Column: C8-2	1 ng/L	BDL-145 ng/L	Pojana et al. [32]
Italy	Elution with acetonitrile, methanol, water	Mobile phase: gradient A acetonitrile B water Internal standard: BPA- <i>d</i> 16			
Precipitation	LLE with dichloromethane	LC-MS, ESI, NI	5 ng/L	bdl-357 ng/L	Peters et al. [33]
The Netherlands		Column: Symmetry C18			
Surface waters	Filtration	GC-MS Column: SGE BPX5	14 ng/L	BDL-330 ng/L	Belfoid et al. [34]
		Diagnostic ions: <i>m/z</i> 357 Internal standard: BPA- <i>d</i> 16			
The Netherlands	SPE (SDV-XC disks)	LC-MS/MS, NI, MRM	2 ng/L	0.15-1.55 µg/L	Mauricio et al. [35]
Wastewater	Elution with methanol Derivatization with SIL A Filtration	Column: Purospher STAR RP-18 Mobile phase: gradient A methanol B water Precursor ion: <i>m/z</i> 227 Product ions: <i>m/z</i> 133, 211 Internal standard: oxybenzoic acid ELISA	5 µg/L		
	SPE (Oasis HLB)				
Portugal	Elution with dichloromethane	HPLC-MS, ESI, NI	0.09 µg/L	BDL-2.97 µg/L	Céspedes et al. [36]
Surface waters	Filtration	Column: 100RP-18 Mobile phase: gradient A methanol B water Diagnostic ions: <i>m/z</i> 227 Internal standard: 4-heptylpheno		0.06-1.51 µg/L	
	SPE (Lichrolut RP-18)				
Wastewater	Elution with acetonitrile				
Spain					

BDL below detection limit

Table 2.4 Analytical methods and levels of BPA in food samples

Samples	Pretreatment	Analytical method	LOD	Concentration of BPA	Reference
Bottled water	LLE with dichloromethane Derivatization BSTFA	GC-MS, EI Column: 5 % diphenyl-95 % dimethyl polysiloxane Diagnostic ions: m/z 357.2, 358.2 Internal standard BPA- <i>d</i> 16	2.3 ng/L	3.5–150 ng/L	Nathanson et al. [12]
Bottled water	SPE (OASIS HLB or C18) Elution two steps A dichloromethane/hexane (4:1 v/v) B ethanol/dichloromethane (9:1 v/v) SPE (OASIS HLB)	GC-MS, EI Column: HP 5MS Diagnostic ions: m/z 213, 119, 228 Internal standard: 4nNP	0.005 µg/mL	bdl-0.011 µg/L	Inoue et al. [76]
Mineral water	Elution with methanol/ dichloromethane SPE (C18)	LC-MS/MS, ESI, NI, MRM Column: A symmetry C-18 Mobile phase: A methanol and B water Precursor ion: m/z 227.2 Product ions: m/z 93.1, 133.4, 212.4	0.01 ng/L 0.60 ng/L	BDL	Gallart-Ayala et al. [77]
Soda beverages Canned soft drink products	SPE (C18) Elution acetonitrile-water (1:1 v/v) DLLME	GC-MS, EI Column: HP 5MS Diagnostic ions: m/z 213, 228, 270, 312 Internal standard: BPA- <i>d</i> 16	27–74 ng/L	0.032–4.5 µg/L	Hennion [51]
Soft drinks/beers	Derivatization acetic anhydride	GC-MS, EI Column: HP 5HT, HP 5MS Diagnostic ions: m/z 213, 228, 270, 312 Internal standard: BPA- <i>d</i> 16	5 ng/L	BDL-4.7 µg/L	Joskow et al. [17]

Milk	Mixing with C18	LC-MS/MS, ESI, NI, MRM Column: A symmetry C-18 Mobile phase: A methanol and B water Precursor ion: <i>m/z</i> 227.2 Product ions: <i>m/z</i> 93.1, 133.4, 212.4	0.1 µg/kg	bdl-0.49 µg/kg	Qubiña et al. [78]
Milk	SPE clean up	GC-MS, EI Column: HP 5MS Diagnostic ions: <i>m/z</i> 213, 119, 228 Internal standard: 4nNP	0.15 µg/kg	0.28–2.64 µg/kg	[79]
Milk infant formula	Dilution with methanol	GC-MS, EI Column: HP 5HT, HP 5MS Diagnostic ions: <i>m/z</i> 213, 228, 270, 312 Internal standard: BPA- <i>d</i> 16	60 ng/L	BDL-0.40 µg/L	Joskow et al. [17]
Wine	Derivatization acetic anhydride SPE (C18)	HPLC-FLD Column: LiChrosper 60 Mobile phase: Acetonitrile/water (30:70 v/v) Exc/Emiss wv: 275/305 nm	0.1 ng/mL	bdl-2.1 ng/mL	Mohapatra et al. [80]
Food simulants – water and 3 % w/v acetic acid (leachates from baby bottles)	Elution with acetonitrile Clean up sol-gel	HPLC-FLD Column: ODS-2 C18-bonded Mobile phase: methanol/water (70:30 v/v) Exc/Emiss wv: 255/317 nm	1.8 ng/mL	2.4–14.3 µg/kg	[81]
Food simulants – hot water (leachates from commercially available bottles)	SPME Derivatization with BSTFA/1 % TMCS	GC-MS, EI, SIM Column DB-5MS Diagnostic ions: <i>m/z</i> 357, 372 Internal standard BPA- <i>d</i> 16	0.4 ng/L	0.7–78.5 µg/L	

BDL below detection limit

Table 2.5 Analytical methods and levels of BPA in urine samples

Samples	Pretreatment	Analytical method	LOD	Concentration of BPA	Reference
Human urine	HF-LPME	GC-MS, EI, SIM Column DB-5MS	0.02 ng/mL	0.1–0.4 ng/mL	Kawaguchi et al. [46]
	In situ derivatization with acetic acid anhydride	Diagnostic ions: <i>m/z</i> 213, 228 Internal standard BPA- ¹³ C12 Analyte determined: BPA-glucuronide			
Human urine	Treatment with β -glucuronidase/sulfatase	LC-MS/MS, APCI, Column: a symmetry C-18	0.4 μ g/L	0.5–15.9 μ g/L (10–95th percentiles)	Calafat et al. [47]
	Online SPE	Mobile phase: A methanol and B water Precursor ion: <i>m/z</i> 227.2 Product ions: <i>m/z</i> 93.1, 133.4, 212.4 Analyte determined: total BPA		2.6 μ g/L (geometric mean)	
Human urine	Treatment with β -glucuronidase SPE	HPLC-ED Column: Inertsil ODS-3V Mobile phase: phosphate buffer/ethanol/ acetonitrile (11 : 1 : 7 v/v/v)	0.2 ng/mL	1.92 ng/mL (total BPA)	Fukata et al. [48]
	Dilution with methanol	Analyte determined: total and free BPA Confirmation with LC-MS/MS, MRM Analyte determined: BPA-glucuronide			
Human urine	Online SPE cleanup	LC-MS/MS, APCI, NI with column switching Column: LiChrospher RP-18	0.4 ng/mL	BDL–11.5 ng/mL (10–95th percentiles)	Ye et al. [49]
	With and without glucuronidase and sulfatase treatment	Mobile phase: A methanol and B water Precursor ion: <i>m/z</i> 227 Product ions: <i>m/z</i> 133, 212 Analyte determined: BPA, BPA-glucuronide		3.5 ng/mL (mean)	

Human urine	Treatment with β -glucuronidase	HPLC-FLD Column: symmetry C-18 Mobile phase: A acetonitrile/methanol +0.1 mM octanesulfonic acid and B acetonitrile/water+0.1 mM octanesulfonic acid Exc/emm wv 275/300 nm Analyte determined: total BPA	0.28 ng/mL 0.85–9.83 ng/mL (men) 1.00–7.64 ng/mL (women)	Kim et al. [50]
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BDL below detection limit

concern associated with the relatively large volumes of organic solvents used. The selection of suitable solvent is based on its extraction efficiency and selectivity, its inertness, and its boiling point. Other factors which are usually considered are the toxicity of the extracting solvent, relative densities of the two phases, and its tendency to form emulsions [51, 52]. The solvents that have been used for the recovery of BPA are dichloromethane, ethyl acetate, and chloroform. Repeated extractions are necessary to ensure complete recovery of BPA [37, 53]. Solvent microextraction techniques representing the miniaturization of liquid–liquid extraction have received major attention, because they resulted in a more efficient analyte enrichment, faster sample preparation, and lower solvent consumption. For the determination of BPA, various techniques such as dispersive liquid–liquid microextraction (DLLME), vortex-assisted liquid–liquid microextraction (VALLME), and ultrasound-assisted emulsification microextraction (USAEME) have been recently introduced [54–58].

Solid-phase extraction (SPE) is the technique usually employed for the recovery of BPA from liquid samples (Tables 2.2, 2.3, 2.4, and 2.5). SPE is the extraction proposed for isolation of BPA from waters in the methods ISO 18857-2:2009 and ASTM D7574-09. SPE does not require large volume of toxic organic solvents, analysis times can decrease significantly, and online and/or automated procedures are easily designed. SPE can be performed with commercially available extraction cartridges with the suitable sorbent [51, 52]. The divinylbenzene/*N*-vinylpyrrolidone copolymer (Oasis HLB) has been the most used sorbent (Table 2.3). Chemically bonded reversed-phase silica (C18) and PS-DVB have been also proposed as SPE sorbents.

Solid-phase microextraction (SPME) is another technique for isolation of BPA from various types of samples. The advantages of the method are the absence of solvents and the relatively small volumes of sample required compared to other methods. SPME utilizes a small fused-silica fiber, coated with a suitable polymeric stationary phase for isolation and preconcentration of analyte. The extraction can be performed with direct immersion of the fiber to the liquid sample or through headspace by suspending the fiber in the vapor phase. The extraction efficiency of SPME depends on many factors such as the matrix, the stationary phase, the exposure time, and the desorption temperature [51, 52]. SPME followed by GC-MS has been applied to the determination of BPA in aqueous samples. Different stationary phases have been used for isolation of BPA such as polydimethylsiloxane (PDMS), polyacrylate (PA), carbowax/divinylbenzene (CW/DVB), and carboxen/polydimethylsiloxane (CAR/PDMS) with PA showing better extraction efficiency [59, 60].

A stir bar sorptive extraction (SBSE) is another technique lately used for the isolation of various analytes from environmental and biological samples. SBSE uses a stir bar into a sealed glass tube that is coated with suitable stationary-phase sorbent. The stir bar can be immersed into the liquid sample or can be held in the headspace above the liquid sample. Removal of the analyte from the bar is achieved by GC thermal desorption or by a suitable solvent. Stationary phase such as PDMS or β -cyclodextrin (PDMS/ β -CD) has been used for the extraction of BPA from waters, saliva samples, and biological fluids (human urine and plasma) [61–63].

The demand of highly selective sorbent materials for the determination of traces contaminants in complex samples leads to development of molecularly imprinted polymers (MIPs). MIPs are synthetic polymers having molecular recognition ability for a target analyte. MIPs offer stability against organic solvents, strong acids or bases, and elevated temperatures and pressures. Furthermore, they permit larger sample volumes and reusability [64]. MIPs for isolation of BPA from various types of samples have been developed and used as SPE sorbent materials, as SPME fibers coatings, and as online pretreatment devices [65–69].

2.4 Analytical Techniques

Chromatographic based analytical techniques are used to identify and quantify BPA in environmental, food, and biological samples. Gas chromatography and liquid chromatography coupled with mass spectrometry are the most widely employed techniques.

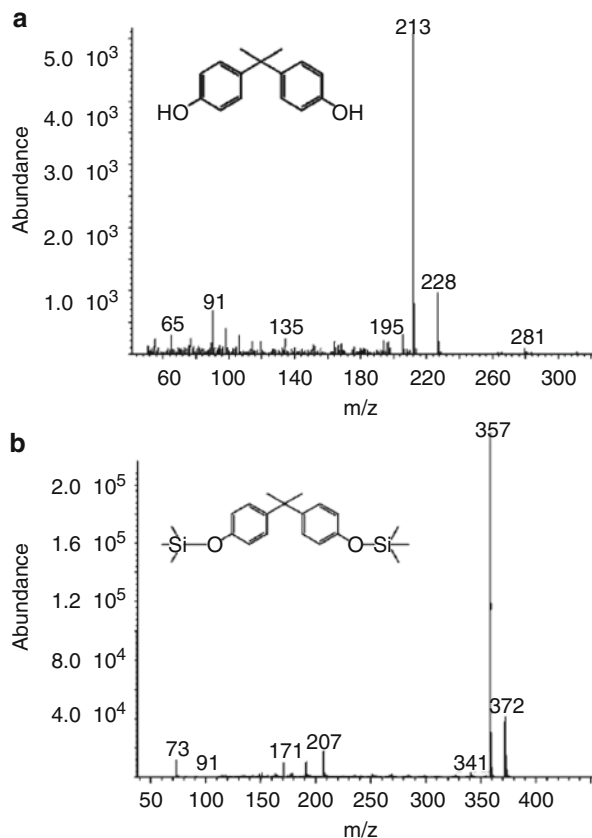
2.4.1 Gas Chromatography-Mass Spectrometry

Gas chromatography coupled with mass spectrometry (GC-MS) has often been applied for determination of BPA (Tables 2.2, 2.3, 2.4, and 2.5). The analytical columns are phenyl-methyl polysiloxane. The electron impact (EI) is the most common ionization source, although negative ion chemical ionization (NICI) has been applied [70]. The EI mass spectra of BPA are shown in Fig. 2.1. The spectra are characterized by a molecular ion at m/z 228 ($[\text{C}_{15}\text{H}_{16}\text{O}_2]^+$), whereas the most abundant fragment ion at m/z 213 ($[\text{C}_{14}\text{H}_{13}\text{O}_2]^+$) corresponds to the loss of methyl group. An alternative minor fragmentation pathway involves the loss of one of the aryl groups from the molecular ion to give *tert*-benzylic carbocation ($[\text{C}_9\text{H}_8\text{O}]^+$) at m/z 135 and the subsequent loss of methane to give a fragment ion at m/z 119. GC-MS is employed for the determination of BPA in environmental waters along with other compounds (NP, tOP, NPE1EO, NPE2EO) in ASTM D7065-06 method. This method adheres to selected ion monitoring mass (SIM) spectrometry, but full-scan mass spectrometry has also been shown to work well under these conditions. The method detection limit for BPA is 0.9 $\mu\text{g/L}$.

Tandem mass spectrometry (GC-MS/MS) can be also used for determination of BPA. In this case the most intense fragment ions on the EI spectrum are those corresponding to the loss of a methyl group ($[\text{M}-15]^+$) and part of the aliphatic chain ($[\text{M}-83]^+$) and are considered as precursors ions resulting in daughter ions (198, 119, 165) which are indicative of the structure of the compound [71].

In order to overcome the drawbacks of low volatility polar characteristics of BPA and improve the selectivity, sensitivity, and performance of gas chromatography, a derivatization procedure is usually employed. Derivatization approaches such as silylation, acetylation, and methylation have been used for determination of low

Fig. 2.1 Mass spectra of bisphenol A (a) and bisphenol A trimethylsilylated derivative (b) by employing GC-EI-MS [60]



concentrations of BPA in various matrices. Silylation is the method most commonly applied (Tables 2.2, 2.3, 2.4, and 2.5) because the derivatization reaction is fast and quantitative and yields thermally stable and highly volatile derivatives. The most popular silylation reagent is *N*-*O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA). Moreover, BSTFA containing 1 % of trimethylchlorosilane (TMCS) or *N*'-*N*'-methyl-(trimethylsilyl)trifluoroacetamide (MSTFA) has been also used. The EI mass spectra and the fragmentation pattern of BPA silylated derivative are characterized by the base peak at m/z 357 that corresponds to the loss of methyl group ($[C_{20}H_{29}Si_2O_2]^+$) from the molecular ion ($[C_{21}H_{32}Si_2O_2]^+$) at m/z 372 (Fig. 2.1). The ISO 18857-2:2009 method specifies GC-MS determination of bisphenol A and selected alkylphenols and their ethoxylates in drinking, ground-, surface, and wastewaters following solid-phase extraction and derivatization with MSTFA.

Acetylation of hydroxyl groups of BPA by using acetic anhydride or trifluoroacetic anhydride as derivatizing reagents is also used. Fluoro-derivatizing reagents are also used to analyze phenolic compounds. The mass spectra of *O*-bis(trifluoroacetyl) derivatives of BPA have the base peak at m/z 405 that corresponds to the loss of methyl group $[M-15]^+$ from molecular ion at m/z 420. The mass spectra of diacetate BPA have the base peak at m/z 213 [53, 62, 72, 73]. In order to

minimize possible interferences or loss of BPA in situ, derivatization has been proposed [46, 61].

2.4.2 *Liquid Chromatography*

Liquid chromatography (LC) has been employed for the determination of BPA in various samples (Tables 2.2, 2.3, 2.4, and 2.5). LC is usually carried out in reverse-phase C18 columns. The detectors that have been used are ultraviolet (UV), fluorescence (FLD), electrochemical (ED), and mass spectrometry (MS). Solvents in mobile phase include water, acetonitrile, and methanol. Gradient elution is usually performed since BPA is often determined simultaneously with other phenolic endocrine-disrupting compounds.

2.4.2.1 **Liquid Chromatography-Ultraviolet Detection**

Reverse-phase liquid chromatography coupled with UV detector (HPLC-UV) at various wavelengths (215–280 nm) has been applied for the determination of BPA (Table 2.2). UV detector exhibits relative low sensitivity for BPA. This method offers poor selectivity for the determination of traces of BPA in complex matrices such as environmental and biological samples [72].

2.4.2.2 **Liquid Chromatography-Fluorescence Detection**

Reverse-phase liquid chromatography coupled with fluorescence detector (HPLC-FLD) has been also employed for the determination of BPA (Tables 2.4 and 2.5). After excitation at 275 nm, BPA shows fluorescence at emission wavelengths range 300–320 nm. The fluorescence intensity is much higher in organic media (methanol and acetonitrile), and thus, the sensitivity is dependent on the composition of the mobile phase [53].

2.4.2.3 **Liquid Chromatography-Electrochemical Detection**

Liquid chromatography coupled with electrochemical detection (HPLC-ED) has been used for the determination of BPA (Table 2.5). It is a sensitive and selective method that presents low detection limits which are 3,000 and 200 times lower than those obtained with UV and FLD detectors [74]. The comparatively high selectivity of electrochemical detector is due to the electroactivity of the phenolic groups of BPA. The pH and electrolyte content of mobile phase influence the electron transfer rate constants, so they have to be optimized in order to get maximum sensitivity. Isocratic elution is preferred otherwise rather large equilibrium time is required [53].

2.4.2.4 Liquid Chromatography-Mass Spectrometry

Liquid chromatography coupled with mass spectrometry (LC-MS, LC-MS/MS) is a valuable tool for determination of BPA since it combines high selectivity and sensitivity (Tables 2.2, 2.3, 2.4, and 2.5). Mass spectrometry offers structural confirmation resulting in higher confidence in identification than LC-UV, LC-FLD, and LC-ED. Moreover, LC/MS has the advantage over GC-MS that derivatization is not required. The most common ionization sources in LC-MS are electrospray ionization (ESI) and atmospheric pressure chemical ionization (APCI), both in negative mode. ESI is more frequently used than APCI because it generally provides better sensitivity. The mass spectra of BPA exhibit the ion m/z 227 that corresponds to deprotonated molecule ion $[M-H]^-$. Under MS/MS conditions the characteristic fragments of product ion mass spectra are shown in Fig. 2.2. The most abundant fragment at m/z 212 can be attributed to the loss of methyl group $[M-H-CH_3]^-$. Another product ion at m/z 133 results from the cleavage of the hydroxybenzyl group $[M-H-C_6H_5OH]^-$ [70, 75–77]. Liquid chromatography coupled with tandem mass spectrometry (LC-MS/MS) is proposed for the determination of BPA in environmental waters according to the ASTM D7574-09 method. The detection limit of this method is 5 ng/L.

2.5 Immunoassays

Immunoassays are analytical tests that utilize antibodies to selectively bind organic compounds and have been employed for the determination of various organic micropollutants. They provide unique selectivity on the basis of molecular recognition, which is particularly suited to complex matrices [78]. The application of immunoassays to the determination of BPA is rather recent. Several enzyme-linked immunosorbent assays (ELISAs) have been developed for the determination of BPA in various media [18, 35, 53]. ELISAs are simple, rapid, and cost-efficient assays. However, special attention has to be given when ELISAs are applied in complex matrices with low concentrations of BPA, due to the cross-reactivity phenomena and matrix effects that may reduce the precision of the method [72, 78].

2.6 Quality Assurance and Quality Control

Analytical methods for the determination of BPA have to lead in reliable measurements of trace levels in complex matrices. At these levels, many factors may affect the reliability of the results. Therefore, the analytical procedure should be subjected to detailed evaluation regarding efficiency.

Samples shall be obtained, handled, and processed in such a way that avoids possible contamination or loss of BPA. The analytical accuracy of the method is normally measured directly by analysis of certified reference materials (CRM).

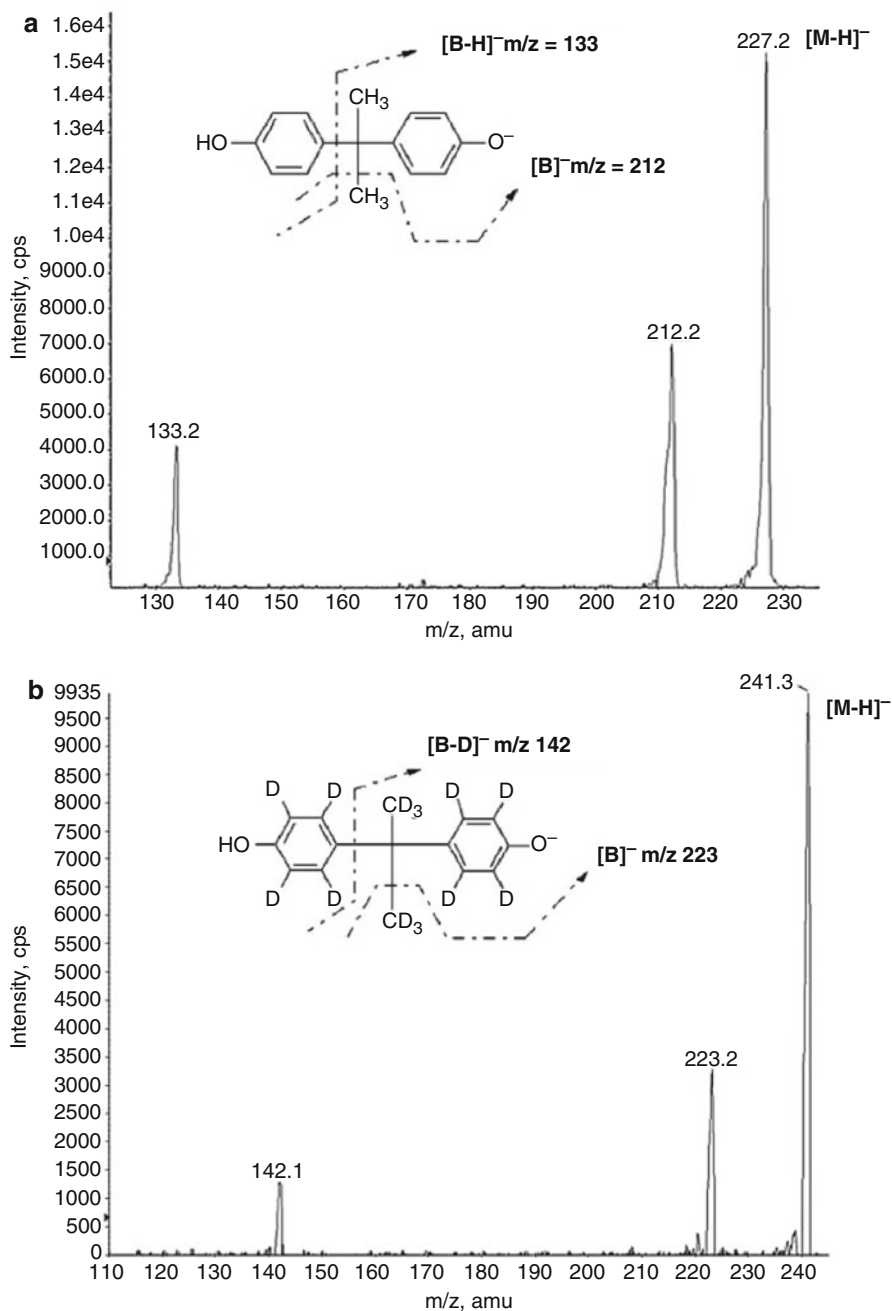


Fig. 2.2 Product ion mass spectra of (a) BPA (precursor ion m/z 227) and (b) BPA-d16 (IS) (precursor ion m/z 241) by employing HPLC/ESI-MS/MS in negative ion mode [75]

In the case of BPA, there is no certified reference material currently available. Instead the recovery can be determined using fortified matrix samples containing known amounts of BPA at least two or more concentration levels. Thus, the recovery efficiency of BPA can be established and appropriate correction can be performed. Fortified samples are also useful to determine whether the sample matrix contributes bias to the results.

One of the most important points in analysis of BPA is the use of appropriate internal standard (IS) to compensate for possible loss of the analyte during sample processing and variations in instrumental performance. The addition of IS at the beginning of the extraction procedure is recommended. Depending on the availability, either stable isotope-labeled forms of BPA, which is particularly suited for mass-spectrometric detection, or compounds that are structurally related to the analyte are used. The isotope-labeled internal standards mostly used in the analysis of BPA by employing mass spectrometry are BPA-*d*16 and $^{13}\text{C}_{12}$ -BPA (Fig. 2.2).

The identification of BPA in chromatographic techniques is based on relative retention time of the eluted peak. The retention time of BPA in the samples has to match that of calibration standard within of specific retention time window, i.e., the retention time of the analyte to that of the internal standard shall correspond to that of calibration solution at a tolerance of ± 0.5 for GC and of ± 2.5 for LC [79]. When mass spectrometric detection is employed (GC-MS and LC-MS methods), additional criteria for identification of BPA, besides retention time, are the characteristic diagnostic ions (molecular ion, characteristic adducts of the molecular ion, characteristic fragment ions, and all their isotope ions) and their relative abundance. The relative intensities of the diagnostic ions of BPA, expressed as a percentage intensity of the most intense ion or transition, shall correspond to those of the calibration standard, at an acceptable tolerance (i.e., $\pm 15\%$ in GC-MS and $\pm 25\%$ in LC-MS) [9, 79].

Cross-checks involving reagent, procedural and field blanks, calibration standards, quality control samples, standard additions on samples, should be carried out through the entire procedure simultaneously with samples in order to ensure the quality of the analytical results.

2.7 Sources and Occurrence of Bisphenol A

2.7.1 Dental Restorative Materials

Bisphenol A is a common ingredient in the resin-based restorative materials used in dentistry. BPA is a precursor to the resin monomer of bisphenol A diglycidyl ether methacrylate (Bis-GMA) and bisphenol A dimethylacrylate (Bis-DMA) that are the main constituents of most commercially available composites and sealants used in dentistry. The resin matrix is initially a fluid containing a monomer that is cured or converted into a rigid polymer by chemical or photo-initiated polymerization reaction.

Various studies reported that BPA is leached from dental materials (Table 2.2). The leaching of BPA could be attributed to (a) unreacted BPA impurities in resins due to incomplete polymerization process, (b) chemical and/or mechanical degradation of these materials, (c) hydrolysis of carbonate linkages of BPA-based epoxy resins at high temperatures, and (d) enzymatic degradation through esterase enzymes present in saliva which can hydrolyze the susceptible ester bond in Bis-DMA monomers.

The leachable BPA concentrations greatly depend on the type of dental material, the polymerization conditions, elution media (i.e., water, methanol, ethanol), and exposure conditions (i.e., pH values, time of elution) [10, 14, 15]. The occurrence of BPA in saliva of patients after treatment with BPA-based dental sealants or composites has been often reported (Table 2.2). The higher concentrations of BPA immediately after placement of dental materials, being decreased within the next hours [17, 19]. However, the magnitude of these exposures, the long-term potential for sealant leaching, and the potential for adverse effects are still controversial.

2.7.2 *Environmental Samples*

In the analysis of environmental samples, BPA is usually determined along with other xenoestrogens such as nonylphenol and octylphenol, thus the methods employed aiming at the simultaneous extraction and determination of all target compounds. Due to widespread application of BPA, it is commonly found in sewage effluents, industrial wastewaters, and surface waters (Table 2.3). BPA occurred at high concentrations in raw wastewaters. Treatment of wastewaters through the conventional or advanced methods results in elimination of BPA in treated effluents [29, 80]. Surface waters usually exhibited relatively low concentrations. However, higher concentrations of BPA have been determined in surface waters directly impacted from specific sources and/or occasional discharges. BPA is subjected for possible identification as priority hazardous substance in water, due to possible adverse effects to aquatic environment [81]. For the protection of aquatic life in freshwaters, a predicted no-effect concentration of 1.6 $\mu\text{g/L}$ is proposed, whereas this value in marine waters is 0.15 $\mu\text{g/L}$ [82].

2.7.3 *Food Samples*

Bisphenol A can be present in foods as a result of migration from epoxy resin coatings used to lacquer-coat the interior of food cans, wine storage vats, water containers, and water pipes. The other main source is polycarbonate plastics used in a wide range of applications such as water carboys, reusable milk containers, food storage vessels, and baby bottles. Incomplete polymerization of these materials during manufacture and increased temperatures imposed during heating result in migration of BPA from these materials into food [3, 72].

The concentration ranges of BPA in various products (water, beverages, wine, milk, and food simulants) are shown in Table 2.4. The reported concentrations of BPA are relatively low, although variations have been observed. The occurrence of BPA in food products depends on the coating/packing materials, type of food, handling, and storage conditions [7, 37, 40].

Food is considered as the major pathway of human exposure to BPA. The tendency of this compound to migrate from food contact materials has been acknowledged in European Union food law. A specific migration limit of 3 mg BPA/kg food has been initially set. However, in 2004, a lower limit value of 0.6 mg/kg food has been proposed in the amending document relating to plastic materials and articles intended to come into contact with foodstuff [83, 84]. The European Food Safety Authority established a tolerable daily intake (TDI) of 0.050 mg/kg bw, which represents a safe level for daily exposure over a lifetime [85]. Similarly, the Integrated Risk Information System (IRIS) of US EPA proposed a reference dose of 0.050 mg/kg bw/day for chronic oral exposure (RfD) [86].

2.7.4 *Biological Samples*

The widespread human exposure to BPA has been highlighted by measurements in human fluids and tissues. The presence of BPA or its metabolites in urine, blood, or various tissues is an indication of human daily or cumulative exposure to this compound. However, an estimation of daily exposure to BPA based on the concentrations found in biological samples requires a detailed knowledge of its biotransformation pathways and toxicokinetics. Based on the known human toxicokinetics of BPA, measurement of urinary concentrations of bisphenol A-glucuronide is the most appropriate and feasible way to assess daily exposure to BPA in humans.

The occurrence of BPA in urine samples along with analytical methodology used is shown in Table 2.5. The biomonitoring data demonstrate that the average concentrations of BPA in urine samples from the general population are relatively low and confirm that BPA is mainly present as glucuronide in human urine [72].

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Chapter 3

In Vitro Assay Systems for the Assessment of Oestrogenicity

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17 β -Estradiol (or simply estradiol) is the predominant sex hormone present in the female mammals, and its impact is vital not only on reproductive and sexual functions but also for many other tissues, most notably the bones. Estradiol acts on target cells through its interaction with two types of specific receptors (oestrogen receptors, ERs) called ER- α and ER- β , which reside in the cytoplasm but upon binding of oestrogen migrate in the nucleus to regulate the transcription of target genes [1, 2]. Recently, membrane-bound receptors for estradiol have also been identified [3].

During the last 50 years, substantial evidence has been accumulated on many exogenous compounds that behave similarly to the endogenous oestrogens, hence termed phytoestrogens – when they are of plant origin – or xenoestrogens, a term mainly referring to chemicals produced industrially [4]. Xenoestrogens belong to a wider group of compounds called “endocrine disruptors” due to the fact that upon their intake by humans or animals, they interfere with the normal hormonal balance of the organism, causing among others reduction in sperm counts and fertility, developmental and/or congenital birth defects, increased incidence of testicular and/or breast cancer in humans as well as gross birth deformities, behavioural abnormalities and both feminisation and masculinisation in animals [4–7].

Xenoestrogens usually are constituents, chemical intermediates or derivatives of industrial products with a huge variety of uses, such as agrochemicals and pesticides, food additives and supplements and medical and pharmaceutical products, to name only a few. Especially in dental practice, there are many products such as restorative materials, liners, adhesives, oral prosthetic devices, tissue substitutes and rebase materials, which possess – or there are reasonable suspicions that they possess – oestrogenic activity [8, 9]. Most important among them is bisphenol-A (BPA),

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a molecule with established oestrogenicity [10, 11] and endocrine disruptive properties [12]. Beyond BPA, there are other bisphenols, e.g. bisphenol-A dimethacrylate (Bis-DMA), bisphenol-A glycidyl dimethacrylate (Bis-GMA) or BPA diglycidyl-ether (BADGE), and phthalates, e.g. n-butyl benzyl phthalate (BBP) or dibutyl phthalate (DBP), in various dental materials that raise suspicions for endocrine disruptive behaviour [9, 13].

As a consequence of all these undesirable effects of xenoestrogens, a whole battery of assays has been developed for the evaluation of the oestrogenic properties of natural or synthetic compounds. This chapter will focus on the presentation of the *in vitro* assays used in the literature for the assessment of the oestrogenicity of various compounds with an emphasis on those used in dental practice.

Generally, these assays can be categorised according to the use or not of various cellular types, as follows [14, 15]:

1. Cell-free assay systems
2. Yeast assay systems
3. Mammalian cell assay systems

3.1 Cell-Free Assay Systems

Cell-free assay systems are based on a chemical reaction and can be performed in a test tube. Typically, they assess the affinity of a test compound for oestrogen receptors. As stated above, the formation of a hormone-receptor complex is required for the manifestation of the hormone effects. Many xenoestrogens bind also to oestrogen receptors; hence, the assessment of their receptor-binding affinity (RBA; also stands for relative binding affinity) may provide a first indication of their ability to mimic the endogenous hormones or to interfere with their activities, as well as an indication of their potency.

Originally, the RBA of a xenoestrogen was calculated based on its capacity to compete with radiolabelled estradiol molecules for binding to the ER, and for this reason, sources rich in ER were used, such as cytosolic extracts from breast tumours [16] or murine uterus extracts [17]. More recently, after the cloning of ER- β , recombinant ER- α and ER- β molecules replaced the crude extracts, and high-affinity fluorescent ligands are being used instead of radioactive hormones for competition experiments [18, 19], thus improving the reproducibility of the assay and allowing for screening compounds that may bind only weakly to ERs and have limited aqueous solubility.

A further variation of this type of assays is the use of fluorescein-labelled synthetic oligonucleotide oestrogen response elements (EREs) of various target genes for the assessment of the xenoestrogen-dependent binding of ER to the ERE [20]. Finally, a fusion protein of the ER with glutathione S-transferase (GST) can be used to study the interaction of the xenoestrogen with ER and a radiolabelled coactivator (such as steroid receptor coactivator-1a, SRC-1a or transcriptional intermediary factor-2, TIF2) by autoradiography [21, 22]. In an analogous approach, fluorescently labelled coactivators and ER can be used to assess the

interaction with the xenoestrogen by fluorescence resonance energy transfer (FRET) [23].

The above-mentioned cell-free assays have the advantage that they are easy to perform, thus allowing for a high-throughput screening of test compounds, except from those that include coactivators. However, they only provide information on the chemical affinity of a compound to the ER, without any clue on the biological phenomena triggered by their interaction, and especially without distinguishing between agonistic and antagonistic activity [14]. Hence, typically the RBA assay is used in combination with other assays that are more informative, such as the yeast two-hybrid or the E-screen (see below) [9, 24, 25].

3.2 Yeast Assay Systems

More biologically relevant are the yeast assay systems, which have been made possible through transfection techniques. These systems are based on the artificial expression in the yeast (*Saccharomyces cerevisiae*) of ERs of human or animal origin or other parts of the molecular machinery conveying the oestrogenic signals. Since it is an artificial system, there is usually a convenient reporter gene included in it, such as β -galactosidase or chloramphenicol acetyltransferase (CAT) or luciferase. For example, an approach similar to the cell-free GST pull-down systems described above is the so-called yeast two-hybrid assay. In this case, the ligand binding domain of the ER and a fusion of galactosidase activation domain with the receptor interaction domain of a coactivator (in most cases TIF2, see above) are subcloned in yeast expression plasmids, in order to transform the appropriate yeast strains, so that the interaction of ER with its ligands can be detected by a chromatic reaction, which can be quantitated measuring absorbance or chemiluminescence [24–27]. In a more focused approach, yeast strains are co-transfected with an ER cDNA and an artificial reporter gene containing the ERE of a known oestrogen target gene linked usually to β -galactosidase [28].

Yeast assay systems are popular to many researchers because yeast is a well-characterised model organism, it is widely accessible, it is readily transformed and it has a broad range of suitable plasmids and promoters available, while the experiments can be performed easily and rapidly. Furthermore, the transfected yeast model is capable of high levels of sensitivity [28], and it is “pure” in the sense that neither ER mechanisms nor other molecules known to interact with them exist in the untransformed organism. However, exactly this lack of the mammalian cell context makes the system highly artificial; hence, the responses observed may not reflect the physiological response in human. It has been reported, for example, that such a system was highly specific for estradiol compared to other molecules with known oestrogenic activity, such as diethylstilbestrol (DES) or mestranol [29]. Moreover – in comparison with a mammalian cell assay system – in the yeast, one could not detect various putative anti-oestrogenic molecules, most probably because some molecules active on mammalian cells cannot cross into the yeast cell through its specialised cell walls [28, 30].

3.3 Mammalian Cell Assay Systems

The vast majority of the studies testing *in vitro* the oestrogenicity of xenoestrogens or phytoestrogens are using at least one mammalian cell assay system alone or in combination with some of the methods described above. The mammalian cell assay systems can be categorised (a) according to the tissue the cells are originating from (usually breast or endometrium, although pituitary cells have also been used), (b) according to the use or not of genetically engineered cell strains and (c) according to the end point assessed by the method, which can correspond to a very broad range of cellular activities: expression of certain genes or proteins, steroidogenesis and activity of marker enzymes as well as DNA synthesis and cell proliferation [14, 15, 31, 32].

Among the various alternatives of the mammalian cell assay systems, the most popular is by far the so-called E-screen [33], i.e. the assessment of the proliferation of the human breast adenocarcinoma cell line MCF-7. This is true also among the researchers studying potential xenoestrogens especially among the materials used in dental practice [10, 11, 25, 34–41]. MCF-7 cell proliferation during the E-screen can be assessed by direct cell counting, usually by a Coulter counter [42] – by estimating DNA synthesis rate through the incorporation of tritiated thymidine [36] or 5-bromo-2'-deoxyuridine [14] into DNA, by measuring total DNA content fluorometrically after binding of an appropriate dye [43] but most often by chromatometric methods such as the MTT [3-(4,5-dimethylthiazol)-2,5-diphenyltetrazolium bromide] assay [33, 34], or the sulforhodamine-B (SRB) assay [25, 37–39, 41], or the utilisation of the neutral red vital stain [11, 44].

The E-screen is appropriate for quantitative analysis of both oestrogenic and anti-oestrogenic activities [15]; it is very sensitive [45] and easy to perform, allowing for high-throughput experimental approaches. Moreover, the fact that it measures a physiological end point of oestrogen action of high biological complexity, i.e. the proliferation of ER-bearing breast cancer cells, affords the opportunity to identify factors that may impact on mixture effect predictability [32].

A scepticism regarding the specificity of E-screen has been expressed, since progesterone and certain 19-nortestosterone derivatives, as well as caffeine, ethanol and various growth factors, have been reported to induce MCF-7 cell proliferation [31, 46]. In contrast, the team of Soto has shown that when the assay is performed in charcoal-dextran stripped serum or plasma, in order to remove endogenous oestrogen – a plasma-borne specific inhibitory activity of ER-bearing breast cancer cell proliferation (termed estrocolony-I and sharing properties with human serum albumin) remains, and only oestrogens can reverse this inhibition [47, 48]; hence, the assay is absolutely oestrogen specific. Nevertheless, a simple way of identifying molecules stimulating or inhibiting MCF-7 cell proliferation non-specifically is to test them in parallel on an oestrogen-insensitive breast cancer cell line, such as MDA-MB-231 (Fig. 3.1) [34, 36].

Scepticism regarding the E-screen has also been expressed because different clones of MCF-7 cultured in identical conditions showed distinct differences in the proliferative response to estradiol and to the xenoestrogens, *p*-nonyl-phenol and

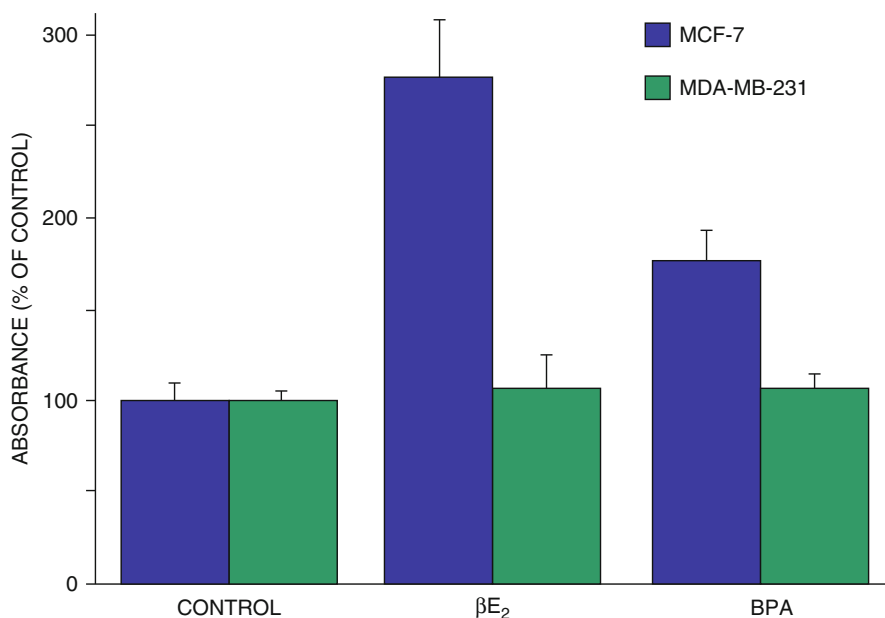


Fig. 3.1 E-screen assay. MCF-7 and MDA-MB-231 cells were grown for 6 days in the absence (control) or presence of 10^{-9} M 17β -estradiol (βE_2) or 10^{-8} M bisphenol-A (BPA), and their viability was assessed using the MTT assay

bisphenol-A [49], as well as to commercial resin-based dental restorative materials [41]. However, this is a common problem when working with cancer cells, and one can overcome it through meticulously uniform cell stocks. Furthermore, apart from MCF-7 cells, other oestrogen-responsive breast cancer cell lines have been used in the E-screen assay, such as T-47D [43] or ZR-75-1 [50].

The proliferation of the Ishikawa human endometrial cancer cell line has also been proposed to be used for the evaluation of oestrogenic activity [51], but there were indications that the response of this cell line is not specific for oestrogenic molecules [52] in contrast to breast cancer ones. However, Ishikawa cells have been shown to respond to oestrogen and phytoestrogens with a potent induction of alkaline phosphatase (ALP) activity, which is oestrogen specific [53, 54]; hence, it can be used for the screening of potentially oestrogenic compounds [19, 55].

Probably, the most important drawback to the use of both the E-screen and the ALP-induction assays as rapid screening tools is that they are time consuming (the assessment can take from 3 to 6 days, depending on the protocol variation). Accordingly, analysis of oestrogen-regulated gene or protein expression in various cell types can be used as an alternative. For example, expression in MCF-7 cells of the genes coding for the progesterone receptor [56] and for the trefoil peptide pS2 [57, 58] or prolactin production by rat pituitary cells [59, 60] has been proposed as tools to study an oestrogen-specific response. However, these assays are not always

as sensitive as the E-screen [32], and they require the use of laborious and/or expensive techniques such as northern blotting or real-time PCR; hence, they are not appropriate for high-throughput screening.

The use of genetically engineered mammalian cell systems was intended to solve some of the above problems. In most of the cases, cells are transfected with an oestrogen-inducible reporter gene, or they are co-transfected with an ER-construct and an ERE-containing reporter gene, similarly to the approaches described above in yeast. The reporter genes usually are designed for measuring CAT or luciferase activity, which due to their high sensitivity offer the possibility to identify even weak oestrogens [15]. The transfection can be transient [61] or stable [19], the latter being more advantageous in terms of reproducibility, as well as rapidity, once the stable line is ready for use [32]. The parental cells used for transfection can be either ER negative, such as HeLa [61] or HEK-293 [19], or ER responsive, like MCF-7 or MG-63 [62]. Consequently, it is clear that apart from the high-throughput capability and the rapidity of these assays (typically gene expression can be assessed within 24 h), their main advantage is their versatility, allowing for separate tests for the various ER subtypes and EREs, recognising both oestrogens and anti-oestrogens and giving the choice of selecting an ER-naive cell context, such as in the case of HeLa cells, or a more physiological context, such as that of MCF-7 cells [15]. Still, these assay systems are artificial, and there are reports regarding the irreversible silencing of the reporter gene after treatment with anti-oestrogens, such as 4-hydroxy-tamoxifen [63, 64].

3.4 Conclusion

In this chapter, a battery of *in vitro* assays for the evaluation of the oestrogenic properties of natural or synthetic compounds was presented. One should not forget that the evidence for *in vitro* oestrogenicity of a test molecule cannot always be conclusive without the knowledge of *in vivo* data regarding its metabolism and bioavailability. However, only *in vitro* testing can respond to the urgent need for screening the huge amount of novel materials produced every day in the industrialised societies. Specifically in dental practice, most often, the E-screen, the RBA and various yeast assay systems are being used, a fact probably reflecting their credibility and/or their simplicity.

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Chapter 4

BPA Effects In Vivo: Evidence from Animal Studies

Efthymia Kitraki

4.1 Introduction

Bisphenol A (2,2-bis-4-hydroxyphenyl-propane, BPA), is a well-known endocrine disruptor that is used as a monomer in the manufacture of dental sealants, epoxy resins and polycarbonate plastics that have extensive use in dentistry or medicine, in food packaging industry and in plastics' production. BPA is contained in many everyday life items, such as house plasticware and baby bottles, from where it is released, for example, by heating, resulting in food or drink contamination. Leached components from dental composites and sealants in the oral cavity are also considered a possible source of human exposure. BPA exposure can also occur by inhalation of contaminated air, for example, from decomposed monomers during medical or dental practice [1].

Animal studies confer a valuable tool for the assessment of BPA effects in vivo. The easiness of experimentation with laboratory animals has allowed a variety of in vivo approaches, summarised in comprehensive recent reviews [2, 3]. Aquatic organisms such as fishes or amphibians have been widely used to assess the effects of BPA in the ecosystem. However, the effects of BPA on animals' physiology have been mainly explored in laboratory animals that are closer to humans. Advantages from the use of small rodents include their genomic similarity to humans, suitability for genetic studies and offspring follow-up, as well as a less stringent legislation, compared to that of non-human primates. Disadvantages include their differences in metabolism from humans that may interfere with BPA degradation kinetics. Intrauterine growth also differs significantly between rodents and humans, and position into the bicornate rodent uterus may differentiate the impact of BPA in each embryo [3]. Furthermore, in contrast to humans, developmental maturation in rodents takes place mainly after birth.

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Although translation from animals to humans should be cautious, given the aforementioned differences in their physiology, growing evidence from animal studies suggests that environmental exposure to BPA may adversely impact human health as well. Initial toxicology studies have exposed animals to rather high doses of BPA and have reported numerous dysfunctions in animals' reproductive physiology. The dose of 50 mg/kg bw/day was set as the LOAEL (lowest-observable-adverse-effect level) dose, based on observations from the reproductive system and tumour growth. To further simulate the low daily exposure of humans, a safe reference dose was determined at 50 µg/kg bw/day. During the last decade, however, many studies have shown that exposures below the safe dose can still affect animals' physiology and behaviour.

Nowadays, there is an ongoing vivid debate on the potential risks for human populations from exposure to low BPA doses. On one side, the majority of independent basic research laboratories emphasise the existence of adverse BPA effects on animals' and subsequently humans' health within the 'safe' exposure [4]. On the other side, governmental agents (US National Toxicology Program and Food and Drug Administration) based on few risk assessment studies [5, 6] assure that there is no risk for human health at current exposures. They only express some concern for possible effects in neural and prostate physiology upon perinatal exposures [7]. The main argument from the side of the scientific community is that the risk assessment studies were not designed to detect delicate developmental effects but were rather focusing on gross BPA-induced changes including mortality, fertility and tumorigenesis. Additional arguments relay on the different mechanisms of BPA actions that do not allow linear extrapolations from high doses to very low ones [8, 9]. Properly designed and reproducible studies have so far provided sound evidence for adverse effects of ecologically relevant BPA exposures during development.

The aim of this chapter is to summarise evidence from rodent studies on the effects of BPA upon exposures that are relevant to humans, that is, exposures around or below the 'safe' daily uptake, estimated at 50 µg/kg bw. The chapter is divided in two parts: In part I, important issues on the design of an animal study will be addressed. In part II, evidence from low-exposure rodent studies will be presented, with an emphasis in the nervous system that appears highly susceptible to low BPA actions.

4.2 Part I: Issues on Experimental Design

Research on the effects of BPA in mammals has produced a wealth of data showing diverse actions of this xenoestrogen in several systems. These effects of BPA often vary significantly, even within the same system/organism, and make it difficult to draw a definite conclusion. The main reason for these discrepancies is the variation of experimental protocols that does not allow direct comparisons from study to study. The aim in the following paragraphs of Part I is to shed light on parameters of the experimental design that may confer diversity in the obtained results.

4.2.1 *Route of Exposure*

Humans are exposed to BPA via both oral and nonoral routes. These include consumption of BPA-containing foods or drinks, leakage from medical/dental devices as well as inhalation of BPA-contaminated air. Animal studies mimicking the above routes are thus all appropriate in evaluating human effects, given that the dosage used is kept within relevant human exposure. In most rodent studies, oral administration is preferred, because it is considered to represent the most common way of human exposure. Oral administration in rodents is met in several variations: provided into the drinking water, by gastric gavages, dissolved in oil or combined with food. Other approaches bypass the digestive track by applying subcutaneous, intravenous, intracisternal, intramuscular or pumping methods. None of these paradigms is ideal however, as they may occasionally preclude inaccurate dosing, variations in exposure over time, psychological stress or vehicle contaminations [3].

In order to mimic the effect of BPA-containing leached substances from dental sealants and resins, Al-Hiyasat et al. [10, 11] have eluted bisphenolic compounds from dental composites and provided the solution in mice by gastric gavages. In such paradigms, using a mixture of compounds, it is important to precisely analyse the composition and concentration of the active components in the starting material. Even so, it is still difficult to attribute a certain effect to a particular component.

The route of exposure may also differently affect the pharmacokinetics and active levels of BPA. Oral administration results in earlier metabolic inactivation of BPA, compared to SC or IV routes, due to the direct passage from gut and liver before entering circulation. This could possibly differentiate its biological effect, although no definite conclusion has yet been reached, since some but not all studies support this possibility [3, 12].

4.2.2 *Pharmacokinetics*

BPA is rapidly metabolised in glucuronide and sulphate compounds that show low estrogenic activity and cannot bind to estrogen receptors. The liver is the major site of conjugation of free BPA to inert metabolites. Intravenously injected BPA in rodents can quickly reach all organs (it peaks at 20–30 min) and is also rapidly transferred across the placenta to the fetuses. Efficient conjugation is witnessed with a decline of active BPA concentration after 2 h [13]. Upon oral administration in rodents, it is estimated that approximately 95 % of BPA is soon inactivated through metabolism in the liver or intestine before reaching the general circulation. It is thus possible that rodents receiving orally BPA are exposed for a shorter time in the active compound, compared to the injected animals [14]. Sex differences may also influence the pharmacokinetics and availability of free BPA in both rodents and humans. Higher active BPA concentrations are detected in males that can be

explained by the lower expression of the main BPA glucuronidating enzyme [UDP-glucuronosyltransferase 2B1 (UGT2B1)] in their liver [15, 16].

In humans, pharmacokinetic studies performed in adult volunteers showed that the ingested BPA is metabolised to inactive compounds more rapidly compared to rodents. The kinetic profile of inactive metabolite d-BPA glucuronide showed a rapid peak and urinary elimination with a half-life of approximately 5 h [17].

Based on the rapid metabolic clearance of BPA that is more effective in humans compared to rodents, the European Food Safety Authority (EFSA) concluded in 2008 that rodent toxicity data are not directly relevant for human risk assessment and that perinatal exposure of humans has a negligible risk [18]. However, there are several arguments against this conclusion, summarised as follows:

- (a) Bio-monitoring studies have detected free BPA in the rat or human placenta and in fetuses, implying that human exposure to BPA is frequent and not negligible.
- (b) The metabolic detoxifying mechanisms are not similarly effective in all tissues, for example, are less potent in the brain.
- (c) There are counteracting mechanisms of de-conjugation that re-provide free BPA. Indeed, extensive de-conjugation of BPA glucuronide in utero and BPA sulphate in neonates has been reported [19].
- (d) The counteracting mechanisms appear particularly effective during the perinatal period. The enzyme activity required for the de-conjugation is higher in the placenta of rodents, and the concentration of BPA in this tissue is higher than that in the maternal or fetal circulation [15], indicating a higher exposure of fetuses to active BPA [20].
- (e) There are evidences for BPA actions (non-genomic) that require very low concentrations of the xenoestrogen and do not depend on receptor binding [21].

4.2.3 Dosage

The dosage of BPA used in animal studies varies from high pharmacological to very low ones that are below the safe limit. Toxicology studies have determined the maximum tolerable dose for BPA at 1,000 mg/kg bw/day. A dose of 50 mg/kg bw/day was set as the LOAEL (lowest-observable-adverse-effect level) dose, concerning effects in the reproductive system and tumorigenesis. The European Food Safety Authority has set the tolerable daily intake (TDI) of BPA for the European Union at 0.05 mg/kg/day [18]. This dose, also termed 'safe dose', is however higher (more than 20 times) than doses reported to cause adverse effects in rodents [3]. This discrepancy could be explained as follows: Initial toxicology studies have used threshold-based or linear non-threshold models to estimate the biological effect of different BPA doses that assume effects over a threshold and increasing number of effects by increasing dose, respectively. However, most hormones appear to follow non-linear biphasic dose responses. According to such biphasic models, the highest

effects can be seen in very low and very high concentrations of the hormone (U-shaped) or in the intermediate doses (inverted U-shaped). The biological effects of BPA in cultured cells appear to follow this biphasic model. In *in vivo* studies, it is more difficult to have a complete confirmation of the model, because data on end point effects at different dosages are missing. Nevertheless, there is some evidence showing a non-linear mechanism of BPA actions [22, 23] that should be taken into consideration when comparing effects of different BPA doses on the same biological system.

4.2.4 Timing and Duration of Exposure

Gonadal steroids exhibit both organisational and activational actions. Organisational actions, taking place mainly during fetal life, refer to the ability of these hormones to program functions of the adult organism. Activational actions exerted in the pubertal and adult organism are driven by the gonadal hormones and regulate relevant physiology and behaviour. Apparently, BPA exposure during development may critically interfere with fetal and neonatal programming. The developing organism is more sensitive to BPA for the additional reason that it lacks fully functional detoxifying and immune systems.

Exposures of adult animals have been used to address the effects of BPA on the mature reproductive system and to study the interactions of this xenoestrogen with the endogenous gonadal steroids. Developmental exposures apply BPA during the whole gestation and/or lactation or during critical time windows within this period that vary depending on the timing of each system's development. BPA is provided to the mother and reaches offspring through the placenta and/or milk. Most developmental studies use long-term exposures to mimic situations in humans. Given the rapid metabolism of BPA, daily exposures for a long time are preferred from acute treatments. In studies with adult exposures, however, BPA is usually administered for shorter periods of time.

4.2.5 Choice of Rodent Species

The choice of rodent species is of importance for the reproducibility of the results obtained, since there are many differences between rats and mice, as well as among strains. In general, mice are considered more sensitive than rats to BPA actions. However, this must be further delineated in light of the specific question to be addressed. For example, mice are preferable for genetic studies, while rats for behavioural testing. Species' differences in the sensitivity of certain tissues also exist. For example, the mammary gland of rats is more susceptible to BPA than that of mice [3, 24].

Attention should be also paid on the strain of rat or mouse, as not all strains show the same sensitivity to BPA [2]. Most reports on strain differences have so far examined effects in the female reproductive system. Fischer 344 rats are considered more sensitive than Sprague–Dawley female rats in the effects of high BPA dose (37.5 mg/kg bw) on vaginal epithelium proliferation [25]. In another study [26], the estrogenic potency of BPA was evaluated in three different rat strains (Sprague–Dawley, Wistar and DA/Han) by determining the uterine weights of adult females exposed for 3 days to high levels of BPA (200 mg/kg bw/day). In contrast to the previous studies, only small differences were observed among the strains, and their blood concentration of BPA did not differ 24 h after the last dose. At this point, it should be noted that the dose used was quite high and that the uterotrophic assay applied has been questioned as to its sensitivity at human relevant exposures. Others have reported reduced sensitivity of the CR–Sprague–Dawley rat strain in the estrogenic actions of BPA, based on the effects of a positive control (estradiol or another potent estrogen) included in the study [2]. However, the ideal positive control to compare BPA actions is still a matter of debate, as xenoestrogens vary significantly in their properties and potential specificity.

Future studies addressing strain sensitivity should take into consideration that animals may differ depending on the biological end point and the dose of BPA used. Furthermore, food content and housing conditions can greatly influence the biological outcome even within a certain laboratory, so it is advised to include all strains to be compared in the same experimental protocol.

4.2.6 Choice of Sex to Study

During the last decade, there is an increasing trend in science towards studying both sexes in basic and clinical research, based on the accumulating gender differences in physiology and disease. The need to study both males and females is more obvious when assessing the biological effects of a xenoestrogen, given the known sex differences in the organism's response to estrogens. Additionally, BPA is acting as a selective estrogen receptor modulator (SERM), and the response of the two sexes cannot always be predicted based on the action of a typical estrogen. Due to the experimental design of most animal studies using developmental exposures, both male and female offspring are available for observation, and there is so far a wealth of evidence concerning sexually dimorphic BPA effects. Special attention should be paid when studying animals in adulthood as to the activational actions of gonadal hormones. Estrous cycle must be monitored and normalised in female subjects, since endogenous estrogens may influence several physiological responses. Similarly, testosterone levels must be measured, and adult males should be individually housed to avoid interference of the testosterone-mediated dominance status in the results.

4.2.7 *Appropriate Controls*

To assure the effects of BPA on a certain system, one is encouraged to include in the study the appropriate positive controls. These are hormonal compounds, whose properties and biological effects have been well established in the system under study. The use of a positive control is important especially in the cases that there is no observable BPA effect. The positivity of the control compound will then confirm the negative results and the hormonal sensitivity of the rodent species used. 17- β -Estradiol, diethylstilbestrol (DES) and ethinylestradiol have been used as positive controls to verify estrogenic effects of BPA in rodents' reproductive system. DES is a synthetic estrogen often used as a positive control for xenoestrogens. It has a higher activity for ER α and equal affinity for ER β , compared to estradiol. The selection of positive control must take into consideration the route of BPA administration. For oral exposures, DES and ethinylestradiol are preferable because they retain their activity better than estradiol.

Given that BPA may not only act as an estrogen-mimetic compound, especially outside the reproductive system [2, 27, 28], the a priori selection of a positive estrogenic control may not confer to the complete elucidation of the results. In these cases, incorporation of more than one control substances of different properties (i.e. estrogenic, anti-androgenic or androgenic and antithyroid) could provide a better solution. Furthermore, caution should be paid on the appropriate dose for each control compound used: So far, the doses are adjusted based on hormones' affinities for the classical estrogen receptors. Evidence for non-genomic actions, exerted at much lower concentrations via membrane-bound entities, requires updating of the used rules.

4.3 Part II: Evidence from Low-Exposure Studies

4.3.1 *General*

Initial toxicological studies for the effects of BPA in vivo have used rather high, pharmacological doses of the agent, close to or higher than 50 mg/kg bw/day, set as the LOAEL dose [29]. These risk assessment studies were focusing on gross BPA-induced changes including fertility, mortality and neoplasia, but were not designed to detect more delicate developmental effects that however may crucially impact individuals' health. Furthermore, in vivo studies using high levels of BPA may be inappropriate to judge for the harmless of lower doses, since BPA actions often follow a non-linear pattern. In compliance with this, low-dose BPA effects in the reproductive system were not witnessed after exposure to high doses [22, 23].

Given the increasing requirement for animal models that simulate human exposure, most studies conducted during the last years have used human relevant doses

of BPA (around or below the safe reference dose of 50 $\mu\text{g}/\text{kg}$ bw/day). In this chapter, only the effects of low BPA exposures will be presented. Information on the impact of high doses can be found in several recent reviews [2, 30]. Most low-dose studies have used perinatal exposures of the animals (during gestation and/or lactation) that preclude the possibility of programming and render the animals more sensitive to BPA actions. In these studies, the biological end points were evaluated either in young offspring or more often in adulthood, to check for possible sustained effects. On the other hand, the number of studies applying adult exposures is limited and focuses on the activational effects of BPA in interaction with the fully developed hormonal system of the animal.

BPA exposure in utero can influence the development of the whole embryo. Imanishi et al. [31] used DNA microarray analysis to define the genes whose expression was altered in the murine placenta at the 18th day of gestation. The daily BPA dose used was only 2 $\mu\text{g}/\text{kg}$ bw and was administered in pregnant mice from day 6 to 17 of gestation. Significant alterations, depending on the sex of the embryo, were detected in mRNA levels of several nuclear receptors upon BPA treatment. These included progesterone receptor and estrogen receptor β genes that were upregulated in male, but not female, BPA-treated embryos. Given the critical contribution of ovarian steroids in the maintenance of pregnancy and fetal differentiation, the BPA-induced changes in placenta sensitivity to these hormones may play a role in the normal embryonic development.

Based on the estrogen-mimicking properties of BPA, the tissues initially selected for studies were the well-known targets of estrogens. These included female and male reproductive organs (vagina, uterus, ovaries, testis) and accessories (mammary gland, prostate), as well as central nervous system (CNS) centres (hypothalamus and pituitary) regulating reproductive physiology and behaviour. The end points assessed were related to sexual maturation, fertility and sexually dimorphic behaviours. Recent studies have also investigated possible effects of BPA on thyroid function and metabolism. Search for possible effects in the CNS outside the hypothalamus has recently unravelled several nonreproductive BPA actions upon particularly low exposures. Given the numerous targets of estrogens in the body and the multiple mechanisms of BPA actions, it will not be surprising to detect novel BPA-endangered systems in the near future.

4.3.2 Effects in the Reproductive System

In contrast to the well-described adverse effects of pharmacological doses in the reproductive system of rodents [for review, see 2], BPA exposures to less than 100 $\mu\text{g}/\text{kg}$ bw/day have only minor effects.

Oral exposure to 2, 20 or even 200 μg BPA/kg/day from gestational day 7 to postnatal day 18 does not significantly change anogenital distances, vaginal opening, fertility or CNS defeminisation in female rat offspring [32]. In this study, treatment with another estrogen (ethinylestradiol) used as a positive control

significantly affected the above parameters, certifying the lack of effects from low BPA exposure. Previous studies in mice exposed perinatally to BPA, via osmotic mini pumps or releasing pellets, have reported increased antral follicles in the ovaries [33] and chromosomal aberrations in the oocytes [34]. Aberrations in the estrous cycle or para-ovarian cysts have also been reported in adult mice offspring perinatally exposed to BPA via subcutaneous injections at doses exceeding 100 µg/kg [35, 36]. The discrepancies in the aforementioned studies may partly reside in the use of different exposure routes, as nonoral administration that avoids first-pass metabolism in the liver could increase active BPA levels.

Short-term exposure of adult female mice to 20 µg BPA/kg/day increased the likelihood of producing aneuploid gametes [37]. In another study, adult female mice were intragastrically administered 5, 25 and 100 µg BPA/kg bw/day for 28 days and then mated to untreated males. Exposure to 25 and 100 µg BPA significantly increased the number of resorptions and the relative uterine weights. Relative ovarian weights were also significantly increased at the 100 µg dose [11]. In the same study, treatment with the leached components of a dental resin, comprising of tri(ethylene glycol)-dimethacrylate (TEG-DMA) (5,945 µg/ml), BPA glycerolate dimethacrylate (BIS-GMA) (2,097 µg/ml) and BPA(78 µg/ml) had similar effects on resorptions and additionally led to increased ovarian weights. This study is one of the few examining the effects of leached components from dental sealants in the reproductive system of rodents, and its findings support the adversity of BPA actions.

Developmental exposure to low BPA doses does not have consistent effects on the adult male reproductive system. Decreased levels of testicular [38] or serum [39] testosterone levels were detected in adult male offspring of mothers exposed to 2–2.4 µg BPA/kg bw during gestation or lactation. Similar reductions in circulating testosterone were detected in juvenile male rats exposed to 40 µg BPA/kg bw during puberty [40]. On the other hand, Kato et al. [41] reported no effects on male rat reproductive system upon neonatal injections of low BPA doses. In recent studies, developmental exposure of rats to 2–200 µg BPA /Kg bw did not affect testes weight or sperm production [6, 42, 43], though earlier studies in mice had reported reductions in the weights of epididymis and seminal vesicles [44].

4.3.3 Effects in the Accessory Reproductive Organs

Exposure to low BPA doses can still have adverse effects in the development of mammary and prostate glands of rodents. Several studies associate perinatal BPA exposure with enhanced mammary gland development in females. BPA administered in pregnant mice via implanted mini pumps (0.025 µg/kg bw/day) stimulated mammary gland development in their offspring [45]. Importantly, this treatment enhanced mammary gland sensitivity to estrogens at puberty onset [46] and the appearance of hyperplasias in adult life [24]. Similar observations have been made in rats, considered more appropriate to model mammary gland pathology [3]. More

specifically, Wistar-Furth rats exposed during embryonic life to 2.5, 25, 250 and 1,000 μg BPA/kg bw developed as young adults increased number of hyperplastic ducts (at all doses) that were positive for estrogen receptor and proliferation markers. The hyperplastic ducts were retained in animals exposed to the lowest dose, whereas carcinomas in situ were detected in adult animals exposed to the higher doses. In another study of the same group, Wistar offspring exposed in utero at 25 μg BPA/kg and challenged at puberty with a chemical carcinogen developed in adulthood more hyperplastic ducts and mammary malignancies than the non-BPA treated. These findings correlate BPA exposure with breast cancer and suggest that prenatal BPA can enhance breast cancer susceptibility by sensitising mammary gland to estrogens and environmental carcinogens experienced later in life. Furthermore, a recent in vitro study, using human breast cancer cell lines, showed that BPA at low doses can antagonise the cytotoxic effects of chemotherapeutics such as doxorubicin, cisplatin and vinblastine [47]. It is proposed that BPA increased the expression of antiapoptotic proteins, probably through interactions with alternative ER receptors expressed in cancer cells.

Adverse effects of developmental BPA exposure have been also reported regarding the rodent prostate. One-month-old male rats exposed in utero to a low BPA dose had transient differences in prostate histology, compared to control males, by means of a larger layer of fibroblasts and increased proliferation in the periductal stroma cells [48]. A transient decrease not observed later in adult life was also detected in the expression of androgen receptor and acid phosphatase in the prostatic cells of BPA-treated males. Several other studies conducted in rats and mice report increased adult prostate size in BPA-treated offspring [for a review, see 2]. The increased prostate size and androgen responsiveness upon perinatal BPA exposure have been suggested to increase the susceptibility of prostate to neoplasia in later life [49]. Male rats neonatally injected with BPA (10 $\mu\text{g}/\text{kg}$ bw) and retreated as adults with testosterone or estradiol showed increased incidence of prostatic neoplasms. Although BPA is studied for its estrogenic properties, its interaction with androgen receptors (ARs) is also possible. In prostate cancer cells, BPA can act as either AR agonist or antagonist depending on the functional state of the receptors. In the presence of wild-type ARs in these cells, BPA blocks androgen actions, whereas in cases of mutated ARs, BPA promotes cell proliferation [50].

4.3.4 Effects in the Central Nervous System

Beyond their traditional roles in the central regulation of reproductive physiology and behaviour, gonadal steroids influence several other brain functions. Estrogens in particular are implicated in cognitive function, neuroprotection and synaptic plasticity, as well as in the neuropathology of Parkinson's and Alzheimer's diseases.

Most estrogen responses are orchestrated by estrogen receptors (ERs) that are ligand-activated transcription factors. Two types of ERs have been identified in

mammals (ER α and ER β) that belong to the nuclear receptor superfamily. The two receptors can mediate distinct actions of estradiol, depending on their interactions with the responsive elements in gene promoters, on the combination of the available co-regulators as well as on the ratio of ER α /ER β in each tissue [51]. ER β may even inhibit the transcriptional activity of ER α . ER α is the predominant receptor in the hypothalamus and the main regulator of reproduction. ER β on the other hand is implicated in nonreproductive estrogen actions and is the principal ER subtype in several brain areas including the cerebral cortex, hippocampus and cerebellum. The dynamics of the two receptors and their distribution are important parameters for the final outcome of estrogen actions in the brain.

BPA has approximately 10,000 times lower binding affinity for ERs, compared to 17 β -estradiol, and based on this, it is characterised as a weak environmental estrogen. BPA has ten times higher affinity for ER β than ER α . Importantly, BPA binding to either ER subtype leads to conformational changes that differ from those induced by estradiol binding. This can critically modify the receptor–ligand complex properties in recruiting co-regulators and/or other interacting proteins. In fact, BPA acts as a selective estrogen receptor modulator (SERM), exhibiting either agonist or antagonist behaviour, depending on the target tissue [52]. Thus, the ratio of the two ERs in a given tissue, along with the dynamic equilibrium of co-activators and corepressors, is critical for the nature of BPA effect.

In addition to nuclear ERs, membrane entities binding estradiol have also been described [21], such as the membrane-bound ER α -like receptor (mER) and a transmembrane G-protein-coupled ER (GPR300). By binding to these membrane receptors, BPA can exert rapid non-genomic actions including increases in cellular Ca⁺⁺ or nitric oxide. These two potent intracellular messages can readily alter enzyme activities or membrane potential and permeability in target cells. Studies in MCF-7 breast cancer cells have shown that BPA via membrane receptors can modify cell properties at concentrations 10⁻¹⁰–10⁻¹² M that are much lower than those needed for an *in vivo* effect. However, it is so far technically difficult to distinguish BPA actions mediated via membrane receptors in experimental animals. Notably, in the above-mentioned *ex vivo* studies, BPA can be equally effective with estradiol. In addition, BPA can bind to other receptors, such as the orphan nuclear estrogen-related receptor- γ or the thyroid hormone receptor; however, the *in vivo* relevance of these interactions has not yet been elucidated.

BPA is able to reach the brain of fetuses or adult rodents shortly upon administration. The estimated time to reach fetal brain is approximately 1 h following subcutaneous injection to the mother [53]. Importantly, the brain does not possess an efficient detoxification system, because the levels of drug-metabolising enzymes are extremely low within this tissue [54], and is thus more susceptible to BPA actions, compared to peripheral tissues.

Perinatal exposure to BPA, as well as to other xenoestrogens, can potentially affect the normal process of sexual differentiation of the brain. In male rodent fetuses, testosterone secreted by the developing testes reaches the brain, where it acts upon conversion to estradiol. This conversion is driven by the enzyme aromatase whose activity increases perinatally, during the critical time windows of brain

masculinisation. Thus, the masculine brain is shaped through the action of estradiol in male fetuses, while at the same time the female fetuses remain hormonally inert [80]. Accordingly, the feminine pattern is the default in a neutral fetus brain. The female fetus is further protected from estradiol of maternal or male littermates' origin by the increased levels of estradiol binding proteins, like alpha-fetoprotein. However, estrogen-binding proteins confer little protection from BPA actions due to its low affinity for them, compared to estradiol.

4.3.4.1 Effects on Brain Structure

Neuronal migration is indispensable during development for the physiological growth of the brain tissue. Migration is particularly important for the formation of multilayered cortex structures. Impaired cortical development and altered connectivity within the brain have been implicated in cognitive deficits and neuropsychiatric disorders in humans [55]. In utero exposure of mice to 20 µg BPA/kg bw results in significant alterations in the migration and differentiation of neocortex precursor neurons [56]. More specifically, BPA enhanced the rate of neuronal migration and differentiation. Furthermore, the expression of neurogenesis-relevant genes (*Math3*, *Ngn2*, *Hes1*) and thyroid receptor-related genes was significantly upregulated in the telencephalus of BPA-treated embryos. When the BPA offspring were examined in later life, they exhibited abnormal positioning of neurons and cortico-thalamic connections. This persistent BPA-induced disturbance in cortical cytoarchitecture resembled the effect of exposure to low ionising radiation [57].

Locus coeruleus is a sexually dimorphic brain area in the brain stem involved in sympathetic stress responses, anxiety and panic. Its dysfunction has been correlated among others with the neurodevelopmental Rett syndrome and the posttraumatic stress disorder (PTSD), while a significant loss of neurons in this area is met in Alzheimer's patients. The volume of this nucleus is larger in females than males, and androgen receptors have been implicated in the emergence of this sex difference in rats, with testosterone lowering neuronal number [58]. BPA treatment of rats at a daily dose of 30 or 300 µg/kg bw, during the fetal and neonatal period, led to the abolishment and inversion of the existing sex difference [59]. The observed increase in locus coeruleus of males has been attributed to the estrogenic action of BPA in this area that harbours both estrogen receptor subtypes.

The anteroventral periventricular (AVPV) preoptic area in the hypothalamus is another sexually dimorphic region that is important for the periodical gonadotropin release and the normal estrous cyclicality. Dopamine-releasing neurons consist the main population in AVPV that is larger in female rodents, compared to males. Tyrosine hydroxylase (TH) is the rate-limiting enzyme in the synthesis of dopamine, and its abundance is also higher in the AVPV of females. The sexual dimorphism of this population of neurons appears to be programmed perinatally by gonadal steroids [60]. Perinatal exposure of mice at low BPA can alter the dimorphic profile of this area. In this study [61], pregnant dams were implanted with osmotic pumps releasing 25 or 250 ng of BPA/kg bw/day from gestational day 8 to

lactation day 16 (in rodents, gestation and lactation last 21 days each). These doses are among the lowest used to date and represent a maximum daily intake of 0.23 $\mu\text{g}/\text{kg}$ bw. A significantly reduced number of TH neurons in the above-mentioned area were observed in the female BPA-treated offspring leading to the abolishment of the documented sexual dimorphism. It is worth noting that other dimorphic nuclei in the hypothalamus, like the sexually dimorphic nucleus (SDN), are not sensitive to BPA actions [59, 62, 63].

Reduction in the number of dopamine-synthesising neurons has also been reported in male rats. Male pups were injected with 0.2–20 μg BPA/pup and tested at 1 or 2 months of age [64]. This reduction was witnessed in the midbrain dopaminergic population and was related with the decreased spontaneous motor activity seen in these offspring (discussed below). No females were used in this study to check for a possible sexually dimorphic effect. Nevertheless, several studies have reported effects of perinatal BPA administration on dopaminergic neurons, indicating the increased sensitivity of this neuronal population to the aforementioned endocrine disruptor during development.

4.3.4.2 Effects on Brain Physiology

Changes in Steroidogenesis and Synaptic Plasticity

Steroids synthesised locally in the brain (neurosteroids) play an important role in many physiological responses including neuroprotection and synaptic plasticity. Neurosteroids' actions have been particularly studied in the hippocampus, a brain area that hosts the required synthesising enzymes. Fetal and postnatal exposure of rats to BPA significantly facilitates the local synthesis of estradiol in the hippocampus [65], implying a role of low BPA concentration in the modulation of brain steroidogenesis and consequently synaptic plasticity. However, in other studies, it is apparent that BPA exposure either in adulthood [66] or in early life [67] can adversely influence the synaptic plasticity in rodent brain. MacLusky et al. [66] showed that exposure of adult ovariectomised rats to a low dose of BPA inhibits the rapid synaptogenic response of pyramidal neurons to estradiol. Given that synaptic remodelling has been related to the rapid effects of estrogens on memory, the authors suggest that BPA exposure may modify the existing sex differences in cognitive function, acting in this case as an estrogen antagonist. Accordingly, BPA exposure during aging could exacerbate the impairment in cognitive function caused in females due to the elimination of endogenous estrogens.

The ability of synaptic contacts to change in response to stimuli is indispensable for the processes of learning and memory. A well-established electrophysiological measure of synaptic plasticity concurrent to memory storage is the long-term potentiation (LTP), indicating the strengthening of certain synaptic contacts upon relevant mnemonic stimuli. Deficits in the development of LTP in the striatum of young male rats have been reported in animals perinatally exposed to 20 μg of BPA [67], providing another example of the antiestrogenic effects of low BPA exposure.

The dorsolateral striatum confers the neuroanatomical substrate for motor control, and BPA-treated male offspring in this study exhibit a significant hyper-locomotion that may relate to the function of dopamine receptors and the improper development of synaptic plasticity.

Changes in Neurotransmission

Brain physiology is greatly depended on synaptic activity that in turn is regulated by neurotransmitters. Significant alterations have been reported in neurotransmitter levels and/or receptors upon exposure to low BPA doses. These alterations so far concern dopamine and other monoamines. Reduced population of dopamine neurons in the midbrain, as detected by the reduced immunoreactivity for tyrosine hydroxylase, was reported in the midbrain of 4- and 8-week-old male rats intracranially injected at postnatal day 5 with 2–20 µg of BPA [64, 68]. The rats had also reduced expression of dopamine receptor D4 and dopamine transporter and exhibited increased spontaneous motor activity. These findings suggest that neonatal low BPA exposure may cause a deficit in the development of dopaminergic neurons that could be causatively linked to the detected molecular and behavioural alterations. Given that the mesocorticolimbic dopamine system has been implicated in the attention deficit hyperactivity disorder (ADHD), a developmental disease characterised by inattention, motor hyperactivity and impulsivity [69], the above results imply a potential contribution of BPA in the appearance of ADHD-like symptomatology. In a recent study using a single intracranial BPA injection (10 µg/pup in 2-day-old male rats), significant alterations were detected 28 days later in norepinephrine, serotonin, dopamine and their metabolites, in the hippocampus, striatum and brain stem [70]. It is noteworthy that although free BPA disappeared from the brain of injected animals 5 h postinjection, BPA effects were well apparent over a period of approximately 1 month.

Nitric oxide (NO) is a gaseous second messenger that also acts as neurotransmitter. It has been implicated in the regulation of several functions including reproductive behaviour. Mice perinatally exposed, through their mothers, to BPA (10–40 µg/kg bw/day) exhibit as adults alterations in the number of cells expressing nitric oxide synthase, the key enzyme for the production of NO, in the medial preoptic area of the hypothalamus and in the bed nucleus of stria terminalis [71]. The observed changes are dose and sex dependent, leading to the loss of normally occurring sex differences or the appearance of novel ones.

Changes in Nuclear Receptors' Levels

Estrogen receptors (ER) are candidate mediators of several BPA actions and at the same time are subjected to modifications induced by this xenoestrogen. In the rodent brain, BPA-induced ER alterations have been detected during adolescence, as well as in the adult life of perinatally exposed animals. Exposure to BPA in utero

resulted in a four-fold increase of ER β mRNA levels in the preoptic area of male rat offspring examined at 1 and 4 months of age [48]. No alterations were observed at the levels of ER α or ER β in the medial basal hypothalamus, denoting that BPA effects within the brain can be locally distinct and area specific. In another study, rats exposed to BPA during early puberty showed altered levels of immunohistochemically detected ER α in a number of hypothalamic nuclei. At puberty, control males had more ER α -positive neurons than females in the arcuate nucleus and medial preoptic area, and BPA treatment further enhanced receptor levels in both areas. In adulthood, BPA also increased ER α levels in another hypothalamic nucleus of treated female offspring [72].

In utero exposure of mice to BPA also increased the expression of ER α and ER β in the dorsal raphe nucleus of male offspring examined at juvenility and adulthood, though variations of the BPA effect were observed at different time periods of adult life [73]. Overall, these results indicate the ability of BPA to change the estrogen responsiveness of neural circuits controlling reproduction during puberty and adulthood, in a sexually and timely distinct way.

Retinoic acid, a vitamin A metabolite, is an essential morphogenetic factor with marked effects on developmental growth and differentiation. Retinoic acid exerts its actions by binding to retinoic acid receptor (RAR) and retinoid X receptor (RXR), belonging to the nuclear receptors' superfamily. In murine embryos exposed to 2 μ g BPA/kg bw/day from day 6 to 17 postcoitum, significant changes in gene expression of the aforementioned receptors were detected in the cerebra and cerebellum that differed between male and female fetuses [74].

Steroid receptor co-activators comprise a class of transcription regulators indispensable for the activation of transcription by steroid hormone receptors. Steroid receptor co-activator-1 (SRC-1) is involved in the transcriptional activity of both thyroid and steroid receptors, including estrogen and glucocorticoid receptors [75]. Transiently increased expression of SRC-1 was witnessed [76] in the hippocampus of male rat pups exposed perinatally through their mothers to a very low BPA dose (100 μ g BPA per liter of drinking water). Given the critical role of SRC-1 in the transcription of many genes regulated by gonadal hormones, its alterations upon low BPA exposure during development provide an extra mechanism through which this compound may interfere with normal growth.

Alterations in the levels of brain glucocorticoid receptors were also detected upon perinatal exposure to a low BPA dose in rats (presented below as part of the stress response system).

Changes in the Stress Response System

The hypothalamic–pituitary–adrenal (HPA) axis is the neuroendocrine system mediating the organism's central stress response. In this circuit, additional brain areas, such as the prefrontal cortex, the hippocampus and amygdala, have a critical contribution [77]. A stressful event activates the sympathetic nervous system and the HPA axis that mobilise catecholamines and adrenal steroids, respectively.

Glucocorticoids (cortisol in humans, corticosterone in rodents) are secreted by the adrenal cortex at high levels during stress. Initially, they synergise with catecholamines to increase sympathetic arousal, cardiac tone and glucose availability in muscle. Later on, glucocorticoids terminate the stress response by lowering HPA axis activation and subsequently their own increased secretion. These glucocorticoid actions are mediated by two types of receptors: the classical nuclear glucocorticoid receptors (GRs), widely distributed in the brain, and the mineralocorticoid receptors (MRs), selectively located in the limbic system [78]. GRs mediate the negative feedback actions that terminate stress response, while MRs maintain basal HPA axis function. Previous data suggest that HPA axis and the hippocampus are potential targets for estrogens' organisational actions [79, 80]. Both the hypothalamus and hippocampus host glucocorticoid receptors and estrogen and androgen receptors and an interplay between gonadal and adrenal steroids appears critical for the fine tuning of hormonal responses in these areas [81, 82].

We recently investigated whether perinatal exposure to a 'safe' BPA dose can affect components of the stress response system in rats [83]. Wistar rats were orally administered 40 µg BPA/kg bw/day for the entire period of gestation and lactation. The dose used had no effect in anogenital distance of the pups or in body weights at puberty onset, compared to the untreated controls. In accordance with the existing literature for similar exposures, the treatment did not alter the time of vaginal opening and cycling in female offspring or the levels of plasma progesterone and testosterone in adolescent females and males, respectively. BPA treatment altered circulating corticosterone and GR levels in the hippocampus of adolescent rats in a sexually dimorphic manner. Under basal conditions, female BPA offspring had higher hormone levels than control females and BPA males, whereas following a mild stressful experience (a Y maze task), corticosterone levels were increased in BPA offspring of both sexes, compared to untreated stressed animals. Additionally, GR levels were altered only in female BPA offspring: They were reduced under basal conditions but increased following stress. These findings show that prolonged perinatal exposure to a weak estrogen can promote the appearance of sex differences in corticosterone levels that normally arise after puberty in rats (adult females have higher levels than the males [84]). Furthermore, they show that the estrogen-mimicking effects of BPA in the enhancement of stress responsiveness [85, 86] are exerted in a sexually dimorphic way. This is the first study to show intervention of a low BPA exposure to the normal maturation of the neuroendocrine stress response system. Future studies are needed to examine whether the observed hormonal and molecular changes are still present in adulthood, under the activational actions of gonadal steroids and their interplay with adrenal hormones and stressful events.

4.3.4.3 Effects on Behaviour and Cognition

The molecular and cellular changes observed in brain physiology have an immediate impact on behaviour. Several studies have shown that perinatal

exposure to low BPA concentrations reduces exploratory behaviour of female offspring and abolishes the sex differences normally existing in this behaviour [59, 61, 68, 76, 87–91]. On the other hand, BPA-treated male rats exhibit reduced anxiety and even increased spontaneous motor activity [68]. Gestational exposure to BPA increases aggressiveness of male mice in early adulthood, without a concomitant increase of testosterone levels, suggesting that other factors are also implicated [39]. Perinatal BPA exposure (40 µg/kg bw) affects social behaviours as well including play behaviour, social grooming and socio-sexual exploration. These behaviours were decreased in young females exposed to BPA [92, 93].

The Morris water maze test is a typical behavioural paradigm to test rodents' ability for spatial learning and memory. Xu et al. [76] have treated rat dams during pregnancy and lactation with 100 µg BPA/liter of drinking water and tested the offspring of both sexes as adults in the water maze. Impaired cognitive performance was detected for male BPA-treated offspring. Using a low dose of BPA (40 µg/kg bw/day), we also detected impairments in the Y maze paradigm of spatial memory in adolescent rats of both sexes perinatally treated [83]. Mice treated with a higher BPA dose (100 µg/kg bw/day) from prenatal day 7 to postnatal day 36 showed also decreased alternation behaviour and decreased novel object recognition [94]. However, the effects of BPA on cognitive abilities may be task specific. Ryan and Vandenberg [95] reported no effects of perinatal BPA exposure (2 µg/kg bw/day) in the spatial memory of adult mice offspring tested in two different short-term spatial memory tests, the radial-arm maze and Barnes maze. In the same study, a higher dose of BPA (200 µg/kg bw/day) was required for the detection of alterations in the anxiety levels of BPA-treated animals.

BPA administration perinatally, or prior to puberty onset, can also affect the sexual activity of treated offspring in adulthood. Perinatal treatment reduced sexual performance of male rats, in terms of latency and frequency of intromissions. In females, BPA produced a small increase in sexual motivation and receptivity [96]. Similar effects were obtained in males upon juvenile exposure to BPA [40]. Interestingly, these findings did not show a potentiation of male behaviour by BPA, as would be expected based on the classical programming actions of estrogens, implying that this xenoestrogen may act in this instance as an estrogen antagonist.

Alterations in maternal behaviour were detected upon administration of 40 µg BPA/kg bw/day in rat dams during pregnancy. Treatment of mothers with BPA significantly reduced maternal care, in terms of licking–grooming behaviour towards the pups and duration of arched-back posture. Moreover, these behaviours were not influenced by the sex of the pup, as is the case in control dams showing more care towards their male offspring, but were rather equally exerted in all pups [97]. Prenatal and neonatal treatment with BPA (40 µg/kg bw/day in dams) also modified the formalin-induced nociception of treated offspring in adulthood in a sexually dimorphic way [98]. The main effects in the nervous system of rodents from low BPA exposures are summarised in Table 4.1.

Table 4.1 Reported effects of low-dose BPA (<50 µg/kg bw/day) in rodent CNS

Altered neuronal migration. Impaired cortex formation [56, 57]
Reversal of sexual dimorphism in the volume of locus coeruleus [59]
Reduced dopamine neurons in the midbrain and in hypothalamic AVPV (loss of sexual dimorphism). Reduced expression of dopamine receptor D4 and transporter in midbrain [61, 64]
Enhanced brain steroidogenesis [65]
Altered synaptogenesis in hippocampal pyramidal neurons [66]
Deficits in the development of synaptic plasticity in the striatum [67]
Altered gene expression of c-fos, dopamine transporter (Dat1) and Hsp70 [68]
Altered monoamine levels in the brain stem, striatum and hippocampus [70]
Altered nitric oxide synthase expression [71]
Altered ER α and ER β levels in preoptic area and hypothalamus [48, 72, 73]
Altered expression of retinoic acid receptors (RAR α , RXR α) in cerebra and cerebellum, depending on the sex of the embryo [74]
Transiently increased expression of SRC-1 in the male pup hippocampus [76]
Altered basal and stress-induced corticosterone levels. Altered hippocampal GR levels in females [83]
Impaired performance in spatial memory tasks [76, 83]
Reduction of motor activity and explorative behaviour in females. Reduction of anxiety in males. Reduction of sexual activity in males and slight enhancement in females [40, 88, 96]
Increased neophobia and anxiety-like behaviour in pubertal females [83, 87]. Male feminisation of adult impulsive behaviour and reduced activity response to amphetamine [87, 91]
Loss of sex differences in explorative and emotional behaviours (open field, novelty test and elevated plus maze forced swim) [59, 76, 89, 90]
Increased spontaneous motor activity [68]
Decrease of playful social interactions in females [92, 93]
Enhanced aggression in males [39]
Alterations in maternal behaviour [97]
Modifications in pain behaviour [98]

4.3.5 *Effects on Other Systems*

4.3.5.1 **Effects in Metabolism**

Estrogens have well-known effects on both peripheral and central energy homeostasis. During development, they regulate adipocyte number, whereas in adulthood they inhibit lipogenesis and adipose deposition exerting an anti-obesogenic action. Both estrogen receptors (alpha and beta) mediate estrogens' action in the adipose tissue, ER α being the principal modulator [99]. BPA, as a weak estrogen, is expected to mimic some of estrogens' actions on energy expenditure and adiposity.

BPA administered at high doses (4 or 5 mg/day) for 15 days in ovariectomised adult female rats had similar effects with estrogens on the reduction of body weight [100]. Another study conducted in adult mice [101] has also reported that BPA imitates the effects of 17beta-estradiol on blood glucose homeostasis through both genomic and non-genomic pathways, depending on the dose used. A single low dose (10 µg/kg) of either estradiol or BPA induced a rapid decrease in glucose levels with concomitant increase of plasma insulin. Longer exposures to estradiol or BPA at doses as low as 10 µg/kg/day induced an increase of insulin in pancreatic beta

cells. Upon 4 days of treatment with either hormone, the mice developed chronic hyperinsulinemia, with altered glucose and insulin tolerance. These findings support an enhancing role of BPA, upon adult exposure, in the development of insulin resistance and consequently of type 2 diabetes, hypertension and dyslipidemia.

Fetal or perinatal BPA exposure has recently been proposed to be a potentially risk factor for the development of obesity and related disorders in adulthood [102]. BPA exposure during this period advances puberty onset [103] and increases body weight gain, adipose tissue mass and cholesterol levels later in life of the exposed mice [35, 104]. However, more recent data do not support a programming effect of low BPA exposures in obesity-associated metabolic disturbances. Ryan et al. [105] reported that perinatal exposure to an ecologically relevant dose of BPA indeed resulted in heavier offspring at 4 weeks of age, compared to the controls, but these differences were no longer apparent when the mice reached adulthood, even when tested on a high-fat diet. These data suggest that perinatal exposure to low BPA doses leads to a faster rate of growth early in development, rather than in an obese, diabetic phenotype in adulthood.

4.3.5.2 Epigenetic Changes

Alterations in the pattern of DNA methylation at CpG-rich promoter sequences of a gene comprise a common epigenetic modification, which can activate (hypomethylation) or silence (hypermethylation) gene transcription. Search for methylation changes in the prostate gland of adult male rats, neonatally exposed to low BPA levels, revealed an altered methylation pattern in several genes involved in cell signalling [49]. One of these genes, the phosphodiesterase type 4 variant 4 gene, coding for an enzyme involved in cyclic AMP degradation, was highly hypomethylated and thus continuously expressed, compared to the controls. Importantly, overexpression of this gene was also detected in prostate cancer cells of neonatally BPA-exposed rats that were as adults treated with gonadal hormones. These findings showed for the first time that perinatal BPA exposure to human relevant doses can lead to epigenetic alterations in genes directly associated with preneoplastic prostatic lesions. In another study in mice, BPA exposure during early life also reduced DNA methylation in the reporter genes, implying a potential of BPA to cause epigenetic alterations in the genome [106]. In this study, the dose of BPA used was high (50 mg/kg of diet) that however is an order of magnitude lower than the dietary non-toxic threshold for rodents [107]. The above observations confer another mechanism of BPA action that directly affects fetal epigenome.

4.4 Concluding Remarks

Animal models provide a powerful tool for elucidating the *in vivo* effects of BPA exposure. There is evidence that low, human relevant, exposure of rodents during development can adversely modify several physiological functions. These include brain neurotransmission and plasticity, behaviour and neuroendocrine responses as

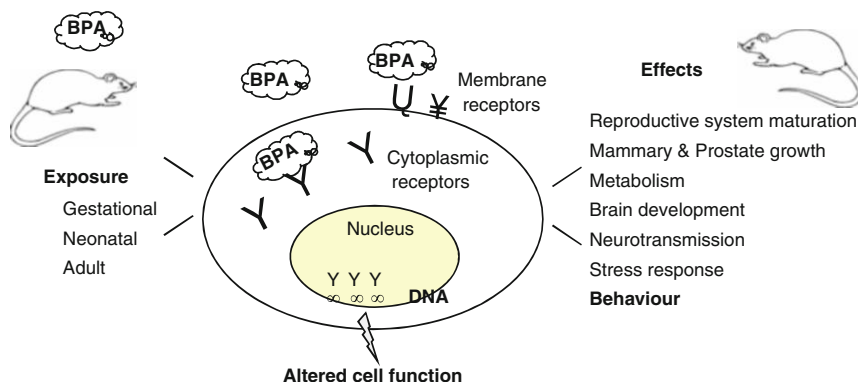


Fig. 4.1 Exposure of rodents to environmental relevant BPA doses during development or in adulthood can affect a number of systems in later life, including the reproductive and the central nervous system. At the cellular level, BPA binds to cytoplasmic receptors for gonadal steroids ($ER\alpha$, $ER\beta$, AR) and mimics or antagonises their actions. BPA can also bind to membrane receptors of steroid hormones or neurotransmitters, further modifying cell function

Table 4.2 Representative low BPA effects in laboratory rats or mice and their possible relevance for human health

Effect in rodents	'Translation' in human health
<i>Developmental exposure</i>	
Abnormal development of brain cortex	Cognitive and neuropsychiatric disorders
Spontaneous motor hyperactivity, related to midbrain dopamine dysfunction	Attention deficit hyperactivity disorder (ADHD)
Increased scores in behavioural tests of 'depression'	Depression
Heightened plasma corticosterone and altered levels of glucocorticoid receptors in the hippocampus	Altered stress response
Enhanced sensitivity of mammary gland to estrogens, hyperplasia	Enhanced sensitivity to mammary hyperplasia
Increased prostate size and incidence for neoplasms	Enhanced risk for prostate neoplasia
<i>Adult exposure</i>	
Decreased sperm production	Infertility
Impaired maternal behaviour	Postpartum emotional dysfunction
Senescence-like disruption of synaptic function in females	Brain aging
Enhancement of insulin resistance	Diabetes type II

well as normal growth of reproductive accessory organs (Fig. 4.1). At low levels, BPA can also exert epigenetic modifications in the whole genome. Several of the observed adversities are reminiscent of human pathologies (Table 4.2) and have sensitised both the public and the scientific communities. Apparently, there is need to revise the risk of such low exposures. However, further research should be conducted before we can safely extrapolate the knowledge from animal data to humans. First of all, the concentrations and kinetics of BPA in different tissues must be precisely determined, both in neonates and older animals, for oral and nonoral

exposures. This will help to delineate the discrepancies of varying effects from the use of rodent species with different genetic backgrounds and sensitivity. Inclusion of appropriate positive controls, taking into consideration all mechanisms of BPA actions, will also facilitate conclusions on BPA properties. Finally, concerted actions for the determination of active BPA levels in human fluids and whenever possible in fetal tissues will allow more reliable comparisons between rodents and humans.

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Part III
Bisphenol-A in Dental Polymers

Chapter 5

BPA and Dental Materials

Jill Lewis

5.1 Introduction and Historical Perspective

BPA has been a controversial component of dental materials since the first reports of its potential toxicities in the mid-1990s [1]. Concerns for the estrogenic and other toxicities of BPA from all environmental sources have escalated over the past decade or so to the point that federal regulatory agencies have begun to address the issue. An early survey was conducted by the Centers for Disease Control and Prevention (CDC) in 2003–2004 (the National Health and Nutrition Examination Survey or NHANES) [2] that found detectable levels of BPA in 93 % of the Americans tested. Because of that survey, the National Toxicology Program (NTP) component of the National Institute of Environmental Health Sciences (NIEHS) focused efforts to evaluate research on the potential adverse effects of BPA [3]. Their findings are expressed on a 5-point scale of concern for adverse effects that includes (from lowest concern level to highest) negligible, minimal, some, concern, and serious. They reported negligible or minimal concern for almost all categories. However, they expressed “some concern for adverse effects” for the effect of BPA on brain, behavior, and prostate gland of fetuses, infants, and children. The US Food and Drug Administration (FDA) issued a statement in 2010 stating that it shares the perspective of the NTP findings concerning the potential effects of BPA [4]. It followed up in July 2012 with a new Food Additive Regulation prohibiting the use of BPA in the production of baby bottles and sippy cups [5].

Thus, what has emerged so far from these federal reviews and surveys is that BPA is extremely ubiquitous in our environment and the most at-risk populations are the very young.

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5.2 What Is the Current “Safe” Level of Exposure?

The Environmental Protection Agency (EPA) and European Food Safety Authority (EFSA) both have set the acceptable BPA exposure limits at $<50 \mu\text{g}/\text{kg}$ body weight/day. This level was based on animal studies performed over 20 years ago that showed rather global adverse effects (decreased offspring, low birth weight, delayed puberty) using high doses of BPA treatment ($50\text{--}500 \mu\text{g}/\text{kg}/\text{day}$). Since then, more recent studies have shown that BPA most likely does not have a monotonic dose response but rather exhibits a biphasic dose response curve suggesting that adverse effects may occur at lower levels than previously thought or, at the very least, responses are unpredictable at low doses [6]. In fact, studies using doses in the $10 \mu\text{g}/\text{kg}/\text{day}$ range are reportedly causing changes in urinary and prostate development and early-onset puberty [7]. Therefore, the current exposure guidelines are somewhat controversial.

5.3 Where Can We Get BPA Exposure from Dental Materials?

Patient exposure to BPA from dental materials is presumed to be primarily through ingestion of released components into the gastrointestinal tract following placement of dental sealants or composite restorations. Although BPA is not itself a component of composite resins, BPA is the starting material for production of common composite monomer, BPA glycidyl dimethacrylate (Bis-GMA), and may persist at trace levels as a contaminant in those preparations. In addition, BPA is released via hydrolysis of another common monomer, BPA dimethacrylate (Bis-DMA), by salivary esterases [7, 8]. Bis-GMA is not susceptible to action of salivary esterases and therefore is unlikely to release BPA into the oral cavity via this route. In addition, exposure to oxygen on the surface of dental sealants inhibits polymerization of monomers and may account for most of the 20–45 % of unreacted monomer that can leach into the saliva. Several studies have shown that BPA can be detected in patient saliva for a short time at detectable levels [9–12]. Salivary BPA levels consistently returned to baseline within a few hours following placement of dental sealants, indicating that BPA exposure from dental materials is primarily an acute event. Some of these studies also monitored BPA levels in serum and urine following sealant placement. While BPA was not detected in serum, elevated BPA levels in urine following sealant placement persisted for up to 5 days [9–12]. Longer-term studies are not available and would be valuable to help determine the contribution of dental materials to chronic BPA exposure. Release of degradation products could occur as composite fillings wear, become more porous, and release unpolymerized monomers that may have been trapped when the restoration was placed and cured. As detection methods have become more sensitive, these studies now are more feasible.

Another largely overlooked potential source for BPA exposure is via uptake of aerosolized or volatile components by the lungs. Dental professionals are at highest risk for occupational exposure by this route, and little research has been done in this arena. However, a 2009 study in Germany reported that low levels of volatile methacrylates have been detected in operatories using solid phase microextraction (SPME) to collect air samples during filling treatment [13]. These investigators found levels of MMA, HEMA, EGDMA, and TEG-DMA in their samplings suggesting that this exposure route deserves further attention.

5.4 Is BPA from Dental Material Sources Significant When Compared to Overall Exposure Levels?

Unfortunately, even after years of research and dozens of studies, controversy over exposure levels and potential adverse effects of BPA from dental materials remains. Since BPA is so pervasive in our environment, attempts to tease out the contributions to overall dose from any single source are exceedingly difficult. In addition, there is no accepted standardization for methodologies that study this problem. As a result, large variabilities in detection limits of various techniques and even variation in the reported units of the results make direct comparisons of studies difficult. Controversies over the studies chosen to set regulatory guidelines also exist [14]. Some investigators have done extensive reviews of the literature to try to standardize the results for comparison as well as recommend guidelines for standardization in future studies [15]. Others have tried to provide some insight into the relative contribution of overall BPA dose from dental materials [16]. These studies sometimes are complicated by the lack of information provided by the manufacturer, whose formulations often are proprietary. However, even with all of these caveats in data interpretation, the contribution to overall dose provided by dental materials often is significant, especially in children, and therefore should not be ignored completely.

5.5 What Effects Have Been Reported from Dental Material-Derived BPA?

The toxic and estrogenic effects of BPA have been studied for many years and summarized elsewhere in this text. More recent papers focus on epigenetic changes, developmental issues, and brain/psychosocial issues. Concerns for BPA toxicities prompted the NIEHS to devote \$30 million in funds from the American Reinvestment and Recovery Act toward the study of BPA. Many of these studies that focused on dental materials examined the effects of BPA exposure on children and development. Based on other *in vitro* and *in vivo* studies, metabolic homeostasis and

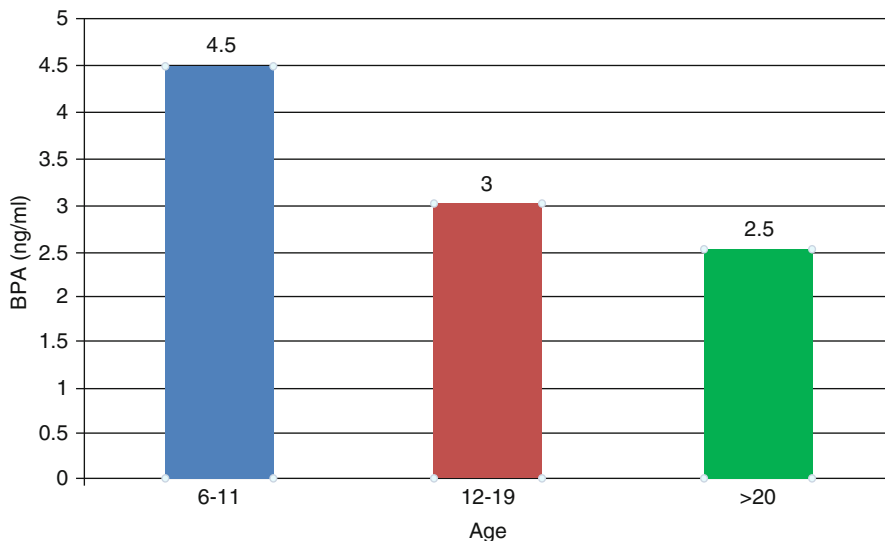


Fig. 5.1 Results of NHANES evaluation of BPA levels in urine

neuropsychological development were two areas of particular interest. Several studies using secondary analysis from the New England Children’s Amalgam Trial conducted from 1997 to 2006 recently have been published. This database includes information on 534 children aged 6–10 with ≥ 2 posterior tooth caries who were randomized into amalgam or composite treatment groups. One such study addressed potential effects on physical growth and found no significant differences in growth rate, percent body fat, or BMI between the two groups. However, they did report some findings that suggest differences in the age of menarche in adolescent girls from the composite group that may warrant further investigation [17]. Other recent studies from this group have focused on neuropsychological development. Results from these studies have been equivocal, sometimes showing an association of composite resin with lower psychosocial scores and others showing trends toward lower executive functioning scores that did not reach statistical significance [18, 19]. This area of research likely will continue to be a topic of focus in the future.

5.6 What Is the Current Impact of BPA on the Dental Profession?

One of the most troubling findings of the NHANES survey was the clear and significant relationship between BPA levels and age. Children (6–11 years of age) had the highest levels of BPA, followed by adolescents, with adults (age >20 years) exhibiting the lowest levels of BPA (Fig. 5.1). No children under age 6 were included in the survey. Developmental effects of BPA exposure are likely to be most

deleterious in the prenatal and perinatal periods. Therefore, in terms of dental patients, pediatric patients and expectant/lactating mothers are most at risk. At the same time, efforts to increase the numbers of children receiving dental sealants have been successful, and the efficacy of dental sealants for reducing the incidence of pit and fissure caries and improving oral health in children is undeniable [20].

Recent articles have reviewed research reports on salivary BPA levels and provide a good perspective on the contribution of BPA exposure from dental materials. The findings of studies using current Bis-GMA-based composites and sealants showed that the acute BPA exposure levels associated with sealant placement were a minimum of 50,000–100,000 times lower than the daily recommended exposure limit for adults set by current EPA and NTP guidelines [21, 22]. Thus, exposure levels from dental materials alone are considered to be safe.

So what should be done? As with any biocompatibility issue, the decision to use these materials is a decision of risk to benefit ratio. The oral health benefits from these materials are well established [23–25]. Because the data on BPA levels and safety remains complex, it seems prudent to limit the exposure contribution from dental materials to these most vulnerable populations as much as possible. Some commonsense guidelines have been suggested in dealing with these issues [26]: Limit elective placement of dental sealant and composite resin restorations in pregnant women. When possible, choose materials that are BPA-free. In pregnant women and children, use precautionary application techniques when restorations or sealants are placed to limit BPA exposure. These techniques are aimed at removing the unpolymerized material on the surface and have been shown to return BPA levels to baseline, greatly reducing the acute, short-term BPA exposure associated with sealant placement [27, 28].

Because of the high profile of BPA safety in the press, many patients may express concerns about the use of these dental materials. The ADA recently published some guidelines to help the dental practitioner address these concerns [21].

BPA and Dental Materials: Addressing Patient Concerns

Here are some key points that can help you answer patient questions about BPA:

- According to manufacturers, BPA is not an added ingredient in dental composites or sealants currently on the market.
- The main ingredient in most commonly used composites and sealants is Bis-GMA, which has been shown to be stable within the mouth and does not decompose to BPA over time.
- Trace amounts of BPA present in raw Bis-GMA are a residue of its manufacturing process.
- Some products contain added Bis-DMA as a Bis-GMA viscosity modifier. Bis-DMA is known to decompose to BPA in the presence of salivary esterases (enzymes). However, many current dental resins severely limit or eliminate all Bis-DMA from their formulations.

- Although trace levels of BPA can be detected in dental products containing Bis-GMA, the potential exposure level is at least 100,000 times lower than current exposure limits.
- BPA exposure from dental materials likely lasts only a few hours after placement of a composite or sealant. Therefore, any BPA exposure is brief and transient.
- The preponderance of scientific data over the past 15 years indicates that the amount of BPA exposure from dental restoratives does not present a health hazard.

Compiled from Ref. [21], used with permission (pending)

5.7 What Can We Do in the Future?

BPA research would benefit from more standardized methodologies to study and report the effects of these compounds in the literature to better facilitate comparisons of the results. In addition, with newer and more sensitive techniques available, studies on longer-term/chronic release of these compounds from dental materials should be possible. Further assessment of volatile release of these compounds during dental procedures should be performed to better delineate potential risks to dental practitioners and staff.

Regulations concerning required product information could be modified to insure that the dental practitioner can make a better informed choice on which materials to use to help limit BPA exposure as much as possible. More extensive studies should be performed on other compounds released from these composite resin systems to assess their estrogenic/toxic effects. Manufacturers should continue to develop new BPA-free materials to offer as options. Of course, these new materials will need biocompatibility testing to insure that new risks are not being introduced.

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Chapter 6

Bisphenol A and Orthodontic Materials

Dimitrios Kloukos and Theodore Eliades

6.1 Introduction

Orthodontic polymers, and their applications, have been instrumental in introducing aesthetics, innovation, and practicality into the orthodontic specialty. Such materials constitute a large class of components including plastic elements and auxiliaries such as adhesives, polycarbonate brackets, and aligners.

The composition and configuration of these materials vary notably. Some of them are based on bisphenol A (BPA), which is used as a precursor of bisphenol A glycidyl dimethacrylate (Bis-GMA) or BPA dimethacrylate (Bis-DMA) during the production of many composite resins. The BPA structure assembles a bulk, stiff chain that offers low susceptibility to biodegradation as well as great rigidity and strength [1]. Although BPA is not used by itself as a raw material in composite resins, it is likely to be present as an impurity from the synthesis process [2, 3].

Since the 1960s, when the use of bisphenol A glycidyl dimethacrylate (Bis-GMA) began to flourish in dentistry, many studies have assessed the effects of dental composites on pulpal impairment [4] and their cytotoxic properties [5–7]. Nevertheless, the systemic health consequences of these chemicals, or their monomers, have not been thoroughly evaluated [8, 9].

Even though the patient may come in contact with significant amounts of unpolymerized monomers during the placement of composites, the release of uncured monomers after polymerization has been assumed to cause most of the unwanted effects [10]. In particular, BPA release from dental resins has attracted recent attention in the literature because of numerous experiments presenting adverse effects of BPA [2, 3, 11]. BPA has shown potential estrogenicity in a significant number of

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studies [12] and is described as an endocrine disruptor chemical (EDC), owing to its ability to bind and activate the human estrogen receptor, however with a capacity of 1,000–5,000 times less than the endogenous 17- β estradiol [13].

Moreover, BPA can interact with other endocrine receptors, as thyroid hormone receptors and peroxisome proliferator-activated receptor gamma [14]. BPA was classified as a reproductive toxic substance of category 3, a significant risk factor for human fertility [15]. The concern is not isolated only at the molecular level. A recently published review indicated that exposure to dental composite resins based on BPA derivatives may even impact psychosocial health in children. Increased levels and duration of exposure (5 years) to composite indicated higher levels of anxiety, depression, social stress, and interpersonal-relation problems in children [16].

The European Food Safety Authority published an initial risk assessment on BPA in 2006, based on a tolerable daily intake (TDI) of 50 $\mu\text{g}/\text{kg}$ body weight/day [17]. Several scientists arguably disputed the use of TDI for risk assessments on EDCs, suggesting that the effects of EDCs are observed at very low doses, non-monotonic dose–response curves, as well as on effects occurring from very specific windows of exposure [18].

The uncertainty in the dental literature was initially provoked by a study published by Olea et al. [19] who reported elevated salivary levels of BPA in patients with dental sealants. Since then, the extensive implementation of new polymers has triggered the investigation of their long-term effects at subtoxic levels. The investigation of the biological properties of materials has deviated from various routine cytotoxicity assays, for example, DNA synthesis or MTT proliferation assay [20].

The orthodontic concerns originate from the fact that monomers equivalent to those used for dental sealants are also used for the construction of orthodontic polymeric adhesives, plastic polycarbonate brackets, and other polycarbonate-made appliances that might also be sources of BPA. However, the actual effects induced by the possible release of BPA are difficult to be assessed because the mode of application of the materials, the growth stage and age of the individual, and potential other environmental factors might alter the extrapolation of results.

The purpose of this chapter is to briefly summarize the limited evidence available on the topic, which is associated with (a) polymeric orthodontic adhesive resins, (b) plastic polycarbonate brackets, and (c) polymeric aligners and their relationship to the possibility of bisphenol A (BPA) release and the subsequent phenomena of estrogenicity. A recently published systematic review was utilized as basis [21] for providing the evidence discussed in this chapter.

6.2 Orthodontic Adhesives

Bonding of brackets to enamel has been an enduring critical issue in orthodontics research.

Biomechanical principles necessitated a relatively inelastic interface that would transfer a load applied to the bracket directly to the tooth or to its root. Furthermore, the engagement of an archwire to the bracket should not exceed the bond strength between bracket and tooth [22]. Based on these requirements a considerable volume

of research was undertaken, aimed to find new materials and new perspectives in the province of orthodontic adhesives.

Orthodontic adhesive exposure to the oral environment involves three patterns:

(a) The bracket peripheral margins

The average thickness of these margins is quantified as between 150 and 250 μm [23]. The effect of aging and leaching of the material throughout these margins and under oral conditions might not be that potent.

(b) Bonded fixed lingual retainers

Fixed retainers have been used in orthodontics for many years. In both arches, mandibular and maxillary, they are routinely used for a prolonged period of time or even permanently. The use of these bonded retainers has been proven and well documented to be efficient in preventing relapse of the orthodontic treatment in most patients [24]. Two main types of fixed retainers are generally used: large-diameter wires, usually made of stainless steel, bonded only to the lingual surfaces of the canines, or small-diameter wires bonded to the lingual surfaces of all six anterior teeth.

For bonding both retainer types, specific orthodontic adhesives, mainly light-cured, are used. The adhesive in this case is used in a mode that involves full exposure of its surface to the oral environment. An extremely large surface-to-volume ratio of the applied adhesive is the main reason that increases its reactivity with the surrounding oral environment and facilitates aging and degradation, with volatile BPA release [25].

(c) Removal of the brackets and cleaning up of the enamel surface

This procedure follows the completion of orthodontic treatment [26]. This standard technique involves grinding and removal of the adhesive layer that existed between the bracket and the tooth with rotating instruments at low or high speed. This process discharges three main fragments in the aerosol that is created: polymer matrix pieces, filler degradation by-products, and particles descending from the wear of the bur [27].

The potentially hazardous nature of this aerosol is double. Potential concerns deal with the respiratory health of the patient and the treatment-providing team, since the produced dust is capable of reaching the alveoli of the lungs [27–29]. If we also take into consideration that the medical team is exposed on a long-term basis to this condition, we can easily assume the importance of these concerns.

Secondly, the particles attained from the presence of a double benzoyl ring in the released Bis-GMA monomers lead, as proclaimed, to the formation and release of bisphenol A (BPA) and hence to potentially disruptive hormonal action [30–33].

6.3 Orthodontic Adhesives: In Vitro BPA Release

Published studies are contradictory with respect to the qualitative and quantitative parameters of elution and BPA release from adhesives, probably because of the varying methodologies that have been employed. Eliades et al. were the first to

investigate the release of bisphenol A from orthodontic adhesives after their artificial accelerated aging with an *in vitro* study [34]. The results showed no indication of BPA identified for either type of adhesive across all time intervals used in the study, i.e., 1 day and 1, 3, and 5 weeks. Nevertheless the authors concluded that although the lack of BPA release was demonstrated in a particularly severe environment and under artificial accelerated aging conditions, these results should not be unquestionably extrapolated to real-life clinical conditions. The given reasons were three: Initially, the analysis of the adhesive extracts should be handled with caution, as far as it concerns the estrogenicity of polymers, because of the documented reactivity of BPA at very low levels [35]. In addition, the detection threshold level of the analytical apparatus used could be well above the potential BPA levels in the analyzed samples. Finally, intraoral aging, which is rather inconsistent with the extraoral reproductive aging, involves complex mechanical and chemical aging with the action of human enzymes, such as esterases, that induce degradation [36].

Similar protocol and techniques for assessing BPA release with the previous research were also used in a recent *in vitro* study of Sunitha et al. [37]. The scope of this study was to assess the BPA released from an orthodontic adhesive by varying the light cure tip distance and correlate it with the degree of conversion (DC). The degree of conversion of a resin composite material is the range of transformation of carbon double bonds (C=C) that exist in the monomer into carbon single bonds (C-C) to form polymers during the polymerization process. This has been found to significantly affect the physical [38, 39], mechanical [40–42], and biological [43] properties of dental composites.

The outcomes of the study displayed that increase in light cure tip distance from the adhesive caused a decrease in the degree of conversion of the substance which, in turn, led to a greater BPA release.

The release of bisphenol A from an orthodontic adhesive used to bond lingual fixed retainers on the surface of teeth was also studied recently from Eliades et al. [25]. Eighteen recently extracted teeth, divided into three groups of six teeth each, were used for this study. A light-cured adhesive was bonded to a twist flex wire adjusted to the lingual surface of the teeth. Then the arches were immersed in double-distilled water for 10, 20, and 30 days. Thereafter, the concentration of BPA in the three eluents was investigated with gas chromatography–mass spectroscopy. The results certified measurable amounts of BPA that were identified for all groups, with the highest found in the immersion media of the 30 days groups: 2.9 mg/L. The control group, which consisted of teeth maintained in immersion media, showed BPA in the mean of 0.16 mg/L.

6.4 Orthodontic Adhesives: *In Vitro* Estrogenicity

The actual contribution of the above amounts of BPA to adolescents and adults remains indefinite, and it is not likely that it would have a direct effect, considering the age of the average orthodontic patient in the retention phase of the treatment,

which may be well above 14 years of age. At such developmental stages, the action of BPA might not have the distinct effects reported for utero or early stages of life.

On the other hand, infants and children, examined on a pound-for-pound basis, have higher relative intakes of many widely detected environmental chemicals because they eat, drink, and breathe more than adults [44].

A recent statement of the US National Toxicology Program concluded that, along with high doses, BPA may show a diversity of effects at much lower ones [12]. A close example is that of phthalate esters, for instance, octaphenol, a substance added to plastics to make them more flexible, durable, and transparent. These plasticizers are capable of altering the uptake of dopamine by hypothalamic cells, at levels as low as 10 parts per trillion [45].

Therefore, there is unfortunately a large window of uncertainty on BPA potential estrogenicity, even if a precise and reliable quantitative estimation is attained. Moreover, there are about 20 different formations of bisphenol, and some of them share estrogenic action with BPA, such as Bis-DMA [30]. Therefore, the direct analysis of the estrogenic action of, artificial or not, aged adhesive eluents may be the method of choice for the inquiry about the potential estrogenic action of orthodontic polymers.

Appraisal of estrogenicity of orthodontic adhesive resins with *in vitro* studies has started to blossom mainly in the last 10 years. Eliades et al. assessed the estrogenic action of a chemically cured and a light-cured orthodontic adhesive resin [46]. The adhesives were bonded to 40 stainless steel brackets divided into two equal groups. The clinical handling of materials was reliably simulated. In total, three representative series of samples were prepared for each adhesive and bracket group. After immersion of the specimens in normal saline, samples of eluent were discharged from each group at 1 day and 1 week following incubation. The probable estrogenicity was measured by the effect of the eluents on the proliferation of cells. Estrogen-responsive MCF-7 breast cancer cells and estrogen-insensitive MB-231 human breast adenocarcinoma cells were used as active group and as control, respectively. The data from both cell lines indicated that no estrogenic activity was detected in the eluents from the resins tested.

Gioka et al. considered that whereas bulk, unimpaired orthodontic adhesive samples, used for the previous research, had not demonstrated estrogenic action, the biological features of their small-scale particles had not been assessed. One of the purposes of her study was to evaluate the estrogenicity of orthodontic adhesive particulates assembled by simulated debonding [26]. A chemically cured and a light-cured adhesive were included in the study. Specimens were prepared by simulating clinical bonding procedures. The adhesives prepared with this method were grounded in glass chambers with a high-speed handpiece. The collected amounts of the ground adhesives were immersed in saline for 1 month at 37 °C, replicating body temperature. Estrogenicity was assessed with estrogen-responsive cell line derived from human breast adenocarcinoma (MCF-7). Estradiol and bisphenol A as positive and saline as negative controls were also used. The proliferation rate of MCF-7 cells was clearly elevated, 160 and 128 %, compared to control for both chemically cured and light-cured adhesives, respectively. Both adhesives

demonstrated therefore an estrogenic behavior. The possibility of irrelevant effects to estrogenicity interfering with proliferation was excluded as the estrogen-insensitive cell line MB-231 did not show any discrepancy in the experimental groups.

6.5 Orthodontic Adhesives: In Vivo BPA Release and Estrogenicity

The estrogenicity in eluents of tested adhesives with *in vitro* studies is usually measured by an established assay, for example, as seen before through the estimation of the proliferation of the estrogen-responsive cell line. These cells are known to express estrogen receptor- α (ER α), which is of paramount importance for the proliferative effect of estrogens. The typical method for measuring estrogenic action *in vivo* is the increase of mitotic indices of rodent epithelia [47]. This strategy may have, however, limited relevance to humans. That is because estrogenicity is diminished from rat hepatic microsomes in contrast with human liver [48]. Receptors for estrogens have been additionally identified in human gingival tissues, supplying evidence that this tissue can be a target organ for human sex hormones [49]. There are also indications of a sex hormone influence on the oral human epithelium reacting to chemical challenge [50]. It has been reported that the oral mucosa of premenopausal woman was appreciably more sensitive to sodium lauryl sulfate found in toothpastes than that of postmenopausal woman.

Up-to-date information about *in vivo* assessment of BPA released from orthodontic adhesives in humans has to do mainly with a recent study of Kang et al. [51]. This study assessed the changes in bisphenol A level in the saliva and urine before and after placing a lingual bonded retainer on the lower dentition of 22 volunteers. The samples were obtained immediately before placement of the retainer and 30 min, 1 day, 1 week, and 1 month after placement. The only significant level of BPA was detected in the saliva collected immediately after lingual retainer placement. Age and gender of the volunteers did not seem to affect the BPA level in the saliva or urine. The salivary BPA level (mean 5.04 ng/mL, levels ranging from 0.85 to 20.88 ng/mL) observed in the immediately collected sample was, as implied by the authors, far lower than the reference daily intake dose. Nevertheless, they concluded that, since some evidence of “low-dose effect” exists, clinicians should reduce the uncured layer of the material, using pumice surface prophylaxis of the adhesive.

The US human exposure limit and European Food Safety Authority have set the tolerable daily intake level of BPA to 50 $\mu\text{g}/\text{kg}/\text{day}$ [17, 52]. The BPA released level from the lingual bonded retainer in this study was far below these doses. However and as already mentioned before, there is some controversy regarding the safe level of BPA exposure. Vom Saal and Hughes [53] proposed the need for a new risk assessment for BPA. They based this proposal on more than 100 *in vivo* and *in vitro* study results indicating that a BPA level far below 50 μg can cause modifications in the biological activities of cultured cells.

Finally, it should be also outlined that there are plenty reports of allergic dermatitis in dental personnel [54–58], which can reasonably be attributed to released monomers

from dental composite resins and, in our case, orthodontic adhesives. A smaller number of case reports of allergic responses in patients, which appear to be linked with the monomers, also exist. The last of these reports [59] described two cases of allergic contact dermatitis to bisphenol A glycidyl dimethacrylate (Bis-GMA) during the application of orthodontic fixed appliances. The authors concluded that these cases highlighted the importance for clinicians of two matters. Firstly, the importance of documenting which bonding agent the clinicians use rather than just recording “bonding upper and lower” and secondly, the conflict to the popular belief that dental adhesives are not eventually all the same, i.e., some have Bis-GMA, others do not.

6.6 Polycarbonate Brackets: In Vitro BPA Release

One of the first to describe BPA release from orthodontic polycarbonate brackets was Suzuki et al. [60]. The materials used in this in vitro experiment were, among others, four different types of polycarbonate orthodontic brackets. Analysis of total and released amounts of BPA resulted in the conclusion that during the synthesis of polycarbonates, nonreacted BPA probably remains inside the materials and is released when they are immersed in water or organic solvents. As for polycarbonates, the thermal conditions during the inclusion of their fillers and fabrication of restorations lead to polymer decomposition and BPA production.

This study was followed by another one from Watanabe et al. [61], who investigated the change in the bisphenol A content in a polycarbonate orthodontic bracket and its leaching characteristics during incubation in water. Polycarbonate brackets were placed in water at 37 and 60 °C. The BPA content in the bracket and the amount of BPA released into the water were analyzed at different time intervals. The BPA content increased in the water with time and was 3.8-fold after 12 months at 37 °C and 12.4-fold after 14 weeks at 60 °C compared with the virgin value. The rate of BPA release also increased with time.

The same team validated the previous findings with their second study [62]. The purpose was to investigate long-term degradation of polycarbonates and the formation of BPA in vivo and in vitro.

The degradation of polycarbonate brackets placed in the oral cavity for up to 40 months was examined regarding surface morphology, BPA content, molecular weight, and glass filler content.

The release of BPA from polycarbonate used in orthodontic brackets, temporary crowns, and denture base resins was examined by immersing them in water at 37 °C for up to 34 months. This study was principally conducted in vitro, but an in vivo attitude was also implied from the brackets retrieved from three patients' oral cavities. The results showed linear relationship for the cumulated amount of BPA eluted into water against time for bracket, denture plate, and temporary crown. BPA eluate increased linearly with time during 12–34 months. The elution of BPA was faster for the polycarbonate bracket. The formation and the release of larger amount of BPA found in the bracket were correlated to larger amount of water absorption in the bracket (2.69 %) compared to that in denture plate and temporary crown (0.07 %).

6.7 Polycarbonate Brackets: In Vivo BPA Release and Estrogenicity

The in vivo BPA release and in vivo estrogenicity from orthodontic polycarbonates are only suggested as logical consequence from the two in vitro studies of Watanabe et al. [61, 62] that were mentioned previously. Specifically, the first one specified the BPA content in the polycarbonate brackets retrieved from patients and attempted to clarify whether the BPA content might change in the oral cavity. It was found that the BPA content in five samples was 56–102 $\mu\text{g/g}$ after 5–15 months. The BPA content was not necessarily correlated with the time the brackets stayed in the oral cavity.

The findings suggested that polycarbonate would degrade in the oral cavity to produce BPA. Based on the in vitro findings, the amount of BPA released in the oral cavity during 5–15 months could be estimated to be a maximum of 3.8 $\mu\text{g/g}$. This estimation was found to be reasonable because the BPA contents in vivo (56–102 $\mu\text{g/g}$) were lower than that in vitro (132 $\mu\text{g/g}$), and the BPA release should be proportional to the BPA content.

The second study suggested that BPA was released from the bracket in the oral cavity more than expected from the in vitro data. However, it was difficult to estimate the amount of BPA released.

The in vitro data obtained in water at 37 °C were as follows: the BPA content in the bracket and the BPA release were 132 and 3.8 $\mu\text{g/g}$ after 12 months and 472 and 37.4 $\mu\text{g/g}$ after 34 months, respectively. Therefore, it was expected that the BPA content would be 132–472 $\mu\text{g/g}$ during 12–34 months. However, the BPA content in vivo was 39–125 $\mu\text{g/g}$ during 18–30 months. Therefore, these results suggested that more amount of BPA was released in the oral cavity compared to that expected from the in vitro data.

Nevertheless, the researchers declared that while in vitro specimens were placed under a static condition in water, the brackets in vivo were, as well understood, exposed to complicated and dynamic conditions. While in oral cavity, toothbrushing, mechanical stresses, thermal alterations, and intake of heterogeneous foods and drinks may all have influenced the degradation of polycarbonates and the release of BPA from the brackets.

Therefore, BPA content released in the oral cavity will not always be correlated to the degradation of polycarbonates, since BPA content is the result of the balance of BPA formed and BPA released in the oral cavity, even if molecular weight decrease is correlated with the degradation of polycarbonate molecules.

6.8 Aligners

The development of clear polymeric aligners as a potential substitutional option instead of conventional brackets and archwires is already a fact in modern orthodontics [63, 64]. Patients are typically required to wear the set of aligners for a

minimum of 2 weeks, for 22 h per day, to achieve progressive tooth movement [64]. Although some controversy exists over the efficiency and limitations of this method, polymeric aligners have become an integral part of the daily orthodontic practice.

The fundamental constituent polymeric module of Invisalign aligners is polyurethane. Polyurethane is not an inanimate or inactive material and is affected by moisture, heat alterations, and sustained contact with enzymes that usually exist in the oral cavity [65, 66]. Eliades et al. assessed the cytotoxicity and estrogenicity of Invisalign appliances (Align Technology, Santa Clara, Calif) [67]. The results failed to demonstrate measurable biologic effects from aligners. Two reasons were thought from the authors to might have contributed to this effect: the short time frame of the study model, although it was longer than in actual clinical conditions, and the stability of the aligners as a material, which are basically, as described before, polyurethane-derived products [68].

6.9 Conclusions

The variety of setups did not allow quantitative synthesis of individual study findings. However, the release of BPA is a well-demonstrated phenomenon in oral conditions, which requires special clinical handling and further research. Despite the lack of consistency in methodological approaches, the qualitative analysis of the studies revealed that:

1. High level of BPA was detected in the saliva collected immediately after lingual bonded retainer placement.
2. Increase in light cure tip distance from the adhesive caused a decrease in the DC of the substance which, in turn, led to a greater BPA release.
3. Direct exposure of the adhesive to the oral fluids appears to play an important role in BPA release. Thus, adhesives used to bond lingual retainers leached more components in contrast to adhesives used to bond brackets (exposure through peripheral margins of the bracket).
4. Polycarbonate was found to show evidence of degradation in both in vitro and in vivo conditions and, under specific conditions, released BPA.

Clinical Recommendations

1. It is recommended to keep the light cure tip as close to the adhesive as clinically possible.
2. The use of pumice prophylaxis after bonding may reduce the potential for BPA release.
3. The use of indirect irradiation (around the bracket edges) instead of direct irradiation (through the bracket) is recommended.
4. Mouth rinsing during the first hour after bracket or retainer bonding may prevent the exposure of patients to the potential hazard of leaching monomers.

Future Research Recommendations

1. Large-scale in vivo studies, focusing on the effects of BPA released in saliva or blood of patients after placement of brackets or lingual retainers on developmental and reproductive toxicity.

Recommendations for Standardization Across Studies

1. It is recommended to express the quantitative data of release in standardized units. When the release is expressed per surface area or volume, the data can be linked to teeth or oral conditions.
2. It can be that a compound is released, even if it is not detected, if the concentration is below the detection threshold; therefore, limits of detection of eluates should be always mentioned.
3. The use of polymer-based materials, such as plastic instruments, plastic containers, or disposable gloves, is discouraged, as they may leach components themselves and they can cause contamination, leading to false-positive results.
4. When human saliva is used as incubation medium, it should originate from volunteers without resin restorations and with baseline check for BPA.
5. If long-term release is to be assessed, refreshing the elution medium at predetermined time periods is recommended. This way the solution is not saturated by leached compounds.
6. A constant temperature of 37 °C is preferable.

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Index

A

- Aryl hydrocarbon receptor (AhR), 12
- Attention deficit hyperactivity disorder (ADHD), 102

B

- Bisphenol A (BPA)
 - aligners, 132–133
 - analytical techniques
 - GC-MS, 65–67
 - LC, 67–69
 - applications, 51
 - chemical structure, 52
 - clinical recommendations, 133
 - definition, 89
 - dental materials, 54–56
 - exposure limits, 118
 - historical perspective, 117
 - NHANES survey, 120
 - patient concerns, 121–122
 - patient exposure, 118
 - research benefits, 122
 - significance of, 119
 - toxic and estrogenic effects, 119–120
 - in environmental samples, 57–59
 - extraction techniques, 54–63
 - liquid-liquid extraction, 53, 64
 - SBSE, 64
 - solid-phase extraction, 53, 64
 - SPME, 64
 - in food samples, 60–61
 - immunoassays, 68
 - in vivo animal studies
 - advantages and disadvantages, 89
 - appropriate positive controls, 95
 - dosage, 92–93
 - exposure duration, 93
 - exposure route, 91
 - low-exposure studies (*see* Low-exposure studies, bisphenol A)
 - pharmacokinetics, 91–92
 - rodents exposure, 108
 - rodent species, 93–94
 - sexes, 94
 - orthodontic adhesives
 - bonded fixed lingual retainer, 127
 - bracket peripheral margins, 127
 - in vitro estrogenicity, 128–130
 - in vitro release, 127–128
 - in vivo release and estrogenicity, 130–131
 - removal of brackets and enamel surface cleaning, 127
 - physico chemical properties, 52
 - polycarbonate brackets, 131–132
 - qualitative analysis, 133
 - quality assurance and quality control, 68, 70
 - research recommendations, 134
 - sampling and storage, 53
 - sources and occurrence
 - biological samples, 62–63, 72
 - dental restorative materials, 54–56, 70–71
 - environmental samples, 57–59, 71
 - food samples, 60–61, 71–72
 - standardization, 134
 - TDI, 126
 - in urine samples, 62–63
- N-O-Bis(trimethylsilyl)trifluoroacetamide (BSTFA)*, 66

- Breast cancer
 epidemiological data, 19
 ER α , 18
 incidence of, 18
- BSTFA. *See* *N-O*-Bis(trimethylsilyl)trifluoroacetamide (BSTFA)
- C**
- Cell-free assay systems, 80–81
- Central nervous system
 behaviour and cognition
 BPA administration, 105
 BPA exposures, 105–106
 BPA, 99
 brain physiology
 neurotransmission changes, 102
 nuclear receptors' levels, 102–103
 steroidogenesis and synaptic plasticity, 101–102
 stress response system, 103–104
 brain structure, 100–101
 estrogen receptors, 98–99
- Cytochrome P (CYP) induction, 13
- D**
- Dental materials and BPA, 54–56
 exposure limits, 118
 NHANES survey, 120
 patient concerns, 121–122
 patient exposure, 118
 research benefits, 122
 significance of, 119
 toxic and estrogenic effects, 119–120
- Diabetes mellitus, 20
- E**
- Endocrine-disrupting chemicals (EDCs)
 biological effects, 4
 compound identification, 26–27
 cytochrome P induction, 13
 definition, 3, 79
 effects
 on bones, 23
 breast cancer, 17–19
 diabetes mellitus, 20
 exposure in utero and during lactation, 23–25
 on female genital system, 15–16
 immunologic, 22
 on male genital system, 16–17
 maternal adipose tissue, 24
 neurologic defects, 21–22
 obesity, 19–20
 prenatal exposure, 24
 risk assessment, 25–26
 estrogenic substances, 4–6
 mechanism
 CYP induction, 12, 13
 estrogen receptor signalling pathway
 (*see* Estrogen receptor signalling pathway)
 physical and chemical properties, 7
- Endocrine disruptors. *See* Endocrine-disrupting chemicals (EDCs)
- Environmental estrogens, 3
- E-screen assay, 82–83
- 17 β -Estradiol, 79
- Estrogen-mimicking properties, 96
- Estrogen receptor signalling pathway
 AhR, 12
 classical and alternative ligand-binding domain, 11
 functional domains, 8
 genomic and nongenomic actions, 10
 isotypes, 7
 peptide growth factors, 9
 signaling pathways, 9
 transcriptional activity, 8
- Extraction techniques, BPA, 54–63
 liquid-liquid extraction, 53, 64
 SBSE, 64
 solid-phase extraction, 53, 64
 SPME, 64
- F**
- Female genital system, 15–16
- G**
- Gas chromatography-mass spectrometry (GC-MS)
 acetylation, 66
 electron impact, 65–66
 silylation, 66
- Glucocorticoids, 104
- Gonadal steroids, 93
- H**
- Human breast milk, 24
- Hypothalamic–pituitary–adrenal (HPA) axis, 103–104

I

- In vitro assays, oestrogenicity
 - cell-free assay systems, 80–81
 - mammalian cell assay systems
 - E-screen, 82–83
 - genetically engineered mammalian cell systems, 84
 - Ishikawa cells, 83
 - yeast assay systems, 81
- In vivo animal studies
 - advantages and disadvantages, 89
 - appropriate positive controls, 95
 - dosage, 92–93
 - exposure duration, 93
 - exposure route, 91
 - low-exposure studies
 - accessory reproductive organs, 97–98
 - CNS (*see* Central nervous system)
 - DNA microarray analysis, 96
 - epigenetic changes, 107
 - estrogen-mimicking properties, 96
 - metabolism, 106–107
 - reproductive system, 96–97
 - pharmacokinetics, 91–92
 - rodents exposure, 108
 - rodent species, 93–94
 - sexes, 94

L

- Liquid chromatography (LC)
 - electrochemical detection, 62–63, 67
 - fluorescence detection, 60–63, 67
 - mass spectrometry, 54–63, 68, 69
 - ultraviolet detection, 54–56, 67
- Liquid–liquid extraction (LLE), 53, 64
- Locus coeruleus, 100
- Low-exposure studies, bisphenol A
 - accessory reproductive organs, 97–98
 - CNS (*see* Central nervous system)
 - DNA microarray analysis, 96
 - epigenetic changes, 107
 - estrogen-mimicking properties, 96
 - metabolism, 106–107
 - reproductive system, 96–97

M

- Male genital system, 16–17
- Mammalian cell assay systems
 - E-screen, 82–83
 - genetically engineered, 84
 - Ishikawa cells, 83

- Maternal adipose tissue, 24
- N-N*-Methyl-(trimethylsilyl)trifluoroacetamide (MSTFA), 66
- Molecularly imprinted polymers (MIPs), 65
- Morris water maze test, 105
- MSTFA. *See* *N-N*-Methyl-(trimethylsilyl)trifluoroacetamide (MSTFA)

N

- National health and nutrition examination (NHANES) survey, 120
- Neurologic defects, 20–22
- Neuronal migration, 100
- Nitric oxide, 102

O

- Obesity, 19–20
- Oestradiol, 16
- Orthodontic adhesives
 - bonded fixed lingual retainer, 127
 - bracket peripheral margins, 127
 - in vitro estrogenicity, 128–130
 - in vitro release, 127–128
 - in vivo release and estrogenicity, 130–131
 - removal of brackets and enamel surface cleaning, 127

P

- PCBs. *See* Polychlorinated biphenyls (PCBs)
- Pharmacokinetics, 91–92
- Pleiotropic effects of estrogens, 7
- Polycarbonate brackets, 131–132
- Polychlorinated biphenyls (PCBs), 19–21
- Polyurethane, 133
- Posttraumatic stress disorder (PTSD), 100

R

- Reproductive system, 96–97
- Retinoic acid, 103
- Rodent species, 93–94

S

- SBSE. *See* Stir bar sorptive extraction (SBSE)
- Silylation, 66
- Solid-phase extraction (SPE), 64
- Solid-phase microextraction (SPME), 64
- Steroidogenesis, 101–102
- Steroid receptor co-activator-1 (SRC-1), 103

Stir bar sorptive extraction (SBSE), 64
Synaptic plasticity, 101–102

T

Testicular cancer, 17
Testicular dysgenesis syndrome (TDS), 17
TH. *See* Tyrosine hydroxylase (TH)
Tolerable daily intake (TDI) of BPA, 92, 126
Tyrosine hydroxylase (TH), 100

X

Xenoestrogens, 79. *See also* Endocrine-
disrupting chemicals (EDCs)
breast cancer, 19

E-screen, 82–83
female genital system, 15
genetically engineered, 84
Ishikawa cells, 83
male genital system, 17

Y

Yeast assay systems, 81